# Jack Well Complaint Phase II Investigations

## Submitted To:

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#### Executive Summary

In July of 2008 the Alberta Research Council (ARC) produced a report for Alberta Environment concluding that the gas present in Mr. Jack's water well included a small component of deeper, thermogenic gas. To further investigate the source of this gas and the possible connection between energy wells in the vicinity of the Jack well, additional data was collected from the nearby energy wells, and a series of additional tests were conducted including: a soil gas migration study, a cement integrity investigation, a lineament study and shut-in test. Alberta Innovates – Technology Futures (AITF, formerly ARC) was contracted by the ERCB to collect or oversee collection of this additional data. In addition to the original recommendations made in the 2008 report, an isotopic mudlog from a new nearby energy well, and a series of pumping tests on the Jack water well prior-to and after remedial cementing of the energy well in question, were also conducted to provide more information about the isotopic labeling of thermogenic gases in the area and the possible effects of remedial work on the energy well on the Jack water well.

The new information collected and reviewed as part of this report supports the original conclusion that gas in the Jack water well has a mixture of biogenic and thermogenic sources. The  $\delta^{13}$ C composition of methane indicates a primarily biogenic source for methane, but the presence of ethane and propane, and their isotopic signatures indicate a contribution of thermogenic gases to the well. A review of the completion details and gas isotope composition of energy wells located near the Jack well identified two energy wells with surface casing vent flows and uncemented zones 100/06-12-078-08 W6M and 100/11-12-078-08 W6M. The results of the soil gas migration study confirmed that gas was leaking from these energy wells. The isotopic signatures of ethane sampled from the surface casing vents of these two energy wells is similar to the isotopic signature of ethane gas in the Jack water well. The 24 hour pressure build-up test conducted on 100/06-12-078-08 W6M showed no evidence of a direct connection between the energy well and the Jack water well over the test duration. The isotopic mudlog from a nearby energy well provided information on the isotopic signatures of the potential source formations. The  $\delta^{13}C$  of ethane sampled from the Jack well is similar to ethane sampled from the Cadotte Formation. The lineament study identified a lineament in the area that aligns with the expected fracture orientation and trends towards energy well 100/06-12-078-08W6M. Water levels compiled from manual measurements made in 2001, 2006, 2008, 2009 and 2010 show a general trend of decreasing levels between 2001 and late 2008 (decrease of 12.6 m over time period), followed by an increasing trend that seems to have started in late 2008 or early 2009 (increase of 2.3 m over that time period). The spontaneous eruptions of the Jack water well stopped at about the same time that water levels in the well began rebounding (late 2008 to early 2009). The new neutron density logs indicate that after the remedial cementing of 100/06-12-078-08W6M, the top of the cement was 490 mKb, leaving an uncemented interval from 298 to 490 mKb. Two bubble tests conducted on the well after remedial work show that there are no longer surface casing vent flow from this well. Pumping tests conducted on the Jack water well before and after the remedial cementing was completed at 100/06-12-078-08W6M indicate some changes in the amount of gas discharging from the well, but these changes could also be related to the increasing trend in water levels that occurred during this period. The  $\delta^{13}$ C signature of methane gas sampled from the Jack well after the remedial cementing occurred at 100/06-12-078-08W6M was more negative, but still within the range of laboratory variability prior to the remedial cementing. The  $\delta^{13}$ C of ethane remained within the pre-remedial cementing range, but concentrations of ethane and propane increased slightly. Despite the cessation of spontaneous eruptions at the Jack water well, the May 2010 sampling indicates that thermogenic gas is still present in the well water and repeat sampling is recommended to evaluate whether these concentrations are decreasing. The ethane isotope composition of the gas in the Jack water well indicates the Cadotte Formation as a possible source. While the results of this study helped to provide a better understanding of the isotopic labeling of gasses in the area and potential sources of gases to the Jack well, the results were not conclusive in identifying the source, pathway or method of introduction of thermogenic gas to the Jack well.

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### 1 INTRODUCTION

In July of 2008, the Alberta Research Council (ARC) produced a report examining all the data in the Alberta Environment and Energy Resources Conservation Board (ERCB) files regarding the Bruce Jack water well complaint (see ARC, 2008). This report concluded that a small component of a thermogenic gas was present in the Jack water well. Several recommendations were made for additional information to be gathered and evaluated to determine the source of this gas. Alberta Innovates – Technology Futures (AITF formerly ARC) was contracted by the ERCB to collect or oversee collection of this additional data. This report details and evaluates the new information collected.

## 2 METHODOLOGY

The results summarized in this report include work that was performed by consulting firms hired directly by the ERCB, data collected by logging companies hired by individual energy well operators as requested by the ERCB, as well as field and laboratory services provided by AITF. The results of the logging have already been reviewed by the ERCB. Methodologies for the gas migration studies (Baseline Water Resources, 2009; Lionhead Engineering, 2009, a, b, c) and the lineament study (Mollard and Associates, 2009) were provided by the consultants who did the work; and are only briefly described below.

### 2.1 Water and Energy Well Gas Sampling

Previous work in the area (ARC, 2008) included sampling of gas composition and carbon isotope data from energy wells within about a 1.5 km radius of the Jack water well (Figure 1). On March 19, 2009 AITF collected samples for composition and carbon isotopes from the following locations:

- Water injection well 100/06-12-078-08 W6M
  - Surface casing vent (SCV).
  - A gas sample was not taken from the production casing because no gas accumulated there.
- Water source well 100/11-18-078-07 W6M
  - Production tubing.
  - Production casing.
  - Exsolved gas sample was separated from the produced water.

The water source well 100/11-18-078-07 W6M was targeted because it was identified as being completed in the Cadotte Formation and is the source of water injected in to 100/06-12-078-08 W6M. In addition to the energy wells, AITF also collected another gas sample from the Jack water well on May 23, 2010.

Samples were collected in FlexFoil<sup>®</sup> gas bags and shipped to AITF in Vegreville for compositional analysis. Gas samples were analyzed for atmospheric gases and hydrocarbon fractions (methane, CH<sub>4</sub>, C1; ethane, C<sub>2</sub>H<sub>6</sub>, C2; propane, C<sub>3</sub>H<sub>8</sub>, C3; and butane, C<sub>4</sub>H<sub>10</sub>, C4). A

second sample was sent to the University of Alberta for carbon isotope ratio analysis of C1 through C4 hydrocarbons and CO<sub>2</sub>.

#### 2.2 Energy Well Soil Gas Migration Sampling

Gas migration studies around several energy wells were performed to determine whether gas was leaking outside the surface casing. On May 31, 2009 Lionhead Engineering (hired directly by Penn West Energy) conducted soil gas sampling at locations on three energy well leases (100/14-01-078-08 W6M, 100/16-12-078-08 W6M and 100/02-14-078-08 W6M). Samples were sent to Maxxam Analytics Inc. for hydrocarbon and atmospheric gas analysis. Samples were sent to the University of Alberta for stable carbon isotope analysis of hydrocarbon gases and CO<sub>2</sub>. A letter report was issued by Lionhead Engineering (June 23, 2009). The ERCB deemed these samples were taken at an inappropriate time (soil moisture conditions were too high) and re-sampling was requested.

On July 16 and 17, 2009 Lionhead Engineering (hired directly by Penn West Energy) conducted soil gas sampling at locations on eight energy well leases (100/14-01-078-08 W6M, 100/08-11-078-08 W6M, 100/02-12-078-08 W6M, 100/06-12-078-08 W6M, 100/11-12-078-08 W6M, 102/11-12-078-08 W6M, 100/16-12-078-08 W6M and 100/02-14-078-08 W6M). Field soil vapour readings were taken with a hand-held RKI Eagle meter. Soil moisture conditions were acceptable at the time of sampling. Sampling was observed by ERCB personnel from the Grande Prairie Field Centre. Samples were sent to Maxxam Analytics Inc. for hydrocarbon and atmospheric gas analysis. Samples were sent to the University of Alberta for stable carbon isotope analysis of hydrocarbon gases and CO<sub>2</sub>. Two letter reports were issued by Lionhead Engineering (July 23, 2009 and August 11, 2009).

On September 15 and 16, 2009, gas migration sampling was performed around two energy wells to confirm results provided by Lionhead Engineering and Penn West Energy. The investigation was performed by Baseline Water Resources Inc. of Calgary under contract to the ERCB. Two energy wells were investigated: 100/06-12-078-08 W6M and 100/11-12-078-08 W6M. Both energy wells had sample points starting as close to the casing as possible and then radiating in a "+" pattern away from the energy well in four directions except where surface obstacles were present. Boreholes were hand augured to a depth of approximately 80 cm and a soil vapour probe was installed. The probe was sand packed from the base of the screen to just above the screen and then sealed with bentonite powder which was saturated with fresh water. The probes were hooked to a pump and the ambient air removed. They were allowed to equilibrate with soil vapour overnight. Field soil vapour readings were taken with a hand-held GasTech® meter and the samples with the three highest readings and a background sample were transferred directly into Tedlar® bags. Samples were shipped to AITF in Vegreville for compositional analysis. Gas samples were analyzed for atmospheric gases and C1 through C4. Samples were sent to the University of Alberta for carbon isotope analysis of C1 through C4 and CO<sub>2</sub>. A letter report of the field activities for these soil vapour investigations has been provided by Baseline (November 25, 2009).

#### 2.3 New Energy Well Gas Mudlog

A new energy well was drilled by Exshaw Oil Corporation at 100/02-04-078-07 W6M on October 30, 2009. During drilling of this energy well a Pason gas detector was used to identify the presence of gas in the mud returns. At the request of the ERCB, gas samples were collected at several selected intervals including the Smoky Group, Doe Creek, Shaftsbury, the Base of Fish Scales, Paddy, Cadotte Formations and the Spirit River Group. A stratigraphic section for the Northwestern Plains and Deep Basin of Alberta is shown in Figure 2. Gas samples were analyzed in the field for composition by Continental Labs using a total hydrocarbon gas meter. Samples were then sent to the University of Alberta for carbon isotope analysis of C1 through C4 and CO<sub>2</sub>.

#### 2.4 Cement Integrity Investigations

The integrity of cement from all energy wells within a 1.5 km radius was investigated in a November 2006 report prepared for Penn West by Lionhead Engineering and Consulting Limited and summarized in the July 8, 2008 report by ARC (ARC, 2008). This information was used to identify energy wells that could potentially be the source of the thermogenic gases to the Jack well (Table 5). The wells included in Table 5 were selected based on having uncemented zones, surface casing vent flows and gas isotope signatures within the range expected for thermogenic gas. Four of the energy wells tested positive for gas migration. Three of these energy wells are located fairly close (less than 800 m) to the Jack water well:

- 100/14-01-078-08 W6M,
- 100/11-12-078-08 W6M,
- 100/06-12-078-08 W6M.

Energy well 100/06-12-078-08 W6M is the closest energy well to the Jack water well (429 m away). This water injection well had an uncemented section between 298 and about 1000m. The ERCB requested that new Radial Cement Bond and neutron density logging be performed on this well to determine the top of the cement in the borehole, the quality/integrity of the cement in isolating hydrocarbon-bearing zones and to identify zones where hydrocarbon gases may have entered the energy well bore. The logging of energy well 100/06-12-078-08 W6M was performed on January 19, 2010 by Hotwell Canada Limited (hired directly by Penn West).

#### 2.5 Lineament Study

A lineament study of the Spirit River area was performed by J.D. Mollard and Associates Limited under contract to the ERCB. The study area extended from Township 73 to 82 and Range 2 to 13 W6M. Digital photography from 1961/'62 and 2008, digital elevation data, satellite imagery and regional surface hydrology data was used to identify lineaments in the Spirit River area that could possibly be related to the regional stress field and deeper bedrock structures (folds and faults). A description of the methodologies is provided in the Mollard and Associates Limited (2009) report.

#### 2.6 Pressure Build-up Test

In order to evaluate whether a direct connection existed between energy well 100/06-12-078-08 W6M and Mr. Jack's water well, the surface casing vent was closed on the energy well and pressure was allowed to build up. The pressure build-up test was performed by Nelgar Oilfield Services Limited (hired directly by Penn West). The surface casing vent was shut in at 9:23 a.m. on March 19, 2009 and pressures were monitored for 24 hours. After the build-up the surface casing vent was opened and allowed to depressurize. During the pressure build-up, water levels and gas release from Mr. Jack's water well was monitored by AITF using a pressure transducer and a Calscan Hawk 9500 gas flow meter.

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### 2.7 Pre and Post 100/06-12-078-08 W6M Remedial Cementing Testing of the Jack Water Well

Penn West decided to abandon the injector well located at 100/06-12-078-08 W6M and the down-hole portion of the well was prepared for abandonment between January 15 and 25, 2010. A pumping test was performed on Mr. Jack's water well prior to the remedial cementing of the energy well, and approximately one week, four weeks and 4 months after cementing, to see if there were any changes in the behavior of the water well. Details of the start and duration of the pumping and recovery phases of the tests are summarized in Table 1.

	Pumping test #1	Pumping test #2	Pumping test #3	Pumping test #4
	Pre-cementing	1 <sup>st</sup> Post-cementing	2 <sup>nd</sup> Post-cementing	3 <sup>rd</sup> Post-cementing
Pump started	11:15 a.m.,	9:10 a.m.,	7:22 a.m.,	3:20 p.m.,
	Sept. 26, 2009	Jan. 30, 2010	Feb. 28, 2010	May 22, 2010
Recovery	4:47 p.m.,	9:00 a.m.,	7:36 a.m.,	4:12 p.m.,
(pump stopped)	Sept. 27, 2009	Jan. 31, 2010	Mar. 1, 2010	May 23, 2010
End of recovery monitoring	6:00 a.m.,	1:05 p.m.,	12:35 p.m.,	9:00 p.m.,
	Sept. 28, 2009	Jan. 31, 2009	Mar. 1, 2010	May 23, 2010

#### Table 1: Summary of pumping test timing.

For each of the tests a transducer was placed in the Jack water well before the start of the test to record the initial pre-pumping water levels. A Calscan Hawk 9500 gas and water flow meter was hooked into the Jack water well to measure gas flow from the casing and water flow from the discharge line. The existing pump in the Jack water well was used to conduct the test. After about 24 hours of pumping the pump was shut off and the recovery portion of the test included monitoring of water levels in the Jack water well. A third post-remedial cementing pumping test was initially started at 9:45 am on May 21, 2010, but was interrupted by a power failure and resumed at 3:20 pm the next day. For this pumping test, the discharge line was run through the gas separator in the AITF trailer so that free gas could be measured and a gas sample could be taken.

The Jack water well has a gas vent attached to the top of the casing (TOC). Water level logger data was measured as depths to water from the top of the vent, but are converted to depths from the TOC in the text of this report. The top of the vent is 0.61 m above the TOC.

The water level data from the pressure build up test and the pumping tests presented in Figures 14-18 are the original measurements from the top of the vent.

#### 3 RESULTS

#### 3.1 Water and Energy Well Gas Sampling

The results of new and previous Jack water well and energy well gas sampling are presented in Table 2, Figure 3 and copies of the new analytical reports are provided in Appendix A.

New samples collected from the Jack water well and energy wells have carbon isotope signatures similar to the ranges identified in previous investigations (ARC, 2008). In general, gas sampled from the Jack water well is characterized by high concentrations of methane and much lower concentrations of ethane or higher order hydrocarbons. The isotopic composition of the methane sampled between 2006 and 2008 stayed fairly consistent at -65.6 ‰ (n=8, stdev=0.3) with the exception of a sample taken on October 19, 2006 and analyzed at Zymax laboratory ( $\delta^{13}$ C = -69.3 ‰). The sample analyzed by Zymax was a duplicate sample that was also analyzed by Maxxam. The  $\delta^{13}$ C value reported by Maxxam was -65.80 ‰ which is similar to previous measurements. The difference between these duplicate samples taken using the same sampling protocol but analyzed at two different laboratories, highlights the magnitude of potential analytical uncertainty. As laboratory standards for methane isotope analyses are not readily available or universally applied there may be significant inter-laboratory variability. The post-remedial cementing sample collected from the Jack well in 2010 had an isotope signature that was slightly more negative than previous results. While this shift is consistent with a reduced thermogenic gas contribution to the well, the observed differences are not sufficient to rule out the possibility that the observed shift may be related to analytical uncertainty from the AITF laboratory. Comparisons of the results of known  $\delta^{13}$ C methane standards submitted to isotope laboratories at University of Alberta, University of Calgary and the University of Victoria, suggest that their  $\delta^{13}C_{\text{methane}}$  analyses are comparable (ARC, 2009). The AITF isotope laboratory was not yet open when this comparison was done, so we recommend that the isotope analyses on any future gas samples from the Jack well be submitted to both AITF Victoria and the University of Alberta, with known standards. Note that the sample submitted to Zymax was only analyzed for isotope content and concentrations are not available. As a result the Zymax sample is not included on any of the plots using gas concentrations.

Another consideration in comparing the data collected between 2006 and 2010 is the different sampling methods used. Free gas samples were sampled from the casing vent when gas flow rates were sufficiently high and from a gas separator when gas flow rates were lower. On the February 20, 2008 sampling visit gas samples were taken directly from the casing vent as well as using a gas separator and submitted for compositional and isotopic analyses. These two samples gave very similar hydrocarbon concentrations and isotopic signatures suggesting that the two different sample types give comparable results.

The Jack water well has methane isotope signatures that fall within the range indicative of primarily biogenic gas (Figure 3). The samples from energy well 100/11-18-078-07 W6M and 100/06-12-078-08 W6M have more positive methane  $\delta^{13}$ C values and a larger percentage of higher order carbon gases (C2+) (Figure 3). The relative proportion of methane to ethane and propane can change as gases migrate, due to the preferential adsorption of higher order hydrocarbons in organic units.

Biogenic gases typically form in low-rank coal (peat to sub-bitumous). They are generated in shallow (<3 km) low-temperature (<100°C) rocks by anaerobic bacteria and are often characterized by low concentrations of C2+ hydrocarbons and isotopically light methane ( $\delta^{13}$ C < -60 ‰) (Jenden, 1993). Thermogenic gases are typically generated from the thermal degradation of kerogen at greater depths and temperatures. The geochemistry of thermogenic gases will depend on the source rock character and maturity, but they typically have a larger proportion of C2+ hydrocarbons and more positive  $\delta^{13}$ C composition of methane (Jenden, 1993).

Based on knowledge of the regional stratigraphy of the area, potential sources of hydrocarbon gases to shallow groundwater would include the following: (i) Charlie Lake Formation, (ii) shale or coal units (e.g. Paddy, Cadotte, or Falher Formations), or (iii) shallow biogenic methane produced in the soil zone.

The concentrations of methane and ethane for the energy and water wells seem to indicate three mixing end-members (Figure 4). These end-members include a) a methane-ethane poor end-member associated with natural recharge and/or injected water, b) a methane-rich, ethane-free end-member consistent with shallow biogenic gas, and c) a mixed methaneethane end-member attributed to the Charlie Lake Formation. Some of the water injection wells have very low concentrations of both methane and ethane, which is consistent with the practice of storing water for injection open to the atmosphere. These waters would degas resulting in low concentrations of both gases. Shallow groundwater is characterized by higher methane concentrations than ethane. Note also that the Cadotte Formation and Jack well samples cluster near the biogenic gas end-member but are distinguished in most cases by a measurable ethane content. Samples from energy well production casings display systematic variation in methane and ethane concentrations. Two energy wells fall on the mixing line between Charlie Lake Formation and the gas-free end-member possibly indicating gas dilution by water injection. Energy well surface casing vent samples (SCV) tend to plot along mixing lines between the Charlie Lake Formation and shallow biogenic gas end-members, along a mixing line with the gas-free end-member. This is attributed to dilution of gas concentrations by recharge water. While it is known that methane:ethane ratios can be sensitive to migration Fig. 4 provides a first look at the ratios in the different end-member waters available for mixing.

The Jack water well is characterized by very high concentrations of methane and fairly negative  $\delta^{13}$ C compositions (Figure 5). Some of the energy wells have similar concentrations of methane, but their  $\delta^{13}$ C is far more positive, plotting in the range expected for thermogenic gases (see also Figure 3). The isotopic composition of methane in the Jack water well (Figure

3 and 5) indicates a primarily biogenic source. The Jack well is located in an area where lowpermeability lacustrine clays overly low permeability Smoky Group deposits (Hackbarth, 1977). The expected water yield for these deposits is quite low and there are few water wells in the area available to obtain information about local background water quality (Hackbarth, 1977) or gas concentrations. A recent survey of shallow groundwater and water from the major coal bed methane producing formations in central Alberta can provide some context for comparison (Cheung et al., 2010). We have included average values for shallow groundwater, the Horseshoe Canyon/ Belly River Group (HSC/BRG) and Manville Formations from the Cheung et al. (2010) dataset for central Alberta, plotted with one standard deviation denoted by the error bars in this study (Figures 5, 6, and 7). The Jack well is located just north of the region included the Cheung et al. (2010) study, so these ranges may not be entirely comparable because of differences in geology. However, in the absence of any local data on the composition of gases in shallow groundwater, these data provide a general idea of the range of gas concentrations and compositions sampled in shallow groundwater in the Central Plains region of southern Alberta.

The multiple sources of mixing for methane (multiple methane producing units at depth, soil zone biogenic methane, etc.) make it difficult to use methane alone to identify the origin of gases. The isotopic composition of higher order hydrocarbons like ethane and propane are generally preferred for identifying potential gas sources (Rowe and Muehlenbachs, 1999). The isotopic composition of ethane sampled from the Jack well is far more positive than the range typically found in shallow groundwater samples from central Alberta and more closely resembles ethane isotope compositions from production casing and SCV gases (Figure 6). This similarity suggests that the ethane in the Jack well has the same source as the ethane sampled in the production casing and SCV.

The  $\delta^{13}$ C values measured for methane and ethane in the Jack water well and surrounding energy wells seem to suggest a mixture of biogenic and thermogenic gases in the well (Figure 7). The methane isotopic composition is within the range of what one would expect for biogenic methane, whereas the ethane composition is similar to some of the samples from water injection wells, and energy well SCV. Average values and one standard deviation for the data collected by Cheung et al. (2010) for central Alberta are included again for comparison. The same  $\delta^{13}C_{methane}-\delta^{13}C_{ethane}$  plot using the data from the soil gas migration study (Figure 7, bottom panel) is illustrative of the large variability one can encounter in methane even within small areas. The bottom panel of Figure 7 includes the production casing and SCV samples for the energy wells located at 100/06-12-078-08 W6M and 100/11-12-078-08 W6M plotted with the soil gas samples obtained from the immediate vicinity of those wellheads (0.3m to 2m away from well head). Both of these wells had uncemented sections and SCV flows. High gas concentrations were consistent with leakage from the wells; however even in the immediate vicinity of the wellhead the isotopic composition of soil gases were often quite different from the production casing. The isotopic composition of soil gas samples more closely resembled the composition of SCV flows.

Time series of the isotopic composition of gases from the Jack water well and 100/06-12-078-08 W6M can be used to evaluate the variability of the isotopic labeling of these different

sources and to identify trends in their composition. (Figure 8) There appears to be a trend of isotopic enrichment of about 2‰ over a 4 year period for all the hydrocarbon gases in the 100/06-12-078-08 W6M energy well (Figure 8). This may be a natural fractionation process in the reservoir/well whereby the isotopically lighter isotopes hydrocarbons have less mass and can move more quickly. The time-series of ethane  $\delta^{13}$ C composition shows the similarity between the isotopic labeling of ethane sampled from the energy well and ethane sampled from the Jack water well.

After the remedial cementing was completed at energy well at 100/06-12-078-08 W6M the  $\delta^{13}$ C value measured for methane from the Jack water well became slightly more negative (Figures 3, 5, 7 and 8) consistent with the post-cementing sample having a lower percentage of thermogenic methane. However, the post-remedial cementing sample was analyzed at a new laboratory (AITF Victoria) and an inter laboratory comparison with results from the University of Alberta should be performed to determine if the shift to more negative values is real, or simply a result of differences in laboratories.

The carbon isotope value of hydrocarbons from the Jack water well were compared to energy wells in the same section (100/06-12-078-08 W6M and 100/11-12-078-08 W6M) that have surface casing vent flows and have been shown to have gas migration issues (Figure 9). Shaded horizontal zones have been drawn from the range in  $\delta^{13}$ C signatures in the Jack water well across energy well fields to facilitate comparison. The carbon isotope ratio of methane in the Jack water well (average -66‰ PDB) is more negative than any of the energy wells sampled and is within the range typical for biogenic methane. The carbon isotope ratio of ethane in the Jack water well is similar to that found in the surface casing vent of 100/06-12-078-08 W6M and 100/11-12-078-08 W6M.

The energy well located at 100/11-18-078-07 W6M was included in the 2009 sampling because it is the water source well that supplies water for the 100/06-12-078-08 W6M injector well. The carbon isotope ratio of ethane from this water supply well is > 2‰ more positive than the range of  $\delta^{13}$ C values measured for ethane in the Jack well. The production casing sample for this well was the only well with a similar propane carbon isotope composition as the Jack well (Figure 9). Concentrations of propane in the Jack water well were generally low (2 to 138 ppm) which makes accurate determination of the carbon isotope ratio more difficult.

The hydrocarbon gas composition and isotopic values from the 2008 sampling of the Jack water well were used to calculate hypothetical mixing scenarios (using equations in Jenden et al. 1993) that could explain both the composition and isotopic signature of gases in the well (ARC, 2008). The model assumed mixing of biogenic and thermogenic end-members with the following characteristics:

- biogenic end-member
  - composition: methane = 999,999 ppm, ethane = 1 ppm
  - isotopic composition  $\delta^{13}C_{\text{methane}} = -65.5 \,\%, \, \delta^{13}C_{\text{ethane}} = -30.8 \,\%$

- thermogenic end-member
  - composition: methane = 838,000 ppm, ethane = 105,300 ppm
  - isotopic composition:  $\delta^{13}C_{\text{methane}} = -50.7 \text{ }$ %,  $\delta^{13}C_{\text{ethane}} = -31.1 \text{ }$ %

Using the original end-members the gas concentrations and isotopic composition of gas sampled from the Jack water well in 2008 could be explained by mixture of 2% thermogenic gas with 98% biogenic gas. This calculation was useful for demonstrating that a gas can still plot in what has traditionally been thought of as the biogenic field on a plot of C1/C2+ vs.  $\delta^{13}$ C methane (upper left section of Figure 3) even if it has a small component of thermogenic gas present. There are however, a number of assumptions that go into this calculation including:

- the gas is a mixture of only two sources,
- the concentrations and isotopic compositions of the two end-members are known and values are constant with time.

The range in concentrations (Figure 4) and isotopic signatures (Figures 3 and 5, 6 and 7) measured in SCV gases, production casing, water injection wells and soil gases suggest that it is unlikely that the gases available for mixing can be characterized by just two endmembers. There is also some uncertainty about the value used to characterize the  $\delta^{13}C_{methane}$  signature of the biogenic end-member. It was set at -65.5 ‰ in the original mixing calculation so that the mixing curve would pass through the data from the Jack water well (ARC, 2008). The shift to a more negative  $\delta^{13}C_{methane}$  signature in gas from the Jack water well in the 2010 data indicates that the biogenic end-member signature should be more negative. The current investigation has also revealed that other formations, like the Cadotte, should be considered as potential sources of gas for mixing. The characteristics of the Cadotte Formation (using average values from 100/11-18-078-07 W6M, n=3):

- Composition: methane = 862,333 ppm, ethane = 3983 ppm
- Isotopic composition:  $\delta^{13}C_{\text{methane}} = -43.7 \text{ \omega}, \delta^{13}C_{\text{ethane}} = -26.5 \text{ \omega},$

are quite different than the biogenic and thermogenic end-members used in the original mixing model calculations.

The increases in the concentrations of ethane and propane measured in the 2010 sample from the Jack water well suggest a small increase in the calculated percentage of thermogenic gas present in the Jack well, but the shift to a more negative  $\delta^{13}C_{methane}$  signatures in the 2010 could indicate a smaller percentage of thermogenic gas. While the thermogenic contribution is likely small, it is our opinion on consideration of all evidence that the source of thermogenic gas is not well defined, and it is therefore unwise to quantitatively apply a simple mixing model as previously described. Repeat sampling and isotopic analyses of gases from the Jack water well is recommended to confirm the negative shift in  $\delta^{13}C_{methane}$  and the increases in C2+. Sampling of the other gas-bearing geological formations is also recommended prior to application of any mixing model.

# t are highlighted in yellow.

	2. Julinia	y or guo u	Dealer 7	Donth	Status	Sample Location	Lab	Consultant	Onnor	(1, ) ) - · · · ·	/0/5
ignation	WELL ID	LSD	Pool or Zone	interval (m)	Gialus		comp/isotope			(d-m-y)	(%)
		0141 40 070 00 14/014	Creaky Croup	47.2 - 54.9	Water well, activ	9	AGAT	Matrix	Bruce Jack	7/2/2005	2.3
k Well	1820001	SVV-12-078-08 VV0fVI	Shoky Group	=7.2 - 04.0		Headspace	GCHEM	GCHEM		16/5/2006	
						Headspace	GCHEM	GCHEM		19/5/2006	
						Headspace	GCHEM	GCHEM		19/5/2006	6.2
			*			Well, after purge	Maxxam/UofA	Matrix	-	19/10/2006	0.2
	-					Duplicate	Maxxam/UofA	Matrix	-	19/10/2006	7.0
	-					Well, pump on	Maxxam/UofA	Matrix	0	19/10/2006	1.1
-		*		-		Well, pump on	Zymax	Matrix		19/10/2006	
						Duplicate	Maxxam/UofA	Matrix		19/10/2006	6.9
						Well	ARC, Veg/Uof/	ARC/AEN	V -	20/2/2008	2.8
	*	*				Duplicate	UofVictoria	ARC/AEN	V	20/2/2008	
		*		-		Casing vent gas	ARC, Veg/Uof/	ARC/AEN	-	20/2/2008	4.8
			*			Duplicate	UofVictoria	ARC/AEN	-	20/2/2008	
		*				Free gas	AITFVeg/AITF	IC AITF		23/05/2010	7.5
rgy Well	100/06-05-078-07 W6M	06-05-078-07 W6M	Charlie Lake Fm	1580.0-1581.5	Pumping oil	SCV	AGAT	Matrix	Petrofund Energy Trus	23/09/2005	0.5
rgy Well	100/12-05-078-07 W6M	12-05-078-07 W6M	Charlie Lake Fm	1547.0-1556.0	Flowing gas	SCV	AGAT	Matrix	Petrofund Energy Trust	23/09/2005	10.
rgy Well	100/06-06-078-07 W4M	06-06-078-07 W4M	Charlie Lake Fm	1618.0-1635.0	Pumping oil	SCV	AGAT	Matrix	Petrolund Energy Trus	23/09/2005	0.4
rgy Well	100/08-06-078-07 W6M	08-06-078-07 VV6M	Charlie Lake Fm	1592.0-1600.0	Pumping oil	SCV	AGAT	Matrix	Petrolund Energy Trus	23/09/2005	4.8
rgy Well	100/14-06-078-07 W6M	14-06-078-07 VV6M	Charlie Lake Fm	1555.0-1558.0	Pumping oil	SCV Destruction Consistent	AGAT	Matrix	Petrofund Energy Trus	23/09/2005	0.2
rgy vveli	100/04-07-078-07 996191	04-07-078-07 VV6IVI	Charlie Lake Fm	1553.5 - 1550.5	Pumping oil	Production Casing	Maxam/UotA	Lionnead	Trico	20/09/2006	3.0
In Alon	100/06 07 079 07 14/014	06.07.078.07 14/644	Charlie Lake E-	1542 0 4545 0	Water Inight	Production Casing	Maxxam/UofA	Maxxam	I FICO	17/10/2006	2.2
gy well	100/08-07-078-07 WOM	08-07-078 09 M/CM	Charlie Lake FM	1543.0-1545.0	Purer Injection	SCV	AGAT	Matrix	Petrofund Energy Trus	23/09/2005	0.2
gy Well	100/14-07-078-07 14/644	14-07-078-09 WOM	Gething Em	1332 0 1225 0	Flowing Cos	SCV	AGAT	Matrix	Petrofund Energy Trust	23/09/2005	1.3
TTY Well	100/11-18-078-07 W6M	11-18-078-07 WGM	CadotteEm	830.0 - 855.0	Water Source	Production Cocing	AGAT	APC	Peroluna Energy Irus	10/02/2005	0.0
av Well	100/11-18-078-07 WeM	11-18-078-07 W6M	CadotteEm		Water Source	Production Tubing	ARC, Vegruois	ARC	PennWest	19/03/2009	16.
uy Well	100/11-18-078-07 W6M	11-18-078-07 W6M	CadotteEm		Water Source	Exclued nas	ARC Vegroop	ARC	PonniAlost	19/03/2009	0.1
gy Well	100/14-01-078-08 W6M	14-01-078-08 W6M	Charlie Lake Em	1592.0 - 1594.5	Pumping oil	SCV	AGAT	Matrix	Pertofund Energy T	13/03/2009	3.0
	н	"	a a		"	SCV	MaxamillefA	Liophood	Petrofund Energy Trust	23/08/2005	3.0
-				*		SCV	Maxam/LofA	Maxwarr	Northstar	17/10/2000	2.8
	*					Production Casing	Maxram/LofA	Liophead	Northstar	22/00/2000	4.5
			*	-		Production Casing	Maxxam/LlofA	Mayyam	Northstar	17/10/2000	3.0
gy Well	100/16-01-078-08 W6M	16-01-078-08 W6M	Charlie Lake Fm	1563.2 - 1566.5	Pumping oil	Production Casing	Maxxam/UofA	Lionhead	Northstar	20/00/2006	2.8
				*		Production Casing	Maxxam/UofA	Maxxam	Northstar	18/10/2006	2.1
gy Well	102/16-01-078-08 W6M	16-01-078-08 W6M	Charlie Lake Fm	1507.8 - 1510.8	Flowing oil	Production Casing	Maxxam/UofA	Lionhead	Trico	22/00/2006	2.1
-						Production Casing	Maxxam/UofA	Maxxam	Trico	18/10/2006	0.2
gy Well	100/08-11-078-08 W6M	08-11-078-08 W6M	Charlie Lake Fm	1569.0 - 1575.0	Pumping oil	Production Casing	Maxxam/UofA	Lionhead	Northstar	22/09/2006	4.8
14/ 11	100/15 11 050 00 1001					Production Casing	Maxxam/UofA	Maxxam	Northstar	17/10/2006	2.7
gy vveil	100/15-11-0/8-08 W6M	15-11-078-08 W6M	Charlie Lake Fm	1563.5 - 1565.5	Pumping oil	Production Casing	Maxxam/UotA	Lionhead	UEI	11/10/2006	3.9
ma Mall	100/46 44 070 00 14/014	10.11.070.00110011				Production Casing	Maxxam/UofA	Maxxam	UEI	17/10/2006	2.2
gy wen	100/16-11-078-08 VV6M	16-11-078-08 VV6M	Charlie Lake Fm	1557.0 - 1562.5	Water Injection	SCV	AGAT	Matrix	Petrofund Energy Trust	23/09/2005	3.2
						SCV	Maxxam/UofA	Lionhead	Northstar	22/09/2006	4.2
						SCV	Maxxam/UofA	Maxxam	Northstar	17/10/2006	1.9
av Well	100/02-12-078-08 M/Ch4	02-12-079 00 16/01+	Chadia Line	-		Production Casing	Maxxam/UofA	Maxxam	Northstar	17/10/2006	33
11	* ************************************	"	Uname Lake Fm	1064.8 - 1566.2	Pumping oil	Production Casing	Maxxam/UofA	Lionhead	UEI	20/09/2006	2.9
	-					Production Casing	Maxxam/UofA	Maxxam	UEI	18/10/2006	2.75
	-					Production Casing	Zymax	Matrix	UEI	18/10/2006	
y Well	100/06-12-078-08 W6M	06-12-078-08 W6M	Charlie Lake Em	1570.0 1572.0	Meter Initation	SCV	GCHEM	GCHEM	Petrofund Energy Trust	11/05/2006	
			"	1370.0 - 1372.0	vvater injection	SCV	AGAT	Matrix	Petrofund Energy Trust	23/09/2005	0.18
80						SCV	Maxxam/UofA	Lionhead	Northstar	22/09/2005	5.05
						SCV	Maxxam/UofA	Maxxam	Northstar	17/10/2006	1.53
		*				Broduction Cosing	Zymax	Matrix	Northstar	17/10/2006	
-			×			SCV	Maxxam/UotA	Maxxam	Northstar	17/10/2006	35.9
101.0			*		*	SCV	GUNEM ADD	GCHEM	Pertofund Energy Trust	12/05/2006	
y vveli	100/08-12-078-08 W6M	08-12-078-08 W6M	Charlie Lake Fm	1556.0 - 1557.5	Pumping oil	Production Casing	Maxyam/LlofA	ARC	PennWest	19/03/2009	3.20
W/A/cll	100/11 12 070 00 110-1		*			Production Casing	Mayyam/ loft	Lionnead	Northslar	22/09/2005	2.32
, vven	100/11-12-078-08 W6M	11-12-078-08 W6M	Charlie Lake Fm	1557.5 - 1601.5	Flowing Gas	SCV	AGAT	Matrix	Northstar	19/10/2006	2.07
						SCV	Maxram/LofA	Liophand	Petrolund Energy Trust	23/09/2005	1.21
н						SCV	Maxcam/LofA	Mayyam	Northstar	20/09/2006	0.09
-						Production Casing	Maxam/LlofA	Lionbood	Northstar	17/10/2006	1.57
Well	PTF 11C-12-078-08\A/6	w				Production Casing	Maxxam/Llof4	Maxxam	Northstar	20/09/2006	36.3
Well	102/11-12-078-08 W/6M	11-12-078-08 \A/614	Charlie Lake E	500 4 10		SCV	GCHEM	GCHFM	Petrofund Enormy Text	1//10/2006	34.7
	"	"	unanie Lake Fm 1	586.4 - 1588.4	Pumping oil	SCV	AGAT	Matrix	Petrofund Energy Trust	11/05/2006	
			*			SCV	Maxxam/UofA	Lionhead	UEI	23/09/2005	0.09
u.		*	× .		-	SCV	Maxxam/UofA	Maxxam	UEI	17/10/2006	17.01
		*				SCV Duplicate	Maxxam/UofA	Maxxam	UEI	17/10/2006	20.81
				-		Production Casing	Maxxam/UofA	Lionhead	UEI	20/09/2006	2.40
Well	PTF 11D-12-078-08W6					Froduction Casing	Maxxam/UofA	Maxxam	UEI	17/10/2006	3.12
Well 1	100/16-12-078-08 W6M	16-12-078-08 W6M	Charlie Lake Fm 1	538.0 - 1540 3	Water Inigation	SCV	GCHEM	GCHEM	Petrofund Energy Trust	11/05/2006	1.69
				"	"	SCV	AGAT	Matrix	Petrofund Energy Trust	23/09/2005	0.14
				-		SCV	Maxxam/UofA	Lionhead	Northstar	20/09/2006	2.65
						Production Casing	Maxxam/UofA	Maxxam	Northstar	17/10/2006	1 72
					-	Production Casing	Maxxam/UofA	Lionhead	Northstar	20/09/2006	26.82
Woll 1	00/02 12 070 00 1000					Prod Casing Durbaste	Maxxam/UofA	Maxxam	Northstar	17/10/2006	13 53
vven 1	00/02-13-078-08 W6M (	J2-13-078-08 W6M	Charlie Lake Fm 1	536.0 - 1542.0	Pumping oil	SCV	Maxxam/UofA	Maxxam	Northstar	17/10/2006	13.33
					" "	SCV	AGAT	Matrix	Petrofund Energy Trust	18/01/2005	0.57
						SCV	AGAT	Matrix	Petrofund Energy Trust	23/09/2005	8 44
						Production Cooling	Maxxam/UofA	Lionhead	Тгісо	22/09/2005	12 25
Well 1	00/03-13 079 00 11001				-	Production Casing	Maxxam/UofA	Lionhead	Trico	22/09/2005	3 20
**CH 1(	" " " " " " " " " " " " " " " " " " "	3-13-078-08 W6M	Charlie Lake Fm 15	43.0 - 1546.0	Pumping oil	Production Casing	Maxxam/UofA	Maxxam	Trico	17/10/2006	2.00
Nell 40	00/04 12 078 00 14/014	"			" "	Production Casing	Maxxam/UofA	Lionhead	Northstar	22/09/2005	6.96
1011	00/02-14-078-08 W6M 0	4-13-078-08 W6M	Charlie Lake Fm 15	44.7 - 1547.0	Flowing oil	Production Casing	Maxxam/UofA	Maxxam	Northstar	17/10/2006	2.20
Noll 11	00/02-14-078-08 W6M 0	2-14-078-08 W6M	Charlie Lake Fm	directional	Pumping oil	Production Casing	Maxxam/UofA	Lionhead	JEI	22/09/2005	3.28
Well 10				anoononu						Bull of Sector and Sector Sect	C 4 3
Well 10					"	Production Casing	Maxxam/UofA	Lionhead I	Northstar	22/09/2005	4.40
Nell 10	00/16-14-078-09 \A/4A4				" "	Production Casing SCV	Maxxam/UofA Maxxam/UofA	Lionhead I Maxxam I	Northstar Northstar	22/09/2005	4.46
Vell 10 Vell 10 Vell 10	00/16-14-078-08 W4M 11	8-14-078-08 W4M (	" Charlie Lake Fm 15	541.0-1544.0	Flowing oil	Production Casing SCV	Maxxam/UofA Maxxam/UofA GCHEM	Lionhead I Maxxam I GCHEM F	Northstar Northstar Petrofund Energy Trust	22/09/2005 17/10/2006 12/05/2006	4.46

)	(mag)	(ppm)	(pom)	) (pr	(D) (D	om) (r	(mag	(mag	(ppm)	(npm)	(%)	(%)	(%)	(%2)	(%)	(%)
	973300	1200	<100	<1 (P)	00 1	00 <	100 <	100	100	300	(100)	(100)	1100/	(100)	( ~~ )	(100)
	843548	782.5	22	0 1	1 0	01 0	34 0	.07		500						
	790343	773.3	2.5	0.1	8 0.	34 2	.81 0	.47								
	933130	929.5	3.6	0.3	3 0.	58 0	.55 0	.34								
	933200	<100	7	<1	00 <1	00 <	100 <	100	<100	<100	-34.47	-65.90	-30.74			
	922300	<100	<100	<1	00 <1	00 <	100 <	100	<100	<100	-33.87	-64.97	-30.53			
	915200	<100	<100	<1	< 00	00 <	100 <	100	<100	<100	-25.10	-65.80	-30.36			
											-0.30	-69.3	-32.2			
	925100	<100	<100	<1	< 00	00 <	100 <	<100	<100	<100	-34.66	-65.57	-30.24	-23.05		
	848000.0	1910.0	14.1	0.0	0.	0					-37.59	-65.48	-29.75	-24.6		
										- 170		-05.8	-30.0	-22 79		
	818000.0	1830.0	18.0	0.4	478 0.	045 0	0.015	0.001	0.000	0.172		-03.44	-29.6	-24.2		
									0.05	-0.05	26 50	-05.7	-301			
	0.000038	4790.0	138.0	0 <0	.05 <	0.05	<0.05	<0.05	<0.05	<0.05	-20.50	-00.4	-50.1			
)	894300	52300	2480	0 41	00 5	500	1500	1100	800	1300						
)	874600	8900	1600	) 70	0 4	00 3	200	200	400	900						
)	873200	67700	3130	0 50	000 6	700	1800	1300	800	900						
)	841500	62800	2750	00 49	900 6	600	1900	1300	1100	2400						
)	893600	63000	2370	00 42	200 5	900	1800	1400	1000	1100		45.51	-35.26	-32.48		-32.98
1	791600	94700	3960	00 63	200 1	1400	2600	2400	1600	1500		45.55	-35.09	-32.34	-33.00	-31.65
0	822400	86600	3600	00 53	200 9	500	2200	2300	2100	3500		-40,00				
0	858800	73500	395	00 6	900 9	700	2800	2100	1400	1000						
0	853300	71100	366	00 6	600 8	3700	2300	1600	1000	800						
0	832200	50200	289	00 6	500	0300	2900	2200	1/00	1500	8.60	-43.60	-26.84	-24.44	-29.61	-22.93
0	745000	3730	154	3	5.1	39.7					5 30	-43.49	-26.27	-26.33	-27.08	-27.75
•	945000	4480	176	5	5.7	93.9					0.00	-44.01	-26.40	-26.61	-25.76	-28.62
	897000	3740	133	3	0.4	31.8		1000	1200	200						
0	834700	67400	370	000 6	100	9700	2800	1900	1200	740		-51.31	-31.05	-27.92	-28.09	07.07
1	837700	69800	342	200 5	000	9400	2500	1800	1100	600		-49,96	-30.94	-27.65	-28.71	-27.87
1	819300	67500	0 350	000 5	5900	9200	2400	1600	1100	4500	n	-45.05	-35.40	-32.94	-33.02	-32.50
	773400	8970	0 480	000	9900	17600	5800	6300	3900	1500	0	-45.62	-35.33	-31.99	-32.89	-31,51
, 1	839000	8090	0 324	400	4600	6900	1200	1100	400	200	0	-44.76	-35.19	-33.03	-32.25	-32.15
0	713100	1310	00 76	200	11700	19500	4000	3500	2/00	220	0	-45.06	-35.39	-32.93	-33,51	-32.05
0	716700	1327	00 79	100	12500	21400	4400	3900	2400	<10	0 -18.13	-46.79	-28.72	-23.11	-30.05	-30.71
~	956300	2500	8		30	6	<100	<100	1200	600	-17.68	-46,10	-35.02	-31.72	-31.09	-20.11
000	841900	7850	0 30	000	2700	7800	1300	1400	440	200		-44.84	-34.63	-32.61	-31.95	-30.01
0	828100	6870	0 27	100	5300	9400	1500	1200	300	<10	0	-44.91	-34.53	-31./3	-33.05	-31.04
0	858700	6880	0 25	200	4600	0008	1500	8700	80	133	00	-45.54	-35.32	-32.11	-32.33	-31,24
0	699400	1131	100 61	000	12700	23100	5600	6200	470	290	00	-45.40	-35.23	-32.06	-32.14	C 1 10 1
0	740500	1187	700 62	400	12000	20700	3000	2000	120	900	)			27 56	-28 91	-27.79
800	843600	6260	00 35	500	5700	0088	2700	1000	160	0 530	00	-50.15	31.16	-27.50	-29.42	-28.33
0	823600	6370	00 33	3200	5400	0018	2000	2100	150	0 110	00	-50.34	31.21	-21.70	-31.64	-30.31
00	855400	656	00 34	4400	5600	8600	13400	0000	940	0 19	500	-46.79	-35.00	-34./1	-34 77	-32.72
00	220000	684	00 79	9400	17800	32100	2500	2700	100	0 41	00	-45.52	-34.89	-34.10	-32.92	-31.53
40	815100	848	00 3	7900	5800	9900	2500	2200	190	0 22	00	-45.46	-35.27	-32.15	-32.50	
00	820900	871	00 3	8400	5500	8800	2100	2201			-9.90	-45.40	-35.60	-31.00	02.00	
~~						74 40	27.74	13 1	8							
	81318	107	0 2	96.8	48.07	71.18	21.11	200	0 120	0 70	0		00.04	20.28	-30 07	-29.71
400	86330	D 714	100 3	8200	6300	9500	2900	200	0 110	0 54	0	-52.63	-33.01	-28.04	-28.95	-28.33
70	80890	0 692	200 3	5600	5900	9300	2000	210	0 14	00 80	00	-49.95	31.30	-27.00	-28,50	
00	85200	0 709	900 3	36400	5900	9200	3000	210			-24.0	0 -51.50	31.00	-32.91	-31.69	-28.79
						20000	1550	0 131	00 11	700 14	4700	-65.30	-30.09			
100	22910	0 563	300 8	85100	19100	35800	2080	134	9				10.00	-27 32	-28.21	-28.35
100	81282	1 72	138	34673	5767	9048	2980	1.54	-			-47.86	25.42	-33.33	-33,67	-32.40
744	79300	0 70	500	36300	7640	10/00	102	00 10	500 90	4	80	-44.94	-35.42	-32.88	-33.28	-31.37
430	66510	0 13	1700	84100	16900	33000	220	0 150	00 50	0 1	00	-45.05	-35.55	-02.00		
400	72320	0 14	0700	78300	11000	7000	220	0 14	00 70	0 4	00		30.96	-28.13	-28.00	
27200	8594	00 55	200	29100	4700	2200	270	0 18	00 10	9 00	20	-50.04	.31.21	-27.67	-28.94	-28.34
60	8737	00 70	900	35800	3300	0000	290	0 20	00 14	100 1	600	-50.15	-35.53	-32.69	-33.56	-31.60
<100	8537	00 69	9300	36100	5900	9000	117	00 11	100 90	) 4	200	-46.87	-36.08	-32.93	-28.63	-33.22
90	3362	00 87	7200	83600	19900	20000	111	00 10	400 5	300 3	3500	-47.21	- 30.00			
<100	3491	00 89	9200	91100	19800	5028	143	2 55	3.1							0.5.50
	6623	12 48	8344	22196	54/0	8400	260	0 18	00 1	000 9	900	10.68	-30,78	-28.40	-28.87	-28.50
400	8763	00 6	7000	34700	4200	6400	190	00 13	300 7	60 9	930	-49.00	-31.60	-27.47	-28.57	-27.01
140	6876	500 5	3400	26600	4200	6300	190	00 14	8 004	00	300	-01.41	-31.42	-28.12	-29.31	-28.47
<100	6468	300 4	9400	24600	4000	0000						-51.10	-35.04	-32.75	-32.57	-32,10
				10110	21400	38800	93	00 91	100 9	0	7300	45.89	35.09	-33.09	-33.32	-31.90
170	5619	900 1	79700	121100	10200	32900	89	00 8	100 7	900	5600	-40.00				
<100	582	800 1	92900	12130	5776	8826	29	19 1	280		000				07.00	27 92
-	837	484 7	2010	34694	6200	9400	28	00 1	900	200	500	-51.26	-31.08	-27.61	-27.99	-27.76
700	877	500 E	52300	35500	5800	8900	25	00 1	600	000	500	-52.01	-31.31	-27.72	-28.98	-31.83
100	847	900 6	54100	34300	5100	8300	27	00 1	900	1100	1200	-45.38	-33.06	-31.65	-30,00	-29.68
<100	876	600	55400	25000	1400	2200	11	00 8	30	940	1200	-44.74	-32.37	-32.18	-29.21	-27.92
270	935	00	3000	2000	1700	3300	15	900 1	500	1500	1200	5.19 -43.24	-32.49	-31.23	-20.44	
<100	726	000	2000	2000	1100					200	200					
			00333	34400	5100	6400	14	400 8	300	300	400			07.00	-20.84	-28.51
700	878	5200	61900	35800	6500	9600	2	700 1	1800	1000	340	-51.85	-31.12	-27.99	-32 79	-31.35
2600	786	5000	61800	29804	5200	7700	2	000	1400	850	4100	-45.07	-35.06	-32.47	-32.10	-31.59
100	74	2800	102600	51404	9300	1800	0 5	300	5300	40	800	-45.34	-35.68	-32.70	-32.57	-31.86
1190	75	9000	103000	3260	5400	9100	2	000	1700	200	5200	-45,28	-35.62	-33.14	-33.80	-31.93
420	81	5800	92500	4220	0 7000	1260	0 3	100	3000	30	1800	-45.24	-35.29	-32.23	-32.67	-30.90
420		2500	91900	4220	0 8600	1610	0 6	500	12000	2000	500	-45.41	-35.68	-32.03	-32.80	-31.57
420 300 390	75		00500	8010	U 00			000	TRUE	-2012	000		25.02	-32.39		
420 300 390 400	75 76	5100	99500	4910	0 7100	) 1170	00 4	200	1000	40	3100	-45.44	-35.04	2262	-31.61	-31.81
420 300 390 400 80	75 76 78	5100 8800	99500	4910	0 7100	) 1170	00 4	1300	4400	40	3100	-45.44	-35.51	-32.63	-31.61	-31.81
420 300 390 400 510	75 76 78 74	5100 8800 5700	99500 109400 10680	4910 0 4750 0 5130 0 5980	0 7100	0 1170 0 1640 00 1750	00 4 00 4	4300 5100	4400	40 5100	3100 4500	-45.44 -45.43	-35.51	-32.63	-31.61	-31.81

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#### 3.2 Energy Well Soil Gas Migration Sampling

The gas migration data are presented in Table 3, Figures 3 (bottom panel), 7 (bottom panel), 10 and 11 and relevant analytical reports are provided in Appendix B along with sample location maps. The original May 31, 2009 gas migration sampling by Lionhead Engineering was deemed invalid by the ERCB due to excessively moist soil conditions. This data is included in Table 3 for completeness but the values are shown in grey to indicate that they have been excluded from this analysis. Most of the energy wells had evidence of thermogenic gas migration near the wellhead. Gas concentrations decreased dramatically within about 4 m from the wellhead. The spatial and temporal variability in gas concentrations are likely related to sampling variability between companies, seasonal differences, soil moisture differences, and laboratory error.

The carbon isotope ratio of methane soil gas around the energy wells has a more enriched (more positive) isotope signature than the methane gas found in the Jack water well (-65.6% PDB) (Figures 3 and 7). The gas composition and isotopic signatures of soil gas sampled in the vicinity of 100/11-12-078-08 W6M have a thermogenic signature (Figure 3). Soil gas sampled in the vicinity of 100/06-12-078-08 W6M appears to reflect a mixture of thermogenic ethane with biogenic methane. The carbon isotope value of ethane from the gas migration study sampling conducted around 100/06-12-078-08 W6M is close (1 to 2 ‰ difference) to that found in the Jack water well (Figure 7 bottom panel).

Soil vapour samples in the vicinity of 100/06-12-078-08 W6M were also taken after the remedial cementing of the energy well on May 13, 2010, and September 22, 2010 (Figures 7 and 11 and Table 3). The May and September 2010 post-cementing samples revealed decreases in gas concentrations for most sampling points. Gas concentrations and isotope compositions for two post-cementing sampling visits are presented on Figure 11. Gas composition (but no isotope) data for an additional sampling point from the September 22, 2010 visit is included in Appendix B. Samples taken 0.3 m from the wellhead had methane concentrations that ranged from 64,900 ppm to 326,000 ppm pre-remedial cementing. Post-cementing gas samples taken 0.3 m from the well head ranged from 54,500 ppm on May 13, between 89,000 ppm and 117,400 ppm on September 22, 2010 and then decreased to 0% LEL on the November 11, 2010 visit. Some of the post-cementing  $\delta^{13}$ C values for methane and ethane in soil gas showed a shift towards more negative values, consistent with a reduced presence of thermogenic gas. The gas migration tests performed on November 11, 2010 measured 0% LEL at the wellhead and all other measurement points in the field screening indicating that there is no longer any gas migration (Lionhead, 2010).

#### 3.3 New Energy Well Gas Mudlog

A new energy well (located 5.6 km SE of the Jack water well) was drilled and completed at 100/02-04-078-07 W6M on October 30, 2009. Gas samples were collected to a depth of approximately 925 m (Spirit River Group). A summary of the gas composition and isotope

data is presented in Table 4 and in Figure 12 (also included on Figure 7 and 9). Raw compositional and isotopic data is presented in Appendix C.

These new data provide better characterization of the isotopic signatures of the Cadotte Formation and show that the  $\delta^{13}C$  of ethane in this unit ( $\delta^{13}C = -29.3\%$ ) is similar to the ethane gas found in the Jack water well (mean value  $\delta^{13}C = -30.1$  %) and the surface casing vents of several energy wells (Figure 7). These data are also consistent with a mud log from an energy well located approximately 65 km to the south (Muehlenbachs et. al. 2000). The method of identifying the source zone of a migrating gas using the gas compositions from mudlogs typically uses a match for ethane, propane and butane (see Rowe and Muehlenbachs, 1999). In the case of the Jack well, there was only a match for ethane. The only time propane was present in sufficient concentrations for isotope analyses was on the February 20, 2008 sampling visit. The  $\delta^{13}$ C signatures for the four gas samples from the Jack water well taken during this visit (average  $\delta^{13}C = -23.66$  ‰) were more positive (~7 ‰) than that in the Cadotte Formation  $\delta^{13}$ C estimate obtained from the mudlog (average  $\delta^{13}$ C = -30.89 ‰). Butane was never present at high enough concentrations in the Jack well for isotopic analyses. The poor agreement between the propane isotope compositions could be due to analytical difficulties at low concentrations (propane was never greater than 100 ppm in gases from the Jack well). A match with multiple hydrocarbon isotopic signatures (e.g. ethane, and propane) between the gases measured in the Jack well and a geological unit would provide more certainty for the identification of a source depth.

#### 3.4 Cement Integrity Investigations

The original neutron density log for energy well 100/6-12-078-08 W6M indicated gas was potentially entering from the Paddy (~850 mKb), the Shaftsbury (595 mKb), the Dunvegan (~500 mKb) and some thin gas bearing zones up hole. Following the request of the ERCB, new radial bond and neutron density logging was performed on January 19, 2010. Penn West decided to perform remedial cementing on the injector well at the same time. The remedial work consisted of cementing the perforations in the injection zone (Charlie Lake) and perforating and attempting a cement circulation to surface. The new radial bond logs collected for 100/06-12-078-08 W6M indicated that as of January 2010 the cement top is at 490 mKb. Previously the cement top for this well was at about 1000 m. There is still an uncemented interval between 298mKb to 490 mKb.

Surface casing vent flow tests performed by Hotwell Canada Ltd on May 13, 2010 and September 22, 2010 have found that there was no longer any vent flow measurable after the remedial cementing of the well. Gas migration tests performed on November 11, 2010 found 0% LEL in all of their soil measurements, indicating that gas migration is no longer occurring at this energy well (Lionhead, 2010).

#### 3.5 Relationship Between Energy Well Activities and Gas in the Jack Water Well

The timing of major events (e.g. drilling, stimulation, acid treatments, remedial cement squeezes and water injection) in energy wells near the Jack water well was examined to see if they coincided with the changes to the Jack water well (appearance of sediment or gas). Figure 13 includes a Gantt chart summarizing these events, presented with manual water levels measured in the Jack well and total annual precipitation measured at the Spirit River meteorological station operated by Environment Canada (station # 3076069). The distance between the Jack water well and the energy well is noted in brackets on Figure 13.

The chronology of problems reported for the Jack water well was compiled using information from the ERCB database, AENV water well database and information supplied by Mr. Jack through his attorney (Mr. Ron Kruhlak) and presented in the report prepared by ARC in 2008. In the fall of 2005, Petrofund Energy Trust (now Penn West Energy Trust) initiated an investigation into a water well complaint by Mr. Jack regarding methane gas in his well. Mr. Jack indicated that the first change in his water well occurred on March 1, 2003 when sediment first appeared in his water. After pumping the well for about four weeks Mr. Jack observed gas started discharging from the well.

There are no energy well events that are coincident with the reported appearance of sediment and subsequent presence of gas in the Jack water well with the exception of an acid treatment of energy well 100/02-14-078-08 W6M. This energy well is located approximately 1370 m away, has no uncemented intervals and has no surface casing vent flow. There does not appear to be any relationship between water injection and the appearance of sediment and gas in the Jack water well (Figure 13). Water injection in the closest energy well (100/06-12-078-08 W6M) started about 10 months after gas was reported in the Jack water well. A hiatus in water injection in this well from September 2006 to May 2008 did not affect gas presence in the Jack water well.

Sometime in early 2009, before the site visit conducted by AITF on March 18, 2009, the behaviour of the Jack water well changed again in that the spontaneous gas eruptions stopped (personal communication with Mr. Jack). The disappearance of the eruptions occurred approximately 5 months after the tubing and packer on the injection tubing was replaced on energy well 100/06-12-078-08 W6M (August 8, 2008). This work should not have affected anything outside the production casing unless there was a hole in the production casing. Pressure testing conducted on this well on August 7, 2008 indicated that the production casing was not leaking at that time. There were water injections at a few nearby energy wells that started four to six months before the eruptions stopped (100/06-12-078-08 W6M, 100/16-11-078-08W6M, 100/16-12-078-08W6M, 100/06-07-078-07W6M).

Remedial cementing of energy well 100/06-12-078-08 W6M occurred in January, 2010 and there were a few changes in the Jack water well that occurred shortly after. After the cementing of the energy well the gas discharging to the water well decreased so that it was

no longer measurable under both pumping and non-pumping conditions (rows labeled "Gas flow rate" in Table 6). The changes in behaviour observed during the pumping tests conducted before and after the remedial cementing are discussed in more detail in section 3.8.6. Linking changes in the amount of gas present in the water well with energy well activities during the period immediately before and after remedial cementing of 100/06-12-078-08 W6M is difficult because changes in static water levels that occurred during this interval would have also affected the solubility of gases in groundwater (Figure 13).

The water level history available for the Jack well (middle panel of Figure 13) was assembled using measurements reported in the original drillers report, a water level reported in the Matrix 2006 report, and measurements made by AITF during site visits made between 2008 and 2010. The water level time-series includes two prominent trends. The first is the decrease in water levels that occurred between 2001 and 2009. Between Nov. 2001 and March 2009 water levels in the well decreased by about 12.6 m. Interestingly, this decrease occurred even though regular usage of the well stopped in 2005, and was not used at all between February 2008 and March 2009 (Mr. Jack, personal communication). Between March 2009 and May 2010 water levels increased by 2.8 m. The decrease in water levels that occurred even when the well was not being pumped indicates a regional lowering of the water table, possibly due to changes in recharge or withdrawal by other users. Total annual precipitation amounts at the Spirit River meteorological station (Figure 13, bottom panel) show fairly consistent precipitation amounts, with the exception of the drought in 2005.

#### 3.6 Lineament Study

The lineament study, covering Townships 73 to 88 and Range 2 to 13 W6M, was performed by J.D. Mollard and Associates Limited (2009). The main conclusions of the report were:

- The study area overlies a major geological structure referred to as the Peace River Arch. The Arch is a major structure in which the Precambrian granitic rocks are uplifted about 1000m and major faults cut the overlying sedimentary rocks.
- The Ft. St. Johns Grabben underlies the study area. Grabben faults extend to the base of the Triassic sedimentary rocks (deeper than all the energy wells), but late movement on the faults may have caused fracturing into the overlying sedimentary rock units.
- Bedrock in the Jack water well study area is characterized by fractures with dominant NW-SE and NE-SW orientations. A NW-SE lineament was identified through the Jack property.
- Correlation of dominant lineament and joint orientations with principal crustal stress directions suggest that bedrock fracture and/or fault systems may provide enhanced permeability pathways for the movement of fluids in the subsurface.

# Table 3: Summary of gas migration data, new data are highlighted in yellow.

WELL ID	Pool or Zone	Depth	Status	Sample Location	Lab	Consultant	Sampling Date	Nitrogen	Carbon Dioxide	Methane	Ethane	Propane	e I-Butane	N-Butan	I-C5	N-C5	C6	C7+ δ	13C CO2	5"°C Methane	δ°C Ethane	ō"°C Propane	ō"°C I-Butane	õ"C N-Butan	e Comment
		Interval (m)			comp/isotope		(d-m-y)	(%)	(ppm)	(ppm)	(ppm)	(ppm)	(ppm)	(ppm)	(ppm)	(ppm)	(ppm) (	(ppm)	(%)	(%)	(%)	(%0)	(%)	(%)	
100/14-01-078-08 W6M	Charlie Lake Fm	1592.0 - 159	4.5 Pumping oil	Soil gas, 8 m	Maxxam/UofA	Lionhead	31/05/2009		W.L.	at a g		1		11	41		20		12 - 4		Contraction of the local distance of the loc				Wet soil conditions
				Soil gas, <1 m	Maxxam/UofA	Lionhead	16/07/2009	56.53	400	279300	1200	200	100	100	<100	<100	<100	<100	-27.62	-57.86	-35.52	-29.29	-29.44	-28.20	
100/02-12-078-08 W6M	Charlie Lake Fm	1564.8 - 156	6.2 Pumping oil	Soil gas, <1 m, 1.1 m depth	GCHEM	GCHEM	11/05/2006			4218	134.3	131.2	71.37	145.1	115.9	61.55									
1				"	GCHEM	GCHEM	19/05/2006			3162	152.3	137	61,56	117.8	125.5	111.6									
1				Soil Gas, <1 m, 2.5 m depth	GCHEM	GCHEM	11/05/2006			40090	361.4	86.13	32.57	27.99	17.65	4.92									
1				Soil Gas, background 30 m	GCHEM	GCHEM	11/05/2006			43.97	3.38	1.47	0.43	0.69	1.91	0.31									
					GCHEM	GCHEM	19/05/2006			9.95	0.52	0.59	0.31	0.79	1.15	1.11									
100/06-12-078-08 W6M	Charlie Lake Fm	1570.0 - 157	2.0 Water Injection	Soil gas, <1 m, 1.1 m depth	GCHEM	GCHEM	12/05/2006			902585	7901	1175	135.7	81.38	32.19	7.44									
					GCHEM	GCHEM	19/05/2006			881436	7380	1215	135.4	81.96	30,19	10.8									
				Soil Gas, <1 m, 2.5 m depth	GCHEM	GCHEM	12/05/2006			575403	5131	803.1	103.8	64.01	0.08	7.25									
				Soil gas, background 100 m	GCHEM	GCHEM	12/05/2006			18.92	1.98	0.62	3.03	0.02	1.11	0.08									
and the second				Soil gas, <1 m	Maxxam/UofA	Lionhead	16/07/2009	72 51	1400	73300	400	<100	<100	<100	<100	<100	<100	<100	-35 22	-54 90	-36 70	-30 78	-24.85	-25.68	To all the set of
				Soil gas, 0.3 m (SVW-1)	ARC. Veg/UofA	Baseline	16/09/2009	55.20	8000	326000	2700	358	55.4	19.5					-44.87	-49.03	-32.51	-31.51	-31.30	-29.26	14/62 4
				Soil gas 0.3 m (SVW-2)	ARC Veg/UofA	Baseline	16/09/2009	71 90	4000	64900	290	29.8	62	25					-25.75	-53 14	-31.90	-31 17			
CARLE CONTRACT				Soil gas 2 m (SVW-6)	ARC. Veg/UofA	Baseline	16/09/2009	38.70	21000	580000	5750	92	11.4	0					-33 82	-49.04	-33 82				A
N 1 1 1 5 1 2 4 1 2 1				Soil gas 20.5 m (SVW-15)	ARC Veg/UofA	Baseline	16/09/2009	77 60	5000	21	0	0	0	0					-19.28						al transition of the
FUEL HIS COLD				Soil gas 0.3 m	Maxxam/UofA	Hotwell	13/05/2010	74 53	6200	54500	120	30	10	0	0	0	0	0	-47.35	-57.76					A STATE OF THE OWNER OF
a har a start a start a start				Soil gas 2.0 m	Maxxam/UofA	Hotwell	13/05/2010	77.05	7200	23700	0	0	0	0	0	0	0	60	-49.73	-58.13	-36.62				
				Soil gas 0.3 m NE of wellhead	Maxxam/UofA	Pennwest	22/09/2010	71.99	1600	89400	350	110	10	10	trace										
				Soil gas 1.8 m NE of wellhead	Maxxam/UofA	Pennwest	22/09/2010	53.89	4000	317900	2500	110	30	10	10	trace			-36.14	-54 94	-36 07	-30.21	-28.64	-24.52	A TRACK AND
				Soil gas 0.3 m SE of wellhead	Maxxam/UofA	Pennwest	22/09/2010	70.31	3600	117400	450	120	10	10	trace				-40 49	-53 39	-37.93	-33 93	-31.33		Call and States
100/11-12-078-08 W6M	Charlie Lake Fm	1557.5 - 160	1.5 Flowing Gas	Soil gas <1 m	Maxxam/UofA	Lionhead	16/07/2009	74 04	28100	31300	1200	1300	400	800	600	700	1300	1900	-31.08	-52 48	-30.93	-30 35	-28.21	-28.15	
				Soil gas 0.3 m (SVW-1)	ARC Veg/UofA	Baseline	16/09/2009	78.2	9000	717	26.6	52.3	20.9	19.5			1000		-28 73	-37 43	-25.26				2457 S 12 1
				Soil gas 0.3 m (SVW-8)	ARC Veg/UofA	Baseline	16/09/2009	79.6	27000	2540	1800	122	40.6	417					-31 84	-40.62	-27 35	-26 14	-27.03	-27.02	
CALL CONTROLS OF				Soil gas 0.3 m (SVW-8 dup)	ARC Veg/UofA	Baseline	16/09/2009	77 4	22000	12700	982	552	114	115					-27 81	-40.62	-27.91	-25.9	-26.85	-26.61	
				Soil gas 2 m (SV/W-9)	ARC Veg/UofA	Baseline	16/09/2009	78.5	2000	32	B	0.4	03	0.5					-16 98	10.02	-26.38	20.0			the good and S
1 Con 1 Con 1 Con 1 Con				Soil gas 21 m (SVW-15)	ARC Veg/UofA	Baseline	16/09/2009	78.5	2000	56	0	0	0	0					-19.09		20.00				a transmission of the second
PTF 11C-12-078-08W6	Charlie Lake Fm	1557.5 - 160	1.5 Flowing Gas	Soil gas <1 m 1.1 m depth	GCHEM	GCHEM	11/05/2006	10.0	2000	13840	967.4	540.2	98.88	150.6	73.22	29.9	-								And the second se
				Soil gas <1 m 2.5 m depth	GCHEM	GCHEM	11/05/2006			13912	417.9	169.5	23 17	34 92	27.63	9 73									
				Soil gas background 100 m	GCHEM	GCHEM	11/05/2006			84 73	6 74	7 17	1.89	3 45	3 37	1 38									
PTF 11D-12-078-08W6	Charlie Lake Fm	1586.4 - 158	8.4 Pumping oil	Soil gas, <1 m, 1.1 m depth	GCHEM	GCHEM	11/05/2006			6342	508.2	280.35	39.05	50.11	33.5	23.65									
			and a reserve a sec	n	GCHEM	GCHEM	19/05/2006			16377	662.2	371 17	45.07	51 75	35.07	12 59									
1				Soil Gas <1 m 2.5 m depth	GCHEM	GCHEM	11/05/2006			4127	282 5	151.65	21.9	33 12	17.07	11.95									
				"	GCHEM	GCHEM	19/05/2006			79314	237.2	889 15	67.69	99.24	37.89	18 72									
				Soil gas, background	GCHEM	GCHEM	11/05/2006			95.85	10 18	7.82	1 95	3 18	3.42	1.34									
				"	GCHEM	GCHEM	19/05/2006			84 73	6 74	7 17	1 89	3.45	3 37	1 38									
				Soil Gas, background 80 m	GCHEM	GCHEM	11/05/2006			21	0.07	0.13	0.04	0.01	0.53	1.55									
100/16-12-078-08 W6M	Charlie Lake Fm	1538.0 - 154	0.3 Water Injection	Soil gas 6 m	Maxxam/LlofA	Lionhead	31/05/2009		10040	6.7	0.01	0.10	0.04	0.01	0.00	1.00									Wet soil conditions
			ere trater injection	Soil das <1 m	Maxxam/LlofA	Lionhead	16/07/2009	74 29	7400	44700	1100	600	100	200	100	100	<100	<100	-34 97	-56 14	-34 54	-27 66	-29.90	-29.39	
100/02-14-078-08 W6M	Charlie Lake Fm	directional	Pumping oil	Soil gas < 1 m 11 m denth	GCHEM	GCHEM	12/05/2006	17.20	7400	238302	1952	270 4	29.48	17.45	6.88	1.66	-100	-100	04.01	00.14	04.04	27.00	20.00	20.00	
	and a concert in		i annenig on	"	GCHEM	GCHEM	19/05/2006			104217	778 2	106	21.04	7 91	5.12	1.50									
				Soil das <1 m	Maxyam/LlofA	Lionhead	31/05/2009			194217	110.2	100	21.04	1.01	0.12	1.55									Wet soil conditions
				Son guo, si m	in and an in OolA	Liointeau	0110012000		1.1				1								1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				

Sample Depth	Date sampled	CO <sub>2</sub> (ppm)	Total Gas (unit)	$\delta^{13}C C_1$	δ <sup>13</sup> C C <sub>2</sub>	δ <sup>13</sup> C C <sub>3</sub>	δ <sup>13</sup> C i-C <sub>4</sub>	δ <sup>13</sup> C n-C <sub>4</sub>	$\delta^{13}$ C CO <sub>2</sub>
340	28-Oct-09	131	1324	-56.78	-36.29	-32.02	-30.49	-33.44	-21.95
407	28-Oct-09	116	752			*		i <del>i</del> ,	
657	29-Oct-09	179	735	-55.18	-40.72	-38.40	-35.16		-18.40
734	29-Oct-09	98	444	-51.45	-35.00	-36.32	-28.88	-36.64	-12.20
844	29-Oct-09	115	705	-44.65	-28.11	-29.04	-30.49		-15.07
844 lab duplicate	29-Oct-09	115	705	-44.77	-28.22	-28.90	-29.23	-31.49	-15.57
871	29-Oct-09	111	607	-45.76	-29.37	-30.89	-29.31	-33.60	-29.13
920	29-Oct-09	50450 50450	442	-42.64	-29.01	-28.57	-27.99	-30.05	-11.74
925	29-Oct-09		846	-42.60	-28.91	-28.22	-28.05	-29.92	

# Table 4: Summary of 100/02-04-078-07 W6M mud log data.

WELLID	Pool or Zone	Statue	Unceme	ented Zones	SCVE	Possible source of	Gas Migration	Distance from
	F001012011e	Status	Top (mKb)	Bottom (mKb)	30 1	SCVF	Outside Casing	Jack Well (m)
100/04-07-078-07 W6M	Charlie Lake Fm	Pumping oil			No		Not Tested	1712
100/14-01-078-08 W6M	Charlie Lake Fm	Pumping oil	290.0	543.0	Yes	Cadotte	Yes	720
100/16-01-078-08 W6M	Charlie Lake Fm	Pumping oil	283.9	400.0	No		Not Tested	1390
102/16-01-078-08 W6M	Charlie Lake Fm	Flowing oil			No		Not Tested	1408
100/16-02-078-08 W6M	Charlie Lake Fm	Abandoned	304.0	429.0	No		Not Tested	870
100/08-11-078-08 W6M	Charlie Lake Fm	Pumping oil	222.0	240.0	No		No	501
100/15-11-078-08 W6M	Charlie Lake Fm	Pumping oil			No		Not Tested	1246
100/16-11-078-08 W6M	Charlie Lake Fm	Water Injection	223.0	335.0	Yes	Cadotte	Not Tested	972
100/02-12-078-08 W6M	Charlie Lake Fm	Pumping oil			No		No	825
100/06-12-078-08 W6M	Charlie Lake Fm	Water Injection	298.0	<sup>1</sup> 1000.0	Yes	Cadotte	Yes	429
100/08-12-078-08 W6M	Charlie Lake Fm	Pumping oil			No		Not Tested	1132
100/11-12-078-08 W6M	Charlie Lake Fm	Flowing Gas	252.0	745.0	Yes	Cadotte	Yes	795
102/11-12-078-08 W6M	Charlie Lake Fm	Pumping oil			Yes	Cadotte	No	774
100/16-12-078-08 W6M	Charlie Lake Fm	Water Injection	224.6	840.0	Yes	Cadotte	Yes	1438
100/02-13-078-08 W6M	Charlie Lake Fm	Pumping oil			Yes	Cadotte	Not Tested	1593
100/03-13-078-08 W6M	Charlie Lake Fm	Pumping oil			No		Not Tested	1417
100/02-14-078-08 W6M	Charlie Lake Fm	Pumping oil			No		Yes	1370

# Table 5: Summary of uncemented intervals, surface casing vent flow and gas migration for energy wells within 1.5 km of the Jack water well.

<sup>1</sup>Note: In January 2010, 100/06-12-078-08 W6M underwent remedial cementing work which brought the top of cement to 490 mkB.

#### 3.7 Pressure Build-up Test

A pressure build-up test was performed to evaluate whether a direct connection existed between energy well 100/06-12-078-08 W6M and Mr. Jack's water well. The surface casing vent was closed on the energy well and pressure was allowed to build up. Water levels and gas flow from the casing of the Jack water well were monitored during the test. Figure 14 presents the surface casing vent pressure data along with the water level in the Jack water well prior and during the same time period.

Water levels in the Jack water well remained fairly constant during the buildup test in 100/06-12-078-08 W6M. It is important to note that fluctuations of up to about 1 cm may arise from instrumental drift and any minor effect of gas bubbling in the water column during non-pumping conditions. Note also that variations in water level of up to 5 cm are attributed to shifts in barometric pressure of the atmosphere. The Jack water well did not explosively erupt from the casing during the duration of the test and gas flow rates were too low to measure with the configuration of the gas flow meter. There does not appear to be a direct relationship between pressure build-up in the surface casing vent of 100/06-12-078-08 W6M and water levels in the Jack water well over the 24 hour test and pressures reached.

#### 3.8 Jack Water Well Testing Pre- and Post- Remedial Cementing of 100/06-12-078-

#### 3.8.1 Pre-Remedial Cementing of 100/06-12-078-08 W6M (September 26, 2009)

The pumping test data for the September 26, 2009 test is presented in Figure 15. For the preremedial cementing test, the Jack water well was pumped at a rate of 8.75 to 9 Igpm. The pumping rate declined slightly (by about 0.25 Igpm) over the test as a result of declining water levels in the water well and a greater head of water the pump was required to lift. After about 3 hours and 35 minutes of pumping, when the water level in the water well had been drawn down by approximately 2.66 m (30.54 m below Top of Casing; TOC) the water well started to erupt with gas (observed at the water well). The pressure head measured with the transducer became variable (by about 1 m) after this time, due to eruption of the gas from the casing and changes in the density of water above the transducer. Gas flow rates (red symbols on Figure 15) were not available until after the sensor on the gas flow meter was repaired on September 27, 2009. Instantaneous gas flow rates (recorded every 10 seconds) from the casing show a fairly regular pattern of gas eruption (at a rate of up to 7.5 cfm) for 80 to 90 seconds, followed by 30 or 40 seconds of rest. Near the end of the pumping portion of the test, the water had drawn down about 6.0 m (similar to the February 18, 2008 pumping test performed by ARC) and the well eruptions lasted for up to almost 4 minutes. The amount of gas exsolving from the pumped water was estimated to be 800 ml of gas per minute of pumping.

When the pump stopped, the water well immediately stopped degassing, and gas flow rates dropped below the detection limit of the instrument (about 0.035 cfm or 1 L/min). Recovery of the water well was monitored until 6:00 a.m. on September 28, 2009. The transducer was left in the Jack water well to record water levels every 15 minutes.

#### 3.8.2 First Post- Remedial Cementing of 100/06-12-078-08 W6M (January 29, 2010)

The pumping test data for the January 29, 2010 test is presented in Figure 16. For the first post-remedial cementing test the Jack water well was pumped at a rate of 7.3 to 5.8 Igpm. Even though the same pump was used, the initial pumping rate was lower than in the pre-remedial cementing test because (at the request of Mr. Jack) about 100 m of discharge pipe was used to divert the water away from farming operations. This extra discharge pipe added friction and reduced the pumping rate. The pumping rate dropped by about 1.5 Igpm over the duration of the test. This was due to loss of pump efficiency when gas started exsolving from the water. After about 6 hours and 30 minutes of pumping, the pressure head measurements started to become variable due to gas exsolving from the water due to pressure head drop. The variability is about 20 cm, much less than the previous September 26, 2009 test. The water level in the water well was drawn down approximately 2.3 m at this time to a depth of about 29.63 m below TOC. There was no casing gas flow detected by the gas flow meter. The amount of gas exsolving from the pumped water was estimated to be 100 ml of gas per minute of pumping.

When the pump stopped, the gas flow rate remained below the detection limit of the instrument. Recovery of the water well was monitored until 1:05 p.m. on January 31, 2010. A transducer was left in the Jack water well to record water levels every 15 minutes.

#### 3.8.3 Second Post- Remedial Cementing of 100/06-12-078-08 W6M (February 27, 2010)

The pumping test data for the February 27, 2010 test is presented in Figure 17. For the second post-remedial cementing test the Jack water well was pumped at a rate of 6.6 to 5.1 Igpm. Even though the same pump and discharge line was used, the initial pumping rate was lower than in the first post-remedial cementing test. The installed pump seems to be getting weaker with each additional test. The pumping rate dropped by about 1.5 Igpm over the duration of the test. This was due to loss of pump efficiency when gas started exsolving from the water. After about 6 hours and 30 minutes of pumping, the pressure head measurements started to become variable due to gas exsolving from the water due to pressure head drop and from pump cavitation. The variability is about 95 cm, which is similar to September 26, 2009 test. The water level in the water well was drawn down approximately 2.5 m at this time, to a depth of about 29.62 m below TOC. There was no casing gas flow detected by the gas flow meter or using a plastic bag over the gas vent on the well. Bubbling could be heard but with little to no gas produced. The amount of gas exsolving from the pumped water was estimated to be 100 ml of gas per minute of pumping in the first 7 hours of pumping. After 24 hours the exsolved gas was estimated to be about 50 ml per minute using the inverted bottle technique. The inverted bottle technique is a simple method for sampling free gas in groundwater that can also be used to estimate gas volumes. Using this technique groundwater is directed into an inverted bottle submersed in a water-filled pail. The gas displaces water in the inverted bottle, accumulating in the headspace.

When the pump stopped, the gas flow rate remained below the detection limit of the instrument. Recovery of the water well was monitored until 12:35 p.m. on March 1, 2010. A transducer was left in the Jack water well to record water levels every 15 minutes.

#### 3.8.4 Third Post-Remedial Cementing of 100/06-12-078-08 W6M (May 22, 2010)

The pumping test data for the May 22, 2010 test is presented in Figure 18. For the third postremedial cementing test the Jack water well was pumped at a rate of 7.75 to 6.7 Igpm. Even though the same pump and discharge line was used, the initial pumping rate was higher than in the first post-remedial cementing test. The pumping rate dropped by about 1.05 Igpm over the duration of the test. This was due to an increase in head and maybe some loss of pump efficiency when gas started exsolving from the water. After about 8 hours and 30 minutes of pumping, the pressure head measurements started to become variable due to gas exsolving from the water due to pressure head drop and from pump cavitation. The variability is about 3 cm, which is about 3% of the variation of the September 26, 2009 test. The water level in the water well was drawn down approximately 3.6 m at this time, to a depth of about 30.18 m below TOC. There was no casing gas flow detected by the gas flow meter or by using a plastic bag over the gas vent on the well. Bubbling could be heard but with little to no gas produced. The amount of gas exsolving from the pumped water in the first few minutes of pumping was estimated to be 140 ml of gas per minute using the inverted bottle technique, (300 ml per minute with the gas separator). After 6 hours of pumping the inverted bottle technique measured 460 ml of gas (800 ml of gas were measure in the gas separator over the same time period). After 24 hours the exsolved gas was estimated to be about 120 ml per minute using the inverted bottle technique while the gas separator still measured about 800 ml per minute of free gas production.

When the pump stopped, the gas flow rate remained below the detection limit of the flow meter. Recovery of the water well was monitored until 9:00 pm on May 23, 2010.

The aquifer test data was analyzed using AQTESOLV, Version 3.50 Professional, Aquifer Test Design and Analysis Computer Software (1996-2003 HydroSOLVE Inc.). This software provides analytical solutions for evaluating parameters in confined, unconfined, leaky, or fractured aquifer systems, and allows evaluation of the aquifer test data by visual curve matching to select the most appropriate interpretation to represent aquifer conditions at the site. The raw data and graphical solutions are included in Appendix D.

The Theis (1935) confined aquifer solution was used to solve both the pumping test and the recovery test for the pumping tests performed on September 26, 2009, January 29, 2010 and February 27, 2010 (Table 7). A pumping recovery test was performed on the Jack well when it was drilled in November 19, 2001. The 2001 recovery data was used to solve the recovery portion of the pumping test (information on pumping rate was not available) giving a transmissivity in the range of  $1.05 \times 10^{-3} \text{ m}^2/\text{min}$  to  $9.79 \times 10^{-3} \text{ m}^2/\text{min}$  (ARC, 2008). A pumping test performed on February 18, 2008 gave apparent transmissivity estimates ranging between  $1.65 \times 10^{-3} \text{ m}^2/\text{min}$  to  $3.28 \times 10^{-3} \text{ m}^2/\text{min}$  (2.4 to  $4.7 \text{ m}^2/\text{day}$ ) (ARC, 2008).

		Pi	umping tests Conducted	Pre-Remedial Cementi	ng of 6-12	Pumping tests	Conducted Post-Reme	dial Cementing of 6-12
		November 19, 2001	February 20, 2008	March 18, 2009	September 24, 2009	January 29, 2010	February 27, 2010	May 22, 2010
	Depth to static water level	16.27 m (from TOC)	26.75 m (from TOC)	29.38 m (from TOC)	27.88 m (from TOC)	27.33 m (from TOC)	27.12 m (from TOC)	26.58 m (from TOC)
Non- Pumping Behaviour	Gas flow rate	NA	Constant flow of ~60 L/min Mr. Jack measured 100 cf/min during eruption	Constant flow of ~5 L/min	Constant flow of ~5 L/min	None detected	None detected	None detected
	Gas eruption frequency	NA	Every 5-10 min	None since Jan. 19, 2009	None	None	None	None
	Gas location	NA	Casing (top slot in liner), small amount exsolved from water	NA	Casing (top slot in liner)	No gas	No gas	No gas
	Relation to pumping	NA	Immediate eruptive degassing, that stopped when pumping stopped	NA	Pumped for 3 hours before eruptive degassing started, stopped when pumping stopped	No eruptive degassing even after pumping, exsolution of gas after 6.5 h of pumping	No eruptive degassing after pumping, exsolution of gas after 6.5 h of pumping	No eruptive degassing after pumping, exsolution of gas after 11 h of pumping
	Relation to water level	NA		NA	2.66 m drawdown before eruptive degassing	No eruptive degassing, exsolution after 2.3 m drawdown	No eruptive degassing, exsolution after 2.5 m drawdown	No eruptive degassing, exsolution after 3.6 m drawdown
Pumping Behaviour	Water depth when eruptive degassing began			14	30.54 m TOC	No degassing, water level drawn down to 29.63 m TOC	No degassing, water level drawn down to 29.62 m TOC	No degassing, water level drawn down to 30.18 TOC
	Gas flow rate	NA	Mr. Jack measured 100cf/min during eruption	NA	~5 L/min	Below detection of flow meter	Below detection of flow meter	Below detection of flow meter
	Gas eruption frequency	NA	Constant	NA	Gas pulse every 50 sec	None	None	None
	Gas location	NA	Majority from casing (top slot in liner), small amount exsolved from water	NA	Majority from casing (~800 mL/min exsolved from water)	None from casing, <100 mL/min exsolved from water	None from casing, <100 mL/min exsolved from water	None from casing, <100 mL/min exsolved from water

# Table 6: Changes in Jack water well behaviour over time

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	Date	Transmissivity (m²/min)
December distances time of	November 19, 2001	1.05 x 10 <sup>-3</sup> to 9.79 x 10 <sup>-3</sup>
Pre-remedial cementing of	February 18, 2009	1.65 x 10 <sup>-3</sup> to 3.28 x 10 <sup>-3</sup>
100/06-12-078-08 00610	September 26, 2009	1.46 x 10 <sup>-3</sup> to 2.73 x 10 <sup>-3</sup>
Post-remedial cementing of	January 29, 2010	2.91 x 10 <sup>-3</sup> to 3.32 x 10 <sup>-3</sup>
100/06-12-078-08 W6M	February 27, 2010	4.21 x 10 <sup>-3</sup> to 4.70 x 10 <sup>-3</sup>

#### Table 7: Summary of pumping test results

The post-remedial cementing transmissivity estimates are within the range estimated from the original recovery test performed in 2001. As was noted in previous report (ARC, 2008) the transmissivity estimates are generally higher than is normally found in shale and sandstone, possibly due to the presence of fractures in the aquifer. There is a slight increasing trend in the estimates of transmissivity, but this could be due to a decrease in gas surging during the pumping tests, allowing for a more representative estimate to be made.

#### 3.8.5 Changes in Behavior of the Jack Well Pre- and Post-Remedial Cementing

The idea behind conducting pumping tests before and after remedial cementing of the energy well at 100/06-12-078-08 W6M was to see if there were any changes in behaviour of the water well after remedial cementing of the energy well. The tests were conducted over a 9 month period, a period when the water level in the Jack well was rising (1.3 m between the September 24, 2009 pumping test and the May 22, 2010 pumping tests). The rise in water levels combined with decreased pump efficiency due to increased discharge pipe length meant that the pumping tests performed after remedial cementing were not drawn down to the same water level as the pre-remedial cementing pumping test (pre-remedial cementing pumping test water levels were drawn down to 30.58 m below TOC, post-remedial cementing pumping test water levels did not go below 30.18 m below TOC). The lack of eruptive degassing during pumping for the post-remedial cementing pumping tests may be the result of not having attained the same degree of drawdown in those tests.

There were changes in the non-pumping gas flow rate before and after remedial cementing of 100/06-12-078-08 W6M. There are three measurements of gas flow rates available for the Jack water well prior to the remedial cementing. On February 20, 2008 Mr. Jack reported a constant gas flow of about 60 L/min, and on March 18, 2009 and September 24, 2009 AITF personal measured a constant gas flow of about 5 L/min prior to pumping. After the remedial cementing of the energy well gas flow rates were measured before each of the three pumping tests, but gas flow could not be detected by the gas flow meter. The 5 L/min of gas discharge were measured when the static water level in the Jack well was lower (27.88 m below TOC) than after the remedial cementing (27.33 m below TOC).

Without more frequent records of water levels, gas discharge and usage for the Jack water well, it is hard to evaluate the causes of the changes in the behaviour of the well, and to separate changes that may have been due to energy well activity versus those due to water level changes. Our best estimates of when gas eruptions first appeared (2003) and

disappeared (2009) are based on anecdotal sources, and there are only a couple of water level measurements made during this period and no information on water usage.

Based on the limited gas flow and manual water level measurements available it appears that there was still constant gas discharge and spontaneous degassing during periods when the water levels were higher than present, indicating that the changes in the form and amount of gas present in the well cannot be entirely due to changes in water level. For example during the pumping test conducted in Feb. 2008 the pre-pumping water level was 26.75 m below TOC yet there was still a large volume of gas discharging before pumping and the well was spontaneously erupting. During the pumping test conducted on September 24, 2009 there was a constant flow of gas measurable at about 5 L/min when the static water level was 27.88m below TOC. The post remedial cementing pumping test conducted on January 29, 2010 even when the water levels were drawn down to 29.63 m below TOC there was no measurable gas flow (Table 6). Changes in the amount of gases present in the groundwater combined with changes in the solubility of those gases due to pressure changes could result in the variable relationship between water levels and gas discharge volumes observed for the Jack water well.

#### 4 CONCLUSIONS

Alberta Innovates – Technology Futures' review of new investigations regarding the Jack water well complaint provides the following conclusions:

- The composition and δ<sup>13</sup>C values of the hydrocarbon gases sampled in the Jack well are consistent with a mixture of biogenic and thermogenic sources.
  - The δ<sup>13</sup>C of methane is within the range expected for predominantly biogenic methane. The concentrations and isotopic compositions of ethane and propane in gases sampled from the Jack water well are consistent with a component of thermogenic gas.
  - The isotopic signature of ethane in the Jack well is similar to that found in the Cadotte Formation and in SCV from 100/06-12-078-08 W6M and 100/11-12-078-08 W6M.
  - The  $\delta^{13}$ C value measured for propane in the Jack well is similar to the propane  $\delta^{13}$ C measured in the production casing from 100/11-18-078-07 W6M, a water source well completed across the Cadotte formation.
  - The more negative  $\delta^{13}$ C composition of methane sampled in the Jack water well after remedial cementing of energy well 100/06-12-078-08 W6M could indicate a decrease in the thermogenic component of methane to this well. This single postremedial cementing sample is within the range of  $\delta^{13}$ C variability measured in water samples prior to the remedial cementing and it was analyzed at a different laboratory, so repeat sampling and submission of a duplicate and standard are included in the recommendations.

- New carbon isotope sampling of the surface casing vent of 100/06-12-078-08 W6M found an isotopic enrichment trend over a four year period that may be a natural fractionation process in the energy well.
- Gas migration studies confirm that gas migration to surface was occurring near the energy well bores at 100/11-12-078-08 W6M and 100/06-12-078-08 W6M. The gas and isotopic compositions of soil gas sampled in the vicinity of 100/11-12-078-08 W6M has a thermogenic signature. Soil gas sampled in the vicinity of 100/06-12-078-08 W6M appears to be a mixture of thermogenic and biogenic sources of hydrocarbon gases. After remedial cementing was conducted at 100/06-12-078-08 W6M repeated gas migration tests showed decreasing concentration of methane and ethane in soil gas, eventually reaching 0% LEL by the November 11, 2010 test. The δ<sup>13</sup>C of methane in soil gas samples from the vicinity of this well became more negative, consistent with less of a contribution of thermogenic methane. It appears that leakage in the form of gas migration has ceased.
- A gas isotope mudlog from a new energy well drilled at 100/02-04-078-07 W6M (located 5.6 km SE of the Jack water well) provides an improved dataset to characterize the isotopic signature of gases from different formations in the area. The δ<sup>13</sup>C composition of ethane gas sampled in the Jack well and surface casing vent from energy wells in the area is within 1 ‰ of the mud sampled from the Cadotte Formation suggesting that this formation could be the source of gas. The propane δ<sup>13</sup>C of the Cadotte Formation did not match the limited propane dataset available for the Jack well.
- The results of the lineament study did not find any evidence of a direct connection between the Jack water well and any of the adjacent energy wells. However, the study did indicate that the study area is part of the Peace River Arch, an area that contains numerous faults and that fractures or fault systems may be pathways for the movement of fluids and gases.
- However, better knowledge of background gas concentrations in shallow groundwater and isotopic compositions and the regional gas characteristics from the Cadotte Formation and other geological formations would improve our ability to identify the source of the gases reaching the Jack well. Even if the gases present in the Jack water well can be linked to specific geological units, the presence of fractures and fault networks in the area will make it difficult to determine the pathways for gases to shallow aquifers. Better characterization of background shallow groundwater gas concentrations and isotopic compositions would help determine if there is a component of thermogenic gases naturally present in shallow groundwater in this region due to the fractured nature of the underlying geological units.
- Water levels compiled from manual measurements made in 2001, 2006, 2008, 2009 and 2010 show a general trend of decreasing levels between 2001 and late 2008 (decrease of 12.6 m over time period), followed by an increasing trend that seems to have started in late 2008 or early 2009 (increase of 2.3 m over that time period). The causes for the observed changes in water level are not clear.

- A lot of the activities summarized in this report were directed towards identifying whether the energy well located at 100/06-12-078-08 W6M could be the source of gases to the Jack well. The δ<sup>13</sup>C signature of ethane sampled in the Jack well is similar to the δ<sup>13</sup>C of ethane sampled from SCV at 100/06-12-078-08 W6M and soil gases sampled as part of the gas migration study, but other than the similarity in isotopic signature, there was no evidence of a direct connection between the two wells.
  - Water injection at 100/06-12-078-08 W6M started 10 months after gas first appeared in the Jack well, indicating that injection was not likely the cause of gas in the water well.
  - The pressure build-up test of the surface casing vent of 100/06-12-078-08 W6M indicates there is no direct connection between the surface casing vent in 100/06-12-078-08 W6M and the Jack water well over the pressures and time period tested.
  - The new cement bond and neutron density log of energy well 100/06-12-078-08 W6M indicate that as of January 2010 the cement top is at 490 mKb, so the uncemented portion of this well is from 298 to 490 mKb. The remedial cementing work conducted in January 2010 appears to have stopped the SCVF as indicated by bubble tests performed in May 2010 and September 2010 and gas migration tests performed in November 2010.
  - The isotopic results for 100/06-12-078-08 W6M seem to indicate a common source for the ethane in the SCV and soil gas from this energy well and the Jack well, but not necessarily a direct connection.
- There have been changes in the occurrence and frequency of eruptive degassing, the volume of gas discharging from the well and the form of the gas present in the well (free gas vs. dissolved gas).
  - Spontaneous degassing apparently stopped shortly before March 2009, more than 9 months before the remedial cementing of 100/06-12-078-08 W6M.
  - After remedial cementing of 100/06-12-078-08 W6M the amount of free gas discharging from the water well decreased, but the amount of dissolved methane in the well water increased. The increase in dissolved gases could be in part due to increasing water levels, with higher pressure heads resulting in higher concentrations of dissolved methane.
  - Prior to remedial cementing of the energy well at 100/06-12-078-08 W6M there was a measurable discharge of gas under non-pumping conditions, whereas on the three visits to the well post-remedial cementing no flow was detected. Over this time period static water levels in the Jack well increased from 27.88 m below the top of casing (TOC) on September, 24, 2009 to 26.58 m below TOC on May 22, 2010. This change in the amount of gas discharging under non-pumping conditions may have been influenced by the increases in water levels.
- These changes in the amount and form (free or dissolved) of gas in the Jack well may be due to differences in the solubility of methane in groundwater due to increases in pressure head as water levels rise. However, these changes would not significantly

affect the isotopic signature of the gas. Because the series of pumping test occurred over a period when there appears to have been a general increase in water levels it is not possible to separate changes related to remedial cementing of the energy well from changes related to the increase in water level with the data currently available.

 Pumping tests conducted on the Jack well before and after the remedial cementing of energy well 100/06-12-078-08 W6M have found no significant changes in aquifer transmissivity estimates. Transmissivity estimates have remained within an order of magnitude of the original estimates.

#### 5 RECOMMENDATIONS

Alberta Innovates – Technology Futures' review of new investigations regarding the Jack water well complaint provides the following recommendations:

- If the Jack water well is to be used, the water well should be equipped in a manner such that there is no explosion potential.
- The Jack water well had a heavy bacterial presence (slime, sulfur reducing, heterotrophic and iron reducing bacteria) measured on February 20, 2008 (ARC 2008). This water well should be maintained with regular shock chlorination and may need cleaning by a water well professional.
- The Jack water well should not be over-pumped. Its yield is fairly unique in the immediate area and the water well likely could not easily be replaced. The water well has been used in the past to fill a dugout on the property. The volume of the small dugout on the Jack property is about 1250 m<sup>3</sup>. The Jack's currently have permission to divert a total of 2386 m<sup>3</sup>/yr. (1136 m<sup>3</sup>/yr., license for stock watering purposes; 1250 m<sup>3</sup>/yr., household purposes). A safe pumping rate for this water well was calculated to be about 3 imperial gallons per minute (ARC, 2008). Filling the dugout over a short period of time or multiple times a year would likely over-pump the water well and aquifer.
- The pump-tests conducted post-remedial cementing did not reach the same level of drawn down as the pre-remedial cementing test and the results relating to the amount of gas present in the well are therefore not directly comparable. We recommend that a pumping test be conducted with a stronger pump and a long enough duration to reach at least the same levels of drawdown (30.8 m below TOC) to see if gases are still discharging. Gas and water samples taken during the pumping test, combined with detailed monitoring of gas flow volumes and water levels will give better insight into the sources of gas as the well is drawn down.

• Resampling of the Jack water well to obtain gas and water samples for geochemical and isotopic analyses is recommended. The gas samples will be submitted to the AITF laboratory in Victoria, and the isotope laboratories at the University of Calgary and the University of Alberta. A methane standard gas sample will also be submitted to all laboratories. This repeat sampling will confirm whether the isotopic shift in the methane  $\delta^{13}$ C composition measured is real and will allow us to see if the methane  $\delta^{13}$ C values measured by the two laboratories are directly comparable.

Further recommendations will be developed once the new pump test and sampling results have been evaluated.

#### 6 CLOSURE

This work was carried out in accordance with accepted hydrogeological practices.

Respectfully submitted,

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Jean Birks, Ph.D.



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February 15, 2011 Alberta Innovates – Technology Futures

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- Rowe, D. and Muehlenbachs, K. 1999. Isotopic fingerprints of shallow gases in the Western Canadian sedimentary basin: tools for remediation of leaking heavy oil wells. Organic Geochemistry, 30, p. 861-871.

### Figure 1: Site map and energy well locations



Aerial Map of Jack Well and Energy Wells in Area

Catronical 2216 Feb.2



## Figure 2: Stratigraphic column for the Northwestern Plains and Deep Basin





ALBERTA INNOVATES - TECHNOLOGY FUTURES » DECEMBER 2010






#### Methane and ethane concentrations measured in water and energy wells. Figure 4:

Figure 5: Summary of isotope data from ARC, 2008 (solid symbols) and new data (open symbols). Average values for the composition of shallow groundwater in Central Alberta, the Horseshoe Canyon/Belly River Group and Manville Formations from Cheung et al. (2010) are included.



Energy Well- Prod.Casing Energy Well- SCV

Jack Water Well

- Water Injection Well- Prod. Casing Water Injection Well - SCV
- Jack Water Well, Post-remediaiton
- Water Source Well- Prod. Casing
- Water Injection Well- SCV

- Water Injection Well- Prod. Casing

32

Figure 6: Summary of isotope data from ARC, 2008 (solid symbols) and new data (open symbols. Average values for the composition of shallow groundwater in Central Alberta, the Horseshoe Canyon/Belly River Group and Manville Formations from Cheung et al. (2010) are included.





Figure 7: Methane and ethane carbon isotope compositions of the Jack water well, energy well gases and the mudlog (top panel) and samples from the soil gas migration study (bottom panel). Average values for the composition of shallow groundwater in Central Alberta, the Horseshoe Canyon/Belly River Group and Manville Formations from Cheung et al. (2010) are included.



34

Figure 8: Carbon isotope values of hydrocarbon gases with time for Jack Water well and 100/06-12-078-08 W6M surface casing vent.







36

### Figure 10: Soil migration study results for 100/11-12-078-08.

100/11-12-078-08 (Sept. 16, 2009)



### Figure 11: Soil migration study results for 100/06-12-078-08.

100/06-12-078-08 Pre-Remediation (Sept. 16, 2009)



100/06-12-078-08 Post-Remediation (May 13, 2010)



100/06-12-078-08 Post-Remediation (Sept.22, 2010)







Spirit River 100/02-04-078-07 W6M



# Figure 13: Gantt chart showing timing of events surrounding gas occurrence in the Jack water well.



Figure 14: 100/06-12-078-08 W6M Surface casing vent build-up test and water levels in the Jack water well.



Figure 15: September 26, 2009 pumping test of the Jack water well.





Figure 17: February 27, 2010 pumping test of the Jack water well.



Note: water levels reported as depth from top of vent. 0.61 m above top of casing (TOC).



Figure 18: May 22, 2010 pumping test of the Jack water well.

## Appendix A

### **Energy Well Analytical Reports**

ALBERTA RESEARCH COUNCIL

ORGANICS ANALYSIS DATA SHEET

ARC SAMPLE NUMBER: T09-0876

						CI	.ient: Jor	nes, Don			
Sample No:			Group	Sample No:	Site	Descrip/Com	ument: 11-	-18 Exsolved			
Station No:				Project No:	Canister:						
Agency:	ARC	Samp	Type:	SampMatrix:	Collection:	Samp	Date: 19-	-Mar-09 Time:	1018	Samplers	ID:
SubGroups	FILE	VM	IV	NAME		ConcRpt	MDL	ConcRptUnit	InjDa	te	
							******				
DG_C1C4											
	w0876	10	6770	Butane		24.30	.01	ug/L	25-Ma	r-09	
	w0876	10	6771	Ethane		905.00	.01	ug/L	25-Ma	r-09	
	w0876	10	6772	Ethylene		0.00	.01	ug/L	25-Ma	r-09	
	w0876	10	6773	Isobutane		14.50	.01	ug/L	25-Ma	r-09	
	w0876	10	6774	Methane		24300.00	.01	ug/L	25-Ma	r-09	
	w0876	10	6775	Propane		168.00	.01	ug/L	25-Ma	r-09	
DC TCD											
00_100	1.0876	10	6776	Carbon dioxide		11 40	1 00	ma/I.	26-Ma	r-09	
	1.0876	10	6777	Nitrogen		9 21	6.00	mg/L	26-Ma	r-09	
	L0876	10	7106	Oxygen		6.12	6.00	mg/L	26-Ma	r-09	
G_C1C4											
	c0876	10	6778	Butane		31.80	.05	ppmv	25-Ma	r-09	
	c0876	10	6779	Ethane		3740.00	.05	ppmv	25-Ma	r-09	
	c0876	10	6780	Ethylene		0.00	.05	ppmv	25-Ma	r-09	
	c0876	10	6781	Isobutane		30.40	.05	ppmv	25-Ma	r-09	
	c0876	10	6782	Methane		897000.00	.05	ppmv	25-Ma	r-09	
	c0876	10	6783	Propane		133.00	.05	ppmv	25-Ma	r-09	
G_TCD		19000	50 S N			andre and			2.29	200	
	G0876	10	6784	Carbon dioxide		555.00	300.00	ppmv	26-Ma	r-09	
	G0876	10	6785	Nitrogen		30000.00	1000.00	ppmv	26-Ma	r-09	
	G0876	10	7107	Oxygen		9380.00	1000.00	ppmv	26-Ma	r-09	

[ARC\_Remarks]: Tedlar Bag

SubGroups				
DG_C1C4 and DG_TCD - Diso	lved Gas in water sample			
G_C1C4 and G_TCD - Free	Gas from canister			
5 				
Certified For: Yogesh Kumar,	Business Unit Manager	Mail To:	Jones, Don	
	Environmental Monitoring		Sustainable Ecosystems	
By:	Alberta Research Council		Alberta Research Council	
	Vegreville, Alberta			
	T9C 1T4		Calgary, Alberta	
Date: 30-Mar-09 Contact:	Grant Prill 780 632-8455		T2L 2A6	

ALBERTA RESEA	RCH COUNCIL		ORGANICS ANAL	YSIS DATA S	HEET		ARC S	SAMPLE N	UMBER:	T09-0877
									7	
				C	lient:	Jones, Don				
Sample No:	Grou	p Sample No:	Site	Descrip/Co	mment:	11-18 Production	Casing	3		
Station No:		Project No:	Canister:							
Agency: A	RC Samp Type:	SampMatrix:	Collection:	Samp	Date:	19-Mar-09 Time:	1012 5	Samplers	ID:	
SubGroups F	ILE VMV	NAME		ConcRpt	MDL	ConcRptUnit	InjDat	te		
G_C1C4										

	c0877	106778	Butane	39.70	.05	ppmv	25-Mar-09
	c0877	106779	Ethane	3730.00	.05	ppmv	25-Mar-09
	c0877	106780	Ethylene	0.00	.05	ppmv	25-Mar-09
	c0877	106781	Isobutane	35.10	.05	ppmv	25-Mar-09
	c0877	106782	Methane	745000.00	.05	ppmv	25-Mar-09
	c0877	106783	Propane	154.00	.05	ppmv	25-Mar-09
G_TCD							
	G0877	106784	Carbon dioxide	4740.00	300.00	ppmv	26-Mar-09
	G0877	106785	Nitrogen	168000.00	1000.00	ppmv	26-Mar-09
	G0877	107107	Oxygen	9470.00	1000.00	ppmv	26-Mar-09

[ARC\_Remarks]: Tedlar Bag

SubGroups			
DG_C1C4 and DG_TCD - Dis	olved Gas in water sample		
	.*.		
G_C1C4 and G_TCD - Fre	e Gas from canister		
Certified For: Yogesh Kumar	, Business Unit Manager	Mail To:	Jones, Don
nan sana na mana na manana na manana na manana 🧰 na mana 🦉 na manana na sana na	Environmental Monitoring		Sustainable Ecosystems
By:	Alberta Research Council		Alberta Research Council
	Vegreville, Alberta		
	T9C 1T4		Calgary, Alberta
Date: 30-Mar-09 Contact	: Grant Prill 780 632-8455		T2L 2A6
	×		

ALBERTA RESEARCH COUNCIL

ORGANICS ANALYSIS DATA SHEET

ARC SAMPLE NUMBER: T09-0878

					Cl	ient: Jon	les, Don		
Sample No:		Grou	Sample No:	Site	Descrip/Com	ment: 11-	18 Production	Tubing	
Station No:			Project No:	Canister:					
Agency:	ARC	Samp Type:	SampMatrix:	Collection:	Samp	Date:	Time:	Samplers	ID:
Cub Cuova a	DITE	12.07	N77 M75		Concent	MDT	ConcEntinit	TaiData	
SubGroups	FILE	VMV	NAME		CONCRPT	MDL	Conceptunit	Injuate	
G_C1C4									
	c0878	106778	Butane		93.90	.05	ppmv	25-Mar-09	
	c0878	106779	Ethane		4480.00	.05	ppmv	25-Mar-09	
	c0878	106780	Ethylene		0.00	.05	ppmv	25-Mar-09	
	c0878	106781	Isobutane		55.70	.05	ppmv	25-Mar-09	
	c0878	106782	Methane		945000.00	.05	ppmv	25-Mar-09	
	c0878	106783	Propane		176.00	.05	ppmv.	25-Mar-09	
G_TCD									
	G0878	106784	Carbon dioxide		0.00	300.00	ppmv	26-Mar-09	
	G0878	106785	Nitrogen		1260.00	1000.00	ppmv	26-Mar-09	
	G0878	107107	Oxygen		5020.00	1000.00	ppmv	26-Mar-09	
	G08782	106784	Carbon dioxide		0.00	300.00	ppmv	26-Mar-09	
	G08782	106785	Nitrogen		1340.00	1000.00	ppmv	26-Mar-09	
	G08782	107107	Oxygen		5130.00	1000.00	ppmv	26-Mar-09	

[ARC\_Remarks]: Tedlar Bag

DG_C1C4 and DG_TCD - Disolved Gas in water sample G_C1C4 and G_TCD - Free Gas from canister
G_C1C4 and G_TCD - Free Gas from canister
Certified For: Yogesh Kumar, Business Unit Manager Mail To: Jones, Don
Environmental Monitoring Sustainable Ecosystems
By: Alberta Research Council Alberta Research Council Vegreville, Alberta
T9C 1T4 Calgary, Alberta
Date: 30-Mar-09 Contact: Grant Prill 780 632-8455 T2L 2A6

ALBERTA RESEARCH COUNCIL

#### ORGANICS ANALYSIS DATA SHEET

ARC SAMPLE NUMBER: T09-0879

						Same -	Tenne Den			
					cifent. bones, bon					
Sample No:		Grou	p Sample No:	Site	Descrip/Con	nment: (	6-12 Surface Cas	sing		
Station No:			Project No:	Canister:						
Agency:	ARC	Samp Type:	SampMatrix:	Collection:	Samp	Date: 2	19-Mar-09 Time:	0905 Samplers ID:		
SubGroups	FILE	VMV	NAME		ConcRpt	MDL	ConcRptUnit	InjDate		
G_C1C4										
	c0879	106778	Butane		10700.00	. (	05 ppmv	25-Mar-09		
	c0879	106779	Ethane		70500.00	. (	05 ppmv	25-Mar-09		
	c0879	106780	Ethylene		0.00	. (	05 ppmv	25-Mar-09		
	c0879	106781	Isobutane		7640.00	. (	05 ppmv	25-Mar-09		
	c0879	106782	Methane		793000.00	. (	05 ppmv	25-Mar-09		
	c0879	106783	Propane		36300.00	. (	05 ppmv	25-Mar-09		
G_TCD										
	G0879	106784	Carbon dioxide		744.00	300.0	00 ppmv	26-Mar-09		
	G0879	106785	Nitrogen		32700.00	1000.0	00 ppmv	26-Mar-09		
	G0879	107107	Oxygen		0.00	1000.	00 ppmv	26-Mar-09		

[ARC\_Remarks]: Tedlar Bag

SubGro	oups					
DG_C1	C4 and DG_TCD	- Diso	lved Gas in water sample			
G_C1C	4 and G_TCD	) - Free	Gas from canister			
Certif	ied For: Yoge	sh Kumar,	Business Unit Manager	Mail To:	Jones, Don	
			Environmental Monitoring		Sustainable Ecosystems	
Ву:			Alberta Research Council Vegreville, Alberta		Alberta Research Council	
			T9C 1T4		Calgary, Alberta	
Date:	30-Mar-09	Contact:	Grant Prill 780 632-8455		T2L 2A6	
					÷	

Alberta Research Council Environmental Monitoring Vegreville, Alberta

## Tabular Data Report

page 1

Sample	No: T09-	0874	Comments: 11-18 Exso	lved						
			SmpDate: 19-Mar-09	Time: 1018 By:	Matı	cix:				
			Canister #:	User Sample No:						
	Sub	Crow			Concontratio	22				
FTTF	DT	MO	NAME		ug/I	511	MM	MolFormula	CAS	
					ug/ 11			HOIFOIMUIA	CAD	
	Analysis	Date	a: 31-MAR-2009 05:40							
V0874A	0.00	Т	Chloromethane			0.0	50	CH3C1	74-87-3	
V0874A	0.00	т	Vinyl chloride			0.0	63	C2H3C1	75-01-4	
V0874A	0.00	Т	Bromomethane			0.0	95	CH3Br	74-83-9	
V0874A	0.00	Т	Chloroethane			0.0	65	C2H5C1	75-00-3	
V0874A	0.00	Т	Trichlorofluoromethane			0.0				
V0874A	0.00	Т	1,1-Dichloroethylene			0.0	96	C2H2C12	75-35-4	
V0874A	0.00	Т	Methylene chloride			.0	84	CH2C12	75-09-2	
V0874A	0.00	Т	trans-1,2-Dichloroethylene			0.0	97	C2H2C12	156-60-5	
V0874A	0.00	Т	MTBE			0.0	88	C5H12O	1634-04-4	
V0874A	0.00	т	1,1-Dichloroethane			0.0	98	C2H4C12	75-34-3	
V0874A	0.00	т	cis-1,2-Dichloroethylene			0.0	97	C2H2C12	156-59-4	
V0874A	0.00	т	2,2-Dichloropropane			0.0				
V0874A	0.00	т	Chloroform			0.0	119	CHC13	67-66-3	
V0874A	0.00	Т	1,1,1-Trichloroethane			0.0	132	C2H3C13	71-55-6	
V0874A	0.00	Т	1,2-Dichloroethane			0.0	98	C2H4C12	107-06-2	
V0874A	0.00	т	1,1-Dichloropropylene			0.0				
V0874A	0.00	Т	Carbon tetrachloride			0.0	154	CC14	56-23-5	
V0874A	0.00	Т	1,2-Dichloropropane			0.0	113	C3H6C12	78-87-5	
V0874A	0.00	Т	Trichloroethylene			0.0	131	C2HC13	79-01-6	
V0874A	0.00	Т	Dibromomethane			0.0				
V0874A	0.00	Т	Bromodichloromethane			0.0	162	CHBrC12	75-27-4	
V0874A	0.00	Т	2-Chloroethoxyethylene			0.0				
V0874A	0.00	Т	cis-1,3-Dichloropropylene			0.0	111	C3H4C12	542-75-6	
V0874A	0.00	Т	trans-1,3-Dichloropropylen	e		0.0	111	C3H4C12	542-75-6	
V0874A	0.00	Т	1,1,2-Trichloroethane			0.0	132	C2H3C13	79-00-5	
V0874A	0.00	Т	1,3-Dichloropropane			0.0				
V0874A	0.00	Т	Dibromochloromethane			0.0				
V0874A	0.00	т	1,2-Dibromoethane			0.0	188	C2H4Br2	106-93-4	
V0874A	0.00	Т	Tetrachloroethylene			0.0	166	C2C14	127-18-4	
V0874A	0.00	Т	Chlorobenzene			0.0	113	C6H5C1	108-90-7	
V0874A	0.00	т	1,1,1,2-Tetrachloroethane			0.0				
V0874A	0.00	Т	Bromoform			0.0	250	CHBr3	75-25-2	
V0874A	0.00	Т	Styrene			0.0	104	C8H8	100-42-5	
V0874A	0.00	Т	1,1,2,2-Tetrachloroethane			0.0	166	C2H2C14	79-34-5	
V0874A	0.00	Т	1,2,3-Trichloropropane			0.0				
V0874A	0.00	Т	Bromobenzene			0.0				
V0874A	0.00	Т	2-Chlorotoluene			0.0				
V0874A	0.00	Т	4-Chlorotoluene		5	0.0				
J0874A	0.00	Т	tert-Butylbenzene			0.0	134	C10H14	98-06-6	
V0874A	0.00	Т	1,3-Dichlorobenzene			0.0	147	C6H4C12	541-73-1	
/0874A	0.00	Т	sec-Butylbenzene			0.0	134	C10H14	135-98-8	
V0874A	0.00	т	1,4-Dichlorobenzene			0 0	147	C6HAC12	106-46-7	
		-	-,			v. v	12/	00114012	100-40-1	

page 2

Sample	No:	T09-	0874
--------	-----	------	------

Comments: 11-18 Exsolved

SmpDate: 19-Mar-09 Time: 1018 By: Canister #: User Sample No: Matrix:

	Sub	Group	: vpp	Concentration			
FILE	RT	MQ	NAME	ug/L	MW	MolFormula	CAS
	Analysis	Date	: 31-MAR-2009 05:40				
V0874A	0.00	Т	p-Isopropyltoluene	0.0	134	C10H14	99-87-6
V0874A	0.00	Т	1,2-Dichlorobenzene	0.0	146	C6H4C12	95-50-1
V0874A	0.00	Т	n-Butylbenzene	0.0	134	C10H14	104-51-8
V0874A	0.00	Τ	1,2-Dibromo-3-chloropropane	0.0			
V0874A	0.00	Τ	1,2,4-Trichlorobenzene	0.0	180	C6H3C13	120-82-1
V0874A	0.00	Т	1,2,3-Trichlorobenzene	0.0			
V0874A	4.70	32	Butane, 2-nitro-	1.7	103	C4H9NO2	600-24-8
V0874A	5.40	40	Pentane	1.1	72	C5H12	109-66-0
V0874A	6.84	25	Acetic acid, methyl ester	1.1	74	C3H6O2	79-20-9
V0874A	7.92	72	Cyclopentane	2.4	70	C5H10	287-92-3
V0874A	11.68	43	1-Pentene, 2-methyl-	2.1	84	C6H12	763-29-1
V0874A	13.12	I	1,2-Dichloroethane-d4	.0			
V0874A	14.05	I	Benzene-d6	.0			
V0874A	14.21	Τ	Benzene	21.1	78	С6Н6	71-43-2
V0874A	19.27	64	Cyclohexane, methyl-	.6	98	C7H14	108-87-2
V0874A	23.12	I	Toluene-d8	.0	100	C7D8	2037-26-5
V0874A	23.49	Т	Toluene	35.5	92	C7H8	108-88-3
V0874A	32.55	Т	Ethyl benzene	3.0	106	C8H10	100-41-4
V0874A	33.30	Т	m,p-Xylene	9.4	106	C8H10	108-38-3 / 106-42-3
V0874A	35.51	Т	o-Xylene	3.4	106	C8H10	95-47-6
V0874A	38.40	I	4-Bromofluorobenzene	.0			
V0874A	38.53	Т	Isopropylbenzene	. 4	120	C9H12	98-82-8
V0874A	40.52	Т	n-Propylbenzene	.2	120	C9H12	103-65-1
V0874A	40.93	93	Benzene, 1-ethyl-4-methyl-	1.2	120	C9H12	622-96-8
V0874A	40.98	70	Benzene, 1-ethyl-3-methyl-	. 4	120	C9H12	620-14-4
V0874A	41.23	Т	1,3,5-Trimethylbenzene	.3	120	C9H12	108-67-8
V0874A	41.72	76	Benzene, (1-methylethyl)-	.8	120	C9H12	98-82-8
V0874A	42.26	Т	1,2,4-Trimethylbenzene	1.9	120	C9H12	95-63-6
V0874A	43.25	94	Benzene, 1,2,3-trimethyl-	2.0	120	C9H12	526-73-8
V0874A	43.66	55	Benzene, 1-propenyl-	.5	118	C9H10	637-50-3
V0874A	44.02	55	Benzene, 1,2-diethyl-	.3	134	C10H14	135-01-3
V0874A	44.06	62	Benzene, 1-methyl-3-propyl-	.5	134	C10H14	1074-43-7
V0874A	44.21	59	5,6-DIMETHYLIDENE-2-NORBORNANONE	.6	134	C9H100	62289-63-8
V0874A	44.25	90	Benzene, 4-ethyl-1,2-dimethyl-	.7	134	C10H14	934-80-5
V0874A	44.50	86	Benzene, 1-methyl-2-propyl-	.5	134	C10H14	1074-17-5
V0874A	44.74	94	Benzene, 1-methyl-3-(1-methylethyl)- (CA	. 9	134	C10H14	535-77-3
V0874A	44.78	94	Benzene, 1-ethyl-2,3-dimethyl-	. 6	134	C10H14	933-98-2
V0874A	44.93	95	Benzene, 2-ethyl-1,3-dimethyl-	1.3	134	C10H14	2870-04-4
V0874A	45.00	72	Benzene, (2-methyl-1-propenyl)-	.5	132	C10H12	768-49-0
V0874A	45.09	90	.alphaMethylstyrene	1.0	118	C9H10	98-83-9
V0874A	45.14	64	Undecane	. 8	156	C11H24	1120-21-4
V0874A	45.32	23	Oxetane, 2-propyl-	.7	100	C6H12O	4468-64-8
V0874A	45.42	83	Benzene, 1-methyl-4-(1-methylethyl)- (CA	.5	134	C10H14	99-87-6
V0874A	45.62	95	Benzene, 1-ethyl-2,3-dimethyl-	. 8	134	C10H14	933-98-2
V0874A	45.71	94	Benzene, 1,2,3,4-tetramethyl-	1.7	134	C10H14	488-23-3
V0874A	46.12	64	Benzene, 2-ethenyl-1,4-dimethyl-	.6	132	C10H12	2039-89-6
V0874A	46.20	47	METHYL N-DIMETHYLTHIOCARBAMATE	.3	119	C4H9NOS	0-00-0
V0874A	46.36	74	Benzene, 1-methyl-2-(2-propenyl)-	1.9	132	C10H12	1587-04-8
V0874A	46.58	55	Naphthalene, 1,2,3,4-tetrahydro-	. 6	132	C10H12	119-64-2
V0874A	46.96	70	Dodecane	.9	170	C12H26	112-40-3

Sample	No: T09-	0874	Comments: 11-18 Exs	olved					
			SmpDate: 19-Mar-09	Time: 1018	By:	Matrix:			
			Canister #:	User Sample	No:				
	Sub	Group	: vpp			Concentration			
FILE	RT	MQ	NAME			ug/L	MW	MolFormula	CAS
	Analysis	Date	: 31-MAR-2009 05:40						
V0874A	47.01	Т	Naphthalene			1.1	128	C10H8	91-20-3
V0874A	47.19	30	trans-2-Phenyl-2-pentene			.4	146	C11H14	0-00-0
V0874A	47.49	т	Hexachlorobutadiene			.6	261	C4C16	87-68-3
V0874A	47.65	35	2,4-Dimethylamphetamine			. 4	163	C11H17N	75659-61-9
				sui	m :	107			

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Sample No: T09-0875

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Comments: 11-18 Exsolved

SmpDate: 19-Mar-09 Time: 1018 By: Canister #: User Sample No:

Matrix:

	Sub	Group	: epp	Concentration			
FILE	RT	MQ	NAME	ug/L	MW	MolFormula	CAS
	Analysis	Date	: 26-MAR-2009 17:45				
E0875	0.00	Т	2-Chlorophenol	0.0	128	C6H5C10	95-57-8
E0875	0.00	Т	Bis(2-chloroisopropyl)ether	0.0	170	C6H12C12O	39638-32-9
E0875	0.00	Т	Hexachloroethane	0.0	234	C2C16	67-72-1
E0875	0.00	Т	N-Nitroso-di-n-propylamine	0.0	130	C6H14N2O	621-64-7
E0875	0.00	Т	Nitrobenzene	0.0	123	C6H5NO2	98-95-3
E0875	0.00	Т	Isophorone	0.0	138	C9H14O	78-59-1
E0875	0.00	Т	2-Nitrophenol	0.0	139	C6H5NO3	88-75-5
E0875	0.00	Т	2,4-Dimethylphenol	0.0	122	C8H10O	105-67-9
E0875	0.00	Т	Bis(2-chloroethoxy)methane	0.0	172	C5H10C12O2	111-91-1
E0875	0.00	Т	2,4-Dichlorophenol	0.0	162	C6H4C12O	120-83-2
E0875	0.00	Т	1,2,4-Trichlorobenzene	0.0	180	C6H3Cl3	120-82-1
E0875	0.00	Т	Hexachlorobutadiene	0.0	261	C4C16	87-68-3
E0875	0.00	Т	4-Chloro-3-methylphenol	0.0	142	C7H7Cl0	59-50-7
E0875	0.00	Т	Hexachlorocyclopentadiene	0.0	270	C5C16	77-47-4
E0875	0.00	Т	2,4,6-Trichlorophenol	0.0	196	C6H3C130	88-06-2
E0875	0.00	Т	2-Chloronaphthalene	0.0	162	C10H7C1	91-58-7
E0875	0.00	Т	Acenaphthylene	0.0	152	C12H8	208-96-8
E0875	0.00	т	Dimethyl phthalate	0.0	194	C10H10O4	131-11-3
E0875	0.00	Т	2,6-Dinitrotoluene	0.0	182	C7H6N2O4	606-20-2
E0875	0.00	т	2.4-Dinitrophenol	0.0	184	C6H4N2O5	51-28-5
E0875	0.00	T	4-Nitrophenol	0.0	139	C6H5NO3	100-02-7
E0875	0.00	T	2 3 4 6-Tetrachlorophenol	0.0	230	C6H2C140	58-90-2
E0075	0.00	T	2 4-Dinitrotoluene	b.o	182	C7H6N2O4	121-14-2
E0075	0.00	T.	4-Chlorophenyl phenyl ether	0.0	204	C1289C10	7005-72-3
E0075	0.00	- -	A-Childfophenyi phenyi ether	0.0	201	C12H1404	94-66-2
EUG75	0.00	T	2-Mothyl=4 6-dipitrophopol	0.0	100	C746N205	524-52-1
EUG/J	0.00	T	2-Methyl-4, 0-dinitiophenor	0.0	104	C12412N2	122-66-7
E0675	0.00	1	1,2-Diphenyinyulazine	0.0	104	CI2HIZN2	101 55 3
EU875	0.00	1	4-Bromophenyi phenyi ether	0.0	240	CIZH9BIO	101-55-5
E0875	0.00	Т	Hexachlorobenzene	0.0	282	COCIO	118-74-1
E08/5	0.00	T	Pentachlorophenol	0.0	264	COHCIO	87-86-5
E0875	0.00	Т	Anthracene	0.0	1/8	C14H10	120-12-7
E0875	0.00	Т	Fluoranthene	0.0	202	C16H10	206-44-0
E0875	0.00	Т	Pyrene	0.0	202	C16H10	129-00-0
E0875	0.00	Т	Benzidine	0.0	184	C12H12N2	92-87-5
E0875	0.00	Т	Di-n-octyl phthalate	0.0	376	C23H36O4	117-84-0
E0875	0.00	Т	Benzo(b)fluoranthene	0.0	252	C20H12	205-99-2
E0875	0.00	Т	Benzo(k)fluoranthene	0.0	252	C20H12	207-08-9
E0875	0.00	Т	Benzo(a)pyrene	0.0	252	C20H12	50-32-8
E0875	0.00	Т	Indeno(1,2,3-cd)pyrene	0.0	276	C22H12	193-39-5
E0875	0.00	Т	Dibenzo(ah)anthracene	0.0	278	C22H14	53-70-3
E0875	0.00	Т	Benzo(ghi)perylene	0.0	276	C22H12	191-24-2
E0875	4.25	90	Pentanoic acid	10.0	102	C5H10O2	109-52-4
E0875	7.19	Т	Phenol	.8	94	C6H6O	108-95-2
E0875	7.40	Т	Bis(2-chloroethyl)ether	. 3	142	C4H8C120	111-44-4
E0875	8.63	97	Pyridine, 5-ethyl-2-methyl-	2.1	121	C8H11N	104-90-5
E0875	9.64	94	Ethanone, 1-phenyl-	9.9	120	C8H8O	98-86-2
E0875	9.91	96	Phenol, 4-methyl-	62.1	108	C7H8O	106-44-5
E0875	10.30	91	Benzenemethanol, .alpha.,.a	61.6	136	C9H12O	617-94-7
E0875	11.20	86	Hexanoic acid, 2-ethyl- (CA	2.1	144	C8H16O2	149-57-5
E0875	12.92	I	Naphthalene-d8	.0	136	C10D8	1146-65-2

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Sample No: T09-0875

Comments: 11-18 Exsolved

SmpDate: 19-Mar-09 Time: 1018 By: Matrix: Canister #: User Sample No:

	Sub	Group	: epp	Concentration			
FILE	RT	MQ	NAME	ug/L	MW	MolFormula	CAS
		-					
	Analysis	Date	: 26-MAR-2009 17:45				
E0875	13.00	Т	Naphthalene	.3	128	C10H8	91-20-3
E0875	15.66	94	Benzeneacetic acid	93.9	136	C8H8O2	103-82-2
E0875	16.05	97	1H-Indole	11.4	117	C8H7N	120-72-9
E0875	20.94	Т	Acenaphthene	.0	154	C12H10	83-32-9
E0875	21.24	97	pentadecane	1.5	212	C15H32	629-62-9
E0875	22.69	72	2,2-Di-(2'-chloroethoxy)-et	1.2	186	C6H12C12O2	0-00-0
E0875	23.27	Т	Fluorene	.0	166	C13H10	86-73-7
E0875	23.59	98	Hexadecane	1,2	226	C16H34	544-76-3
E0875	24.16	Т	N-Nitrosodiphenylamine	.1	198	C12H10N2O	86-30-6
E0875	25.83	98	Heptadecane	1.2	240	C17H36	629-78-7
E0875	27.25	43	4-Hexen-1-ol, 2-ethenyl-2,5	2.8	154	C10H180	50598-21-5
E0875	27.53	I	Phenanthrene-d10	.0	188	C14D10	1517-22-2
E0875	27.63	Τ	Phenanthrene	.0	178	C14H10	85-01-8
E0875	27.95	98	Octadecane	1.1	254	C18H38	593-45-3
E0875	29.98	98	Nonadecane	1.2	268	C19H40	629-92-5
E0875	31.27	Т	Di-n-butylphthalate	.2	278	C16H22O4	84-74-2
E0875	32.29	95	Phenol, 2,2'-methylenebis-	3.6	200	C13H12O2	2467-02-9
E0875	35.54	99	Docosane	1.1	310	C22H46	629-97-0
E0875	37.52	49	Neopentylidenecyclohexane	2.4	152	C11H20	39546-80-0
E0875	38.17	Т	Butylbenzylphthalate	. 8	312	C19H2004	85-68-7
E0875	38.45	91	Squalene	1.3	410	C30H50	7683-64-9
E0875	39.71	т	Benzo(a)anthracene	. 0	228	C18H12	56-55-3
E0875	39.73	I	Benzo(a)anthracene-d12	.0	240	C18D12	1719-03-5
E0875	39.98	Т	Chrysene	.0	228	C18H12	218-01-9
E0875	41.28	Т	Bis(2-ethylhexyl)phthalate	18.4	390	C24H38O4	117-81-7
E0875	46.90	98	Docosane	2.9	310	C22H46	629-97-0
E0875	49.33	97	Eicosane	2.4	282	C20H42	112-95-8
			sum:	298			

 FILE: datafile
 RT: retention time
 MQ: T=target compound or ##=PBM library match quality

 Flg: nd=not detected
 U=non-target compound or Unknown

 MDL: method detection limit
 MW: molecular weight
 CAS: chemical abstracts service

Certified For: Yogesh Kumar, Business Unit Manager Environmental Monitoring By: Alberta Research Council Vegreville, Alberta T9C 1T4 Date: Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 (780) 632-8455 Alberta Research Council Environmental Monitoring Vegreville, Alberta

## Tabular Data Report

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Sampie	NO. 109-0		SmpDate: 19-Mar-09 Time: 1012 By: DFJ Matrix: TEDLAR						
			Canister #:	User Sample No:					
	SubG	roup:	: VOC		Concentration				
FILE	RT	MQ	NAME		ppbv	MW	MolFormula	CAS	
	Analysis	 Date:	· 26-MAR-2009 11.01						
V0881	0.00	T Dace.	1-Butene		0.0	56	C4H8	106-98-9	
V0881	0.00	Ť	trans-2-Butene		0.0	56	C4H8	624-64-6	
10001	0.00	T	cis-2-Butane		0.0	56	САНВ	590-18-1	
V0881	0.00	T	3-Methyl-1-butene		0.0	70	C5H10	563-45-1	
V0881	0.00	T	1-Pentene		0.0	70	C5H10	109-67-1	
V0881	0.00	T	Isoprepe		0.0	68	C5H8	78-79-5	
V0881	0.00	T	trans-2-Pentene		0.0	70	C5H10	646-04-8	
V0001	0.00	T	cis-2-Pentene		0.0	70	C5H10	627-20-3	
V0881	0 00	T	2-Methyl-2-butene		0.0	70	C5H10	563-46-2	
V0881	0 00	Ť	Cyclopentene		0.0	68	С5Н8	142-29-0	
VOREI	0.00	Ť	4-Methyl-1-peptere		0.0	84	C6H12	691-37-2	
10001	0.00	T	2-Methyl-l-poptope		0.0	84	C6H12	763-20-1	
10001	0.00	T	cis=2=Hevene		0.0	84	C6H12	7688-21-3	
10001	0.00	T	trans-2-Hevene		0.0	04	CENIZ	1050-21-3	
VUCCI	0.00		2 4-Dimethylpontane		0.0	100	C7H16	100-00-7	
170001	0.00	m	2 2 4-Trimothylpontane		0.0	114	Coulo	565-75-2	
10001	0.00	T.	2, 5, 4-111Methylpentane		0.0	106	Conio	100-41-4	
10001	0.00	T	Nepapa		0.0	100	CONSO	111 84 2	
10001	0.00	1	Sturene		0.0	104	COH20	111-04-2	
10001	0.00	T	Styrene		0.0	104	CONIO	100-42-5	
VU881	0.00	1	o-xyrene		0.0	100	CONIC	95-47-6	
V0881	0.00	т	Isopropyidenzene		0.0	120	C9H12	98-82-8	
V0881	0.00	т	alpha Pinene		0.0	136	CIUH16	80-56-8	
V0881	0.00	Т	n-Propylbenzene		0.0	120	C9H12	103-65-1	
V0881	0.00	Т	1,3,5-Trimethylbenzene		0.0	120	C9H12	108-67-8	
V0881-	0.00	Т	beta Pinene		0.0	136	CIOHI6	18172-67-3	
V0881	0.00	Т	1,2,4-Trimethylbenzene		0.0	120	C9H12	95-63-6	
V0881	2.79	Т	Isobutane		65,500.0	58	C4H10	75-28-5	
V0881	3.00	Т	Butane		73,400.0	58	C4H10	106-97-8	
V0881	3.11	72	Propane, 2,2-dimethyl-	<b>N</b>	1,700.0	72	C5H12	463-82-1	
V0881	3.16	0	Unknown		1,420.0	0			
V0881	3.68	T	Isopentane		25,400.0	72	C5H12	78-78-4	
V0881	4.00	Т	Pentane		20,600.0	72	C5H12	109-66-0	
V0881	4.25	50	Acetone		459.0	58	СЗН6О	67-64-1	
V0881	4.49	Т	2,2-Dimethylbutane		806.0	86	C6H14	75-83-2	
V0881	4.95	Т	2,3-Dimethylbutane		869.0	86	C6H14	79-29-8	
V0881	4.97	Т	Cyclopentane		856.0	70	C5H10	287-92-3	
V0881	5.01	Т	2-Methylpentane		6,000.0	86	C6H14	107-83-5	
V0881	5.24	Т	3-Methylpentane		3,270.0	86	C6H14	96-14-0	
V0881	5.52	Т	Hexane		6,580.0	86	C6H14	110-54-3	
V0881	5.67	80	2-Butanone		1,260.0	72	C4H8O	78-93-3	
V0881	6.02	Т	Methylcyclopentane		1,510.0	84	C6H12	96-37-7	
V0881	6.60	Т	Cyclohexane		1,300.0	84	C6H12	110-82-7	

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Sample No: T09-0881 Comments: 11-18 Production Casing SmpDate: 19-Mar-09 Time: 1012 By: DFJ Matrix: TEDLAR Canister #: User Sample No:

	SubGroup: voc				Concentration				
FILE	RT	MQ	NAME		ppbv	MW	MolFormula	CAS	
	Analysis	Date	: 26-MAR-2009 11:01						
V0881	6.62	Т	Benzene		572.0	78	C6H6	71-43-2	
V0881	6.64	Т	2-Methylhexane		1,160.0	100	C7H16	591-76-4	
V0881	6.69	Т	2,3-Dimethylpentane		305.0	100	C7H16	565-59-3	
V0881	6.79	Т	3-Methylhexane		1,440.0	100	C7H16	589-34-4	
V0881	7.00	Т	2,2,4-Trimethylpentane		216.0	114	C8H18	540-84-1	
V0881	7.05	81	Isopropylcyclobutane		565.0	98	C7H14	872-56-0	
V0881	7.19	Т	Heptane		1,910.0	100	C7H16	142-82-5	
V0881	7.26	78	Propanal, 2,2-dimethyl-		229.0	86	C5H100	630-19-3	
V0881	7.63	Т	Methylcyclohexane		1,590.0	98	C7H14	108-87-2	
V0881	7.73	0	Unknown		90.5	0			
V0881	7.80	0	Unknown		99.7	0			
V0881	8.01	0	Unknown		134.0	0			
V0881	8.25	Т	2-Methylheptane		261.0	114	C8H18	592-27-8	
V0881	8.35	Т	Toluene		923.0	92	C7H8	108-88-3	
V0881	8.38	Т	3-Methylheptane		234.0	114	C8H18	589-81-1	
V0881	8.79	Т	Octane		417.0	114	C8H18	111-65-9	
V0881	9.59	I	Chlorobenzene-d5		.0	112	C6D5C1	3114-55-4	
V0881	9.94	Т	m,p-Xylene		97.4	106	C8H10	108-38-3 / 106-42-3	
				sum:	221,174				

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Sample No: T09-0882 Comments: 11-18 Production Tubing

SmpDate: Time: By: Canister #: User Sample No:

Matrix: TEDLAR

	Sub(	Froup	: voc	Concentration			
FILE	RТ	MO	NAME	vdaa	MW	MolFormula	CAS
	Analysis	Date	: 26-MAR-2009 11:35				
V0882	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
V0882	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
V0882	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
V0882	0.00	Т	3-Methyl-1-butene	0.0	70	C5H10	563-45-1
V0882	0.00	Т	1-Pentene	0.0	70	C5H10	109-67-1
V0882	0.00	Т	Isoprene	0.0	68	C5H8	78-79-5
V0882	0.00	Т	trans-2-Pentene	0.0	70	C5H10	646-04-8
V0882	0.00	Т	cis-2-Pentene	0.0	70	C5H10	627-20-3
V0882	0.00	Т	2-Methyl-2-butene	0.0	70	C5H10	563-46-2
V0882	0.00	Т	Cyclopentene	0.0	68	C5H8	142-29-0
V0882	0.00	Т	4-Methyl-1-pentene	0.0	84	C6H12	691-37-2
V0882	0.00	Т	2-Methyl-1-pentene	0.0	84	C6H12	763-29-1
V0882	0.00	Т	cis-2-Hexene	0.0	84	C6H12	7688-21-3
V0882	0.00	Т	trans-2-Hexene	0.0	84	C6H12	4050-45-7
V0882	0.00	Т	2,4-Dimethylpentane	0.0	100	C7H16	108-08-7
V0882	0.00	Т	2,2,4-Trimethylpentane	0.0	114	C8H18	540-84-1
V0882	0.00	Т	2,3,4-Trimethylpentane	0.0	114	C8H18	565-75-3
V0882	0.00	Т	Styrene	0.0	104	C8H8	100-42-5
V0882	0.00	Т	o-Xylene	0.0	106	C8H10	95-47-6
V0882	0.00	Т	Isopropylbenzene	0.0	120	C9H12	98-82-8
V0882	0.00	т	alpha Pinene	0.0	136	C10H16	80-56-8
V0882	0.00	Т	n-Propylbenzene	0.0	120	C9H12	103-65-1
V0882	0.00	Т	1,3,5-Trimethylbenzene	0.0	120	C9H12	108-67-8
V0882	0.00	Т	beta Pinene	0.0	136	C10H16	18172-67-3
V0882	0.00	Т	1,2,4-Trimethylbenzene	0.0	120	C9H12	95-63-6
V0882	2.78	т	Isobutane	43,500.0	58	C4H10	75-28-5
V0882	3.00	т	Butane	73,200.0	58	C4H10	106-97-8
V0882	3.10	64	Propane, 2,2-dimethyl-	1,130.0	72	C5H12	463-82-1
V0882	3.20	4	Methanol	80,700.0	32	CH40	67-56-1
V0882	3.69	т	Isopentane	103.000.0	72	C5H12	78-78-4
V0882	4.02	т	Pentane	136,000.0	72	C5H12	109-66-0
V0882	4.50	т	2.2-Dimethylbutane	3,400.0	86	C6H14	75-83-2
V0882	4.96	T	2.3-Dimethylbutane	5,390.0	86	C6H14	79-29-8
V0882	4.99	т	Cyclopentane	8,180.0	70	C5H10	287-92-3
V0882	5 02	т	2-Methylpentane	46,400.0	86	C6H14	107-83-5
V0882	5.26	T	3-Methylpentane	29,000,0	86	C6H14	96-14-0
V0882	5 54	T	Hevane	62 600 0	86	C6H14	110-54-3
10002	5 92	17	Heyane 2 4-dimethyl-	1 250 0	114	C8H18	580-13-5
V0882	6.03	ч, т	Methyl cyclopentane	12, 700, 0	84	C6H12	96-37-7
V0882	6.12	40	1-Hevene, 4-methyl-	12,700.0	98	C7H14	3769-23-1
10002	6 15	72	Pentane 3 3-dimethyl-	281 0	100	C7H16	562-49-2
170002	6 61	72 T	Cuclobeyane	12 600 0	100	CFH12	110-02-7
10002	6 62	T	Ponzone	2 210 0	70	CONIZ	71-42-2
10002	6 64	Ŧ	2-Methylbeyane	6,990,0	100	C7H16	501-76-4
V0802	6 60	T	2 3-Dimethylnentane	1 450 0	100	C7H16	565-50-3
10002	6 70	T	3-Methylbevane	7,450.0	100	C7416	589-34-4
10002	6.06	1	Cuclopentane 1 3-dimethul-	7,000.0	100	C7H14	2452-00-1
10002	0.90	54	Cyclopentane, 1,3-dimethyl-	2,280.0	90	C7U14	2403-00-1
100000	7.01	90	Cyclopentane, 1,5-dimethyl-, Cis-	1,720.0	98	C7H14	2002-08-0
VU682	7.05	94 T	Ventane	2,810.0	100	C7H16	142-02-5
VUODZ	1.17	1	1000000	11.200.0	TUU	U/DIU	

Sample No: T09-0882

Comments: 11-18 Production Tubing

\_\_\_\_\_ SmpDate: Time: By: Matrix: TEDLAR Canister #: User Sample No:

	SubG	roup	: VOC	Concentration			
FILE	RT	MQ	NAME	ppbv	MW	MolFormula	CAS
	Analysis	Date	: 26-MAR-2009 11:35				
V0882	7.54	33	Butane, 1-chloro-2-methyl-	83.6	106	C5H11C1	616-13-7
V0882	7.63	Т	Methylcyclohexane	6,870.0	98	C7H14	108-87-2
V0882	7.70	80	Hexane, 2,5-dimethyl-	382.0	114	C8H18	592-13-2
V0882	7.74	90	Hexane, 2,4-dimethyl-	371.0	114	C8H18	589-43-5
V0882	7.80	91	Cyclopentane, ethyl-	387.0	98	C7H14	1640-89-7
V0882	7.89	87	Cyclopentane, 1,2,4-trimethyl-	531.0	112	C8H16	2815-58-9
V0882	8.01	94	Cyclopentane, 1,2,3-trimethyl-, (1.alpha	471.0	112	C8H16	15890-40-1
V0882	8.18	64	Pyrrolidine	169.0	71	C4H9N	123-75-1
V0882	8.25	Т	2-Methylheptane	1,470.0	114	C8H18	592-27-8
V0882	8.35	Т	Toluene	1,600.0	92	C7H8	108-88-3
V0882	8.38	Т	3-Methylheptane	629.0	114	C8H18	589-81-1
V0882	8.52	93	Cyclohexane, 1,3-dimethyl-, cis-	897.0	112	C8H16	638-04-0
V0882	8.67	35	Cyclohexane, 1-ethyl-2-methyl-, trans- (	162.0	126	C9H18	4923-78-8
V0882	8.73	93	Cyclopentane, 1-ethyl-2-methyl-, cis-	185.0	112	C8H16	930-89-2
V0882	8.79	Т	Octane	1,270.0	114	C8H18	111-65-9
V0882	8.84	86	Cyclohexane, 1,2-dimethyl- (cis/trans)	307.0	112	C8H16	583-57-3
V0882	9.13	59	Heptane, 2,4-dimethyl-	117.0	128	C9H20	2213-23-2
V0882	9.23	80	Heptane, 2,6-dimethyl-	285.0	128	C9H20	1072-05-5
V0882	9.35	50	Nonanal	162.0	142	C9H18O	124-19-6
V0882	9.41	87	Cyclohexane, ethyl-	149.0	112	C8H16	1678-91-7
V0882	9.46	72	Cyclohexane, 1,1,3-trimethyl-	201.0	126	C9H18	3073-66-3
V0882	9.59	I	Chlorobenzene-d5	.0	112	C6D5C1	3114-55-4
V0882	9.67	37	1-Trifluoroacetoxydecane	105.0	254	C12H21F3O2	0-00-0
V0882	9.76	53	Hexane, 3,3-dimethyl-	190.0	114	C8H18	563-16-6
V0882	9.82	Т	Ethyl benzene	79.1	106	C8H10	100-41-4
V0882	9.87	50	Heptane, 3-ethyl-	165.0	128	C9H20	15869-80-4
V0882	9.94	Т	m,p-Xylene	136.0	106	C8H10	108-38-3 / 106-42-3
V0882	10.27	Т	Nonane	140.0	128	C9H20	111-84-2
			sum:	672,646			

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Dampie NO. 109 0000	Sample	No:	T09-0883
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Comments: 6-12 Surface Casing

SmpDate: 19-Mar-09 Time: 0905 By: Matrix: TEDLAR Canister #: User Sample No:

	Sub	Group	: voc	Concentration			
FILE	RT	MQ	NAME	ppbv	MW	MolFormula	CAS
	Analysis	Date	: 26-MAR-2009 12:09				
V0883	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
V0883	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
V0883	0.00	Т	3-Methyl-1-butene	0.0	70	C5H10	563-45-1
V0883	0.00	Т	1-Pentene	0.0	70	C5H10	109-67-1
V0883	0.00	Т	Isoprene	0.0	68	C5H8	78-79-5
V0883	0.00	Т	trans-2-Pentene	0.0	70	C5H10	646-04-8
V0883	0.00	Т	cis-2-Pentene	0.0	70	C5H10	627-20-3
V0883	0.00	Т	2-Methyl-2-butene	0.0	70	C5H10	563-46-2
V0883	0.00	т	Cyclopentene	0.0	68	C5H8	142-29-0
V0883	0.00	Т	4-Methyl-1-pentene	0.0	84	C6H12	691-37-2
V0883	0.00	Т	2-Methyl-1-pentene	0.0	84	C6H12	763-29-1
V0883	0.00	Т	cis-2-Hexene	0.0	84	C6H12	7688-21-3
V0883	0.00	т	trans-2-Hexene	0.0	84	C6H12	4050-45-7
V0883	0.00	Т	2.4-Dimethylpentane	0.0	100	C7H16	108-08-7
V0883	0 00	T	2.2.4-Trimethylpentane	0.0	114	C8H18	540-84-1
100003	0.00	T	2 3 4-Trimethylpentane	0.0	114	C8H18	565-75-3
170883	0.00	T	Toluene	0.0	92	C7H8	108-88-3
10003	0.00	т т	3-Methylhentane	0.0	114	C8H18	589-81-1
170003	0.00	- 	Ethyl bonzone	0.0	106	CONIO	100-41-4
170003	0.00	±	m n-Yulono	0.0	106	CONTO	100-30-3 / 106-42-3
10000	0.00	T T	Menano	0.0	120	CONIO	111-04-2
V0003	0.00		Shumana	0.0	104	CONO	100 42 5
VU883	0.00	T	Styrene	0.0	104	COND	100-42-5
V0883	0.00	T	o-xylene	0.0	106	CSHIU	95-47-6
V0883	0.00	T	Isopropyidenzene	0.0	120	C9H12	98-82-8
V0883	0.00	Т	alpha Pinene	0.0	136	CIUHI6	80-56-8
V0883	0.00	Т	n-Propylbenzene	0.0	120	C9H12	103-65-1
V0883	0.00	Т	1,3,5-Trimethylbenzene	0.0	120	C9H12	108-67-8
V0883	0.00	Т	beta Pinene	0.0	136	C10H16	18172-67-3
V0883	0.00	Т	1,2,4-Trimethylbenzene	0.0	120	C9H12	95-63-6
V0883	2.80	Т	Isobutane	4,180,000.0	58	C4H10	75-28-5
V0883	3.02	Т	Butane	5,670,000.0	58	C4H10	106-97-8
V0883	3.25	Т	cis-2-Butene	17,000.0	56	C4H8	590-18-1
V0883	3.69	Т	Isopentane	2,170,000.0	72	C5H12	78-78-4
V0883	4.02	Т	Pentane	1,650,000.0	72	C5H12	109-66-0
V0883	4.18	27	Cyclopropane	1,120.0	42	СЗН6	75-19-4
V0883	4.50	Т	2,2-Dimethylbutane	28,900.0	86	C6H14	75-83-2
V0883	4.95	Т	2,3-Dimethylbutane	48,400.0	86	C6H14	79-29-8
V0883	4.99	Т	Cyclopentane	33,300.0	70	C5H10	287-92-3
V0883	5.02	Т	2-Methylpentane	303,000.0	86	C6H14	107-83-5
V0883	5.25	Т	3-Methylpentane	171,000.0	86	C6H14	96-14-0
V0883	5.53	Т	Hexane	282,000.0	86	C6H14	110-54-3
V0883	6.02	Т	Methylcyclopentane	55,400.0	84	C6H12	96-37-7
V0883	6.45	23	Pentane, 3,3-dimethyl-	1,030.0	100	C7H16	562-49-2
V0883	6.61	Т	Cyclohexane	25,300.0	84	C6H12	110-82-7
V0883	6.61	Т	Benzene	2,190.0	78	C6H6	71-43-2
V0883	6.64	Т	2-Methylhexane	23,000.0	100	C7H16	591-76-4
V0883	6.69	Т	2,3-Dimethylpentane	6,680.0	100	C7H16	565-59-3
V0883	6.80	Т	3-Methylhexane	24,200.0	100	C7H16	589-34-4
V0883	6.95	76	Cyclopentane, 1,3-dimethyl-	13,500.0	98	C7H14	2453-00-1
V0883	7.00	59	Cyclopentane, 1,3-dimethyl-, trans-	8,330.0	98	C7H14	1759-58-6

Sample No: T09-0883 Comments: 6-12 Surface Casing SmpDate: 19-Mar-09 Time: 0905 By: Matrix: TEDLAR Canister #: User Sample No: SubGroup: voc Concentration FILE RT MQ NAME ppbv MW MolFormula CAS \_\_\_\_\_ \_\_\_\_ Analysis Date: 26-MAR-2009 12:09 V0883 7.05 93 CYCLOBUTANE, ISOPROPYL-11,000.0 98 C7H14 0-00-0 21,700.0 100 C7H16 V0883 7.19 T Heptane 142-82-5 20,100.0 V0883 7.63 T Methylcyclohexane 98 C7H14 108-87-2 V0883 7.89 45 2-(1-Methylethyl)-1,3-propanediol 1,180.0 118 C6H14O2 0-00-0 V0883 8.26 T 2-Methylheptane 1,820.0 114 C8H18 592-27-8 V0883 8.38 45 Octane, 3,4-dimethyl-1,580.0 142 C10H22 15869-92-8 V0883 8.52 43 Cyclohexane, 1,2-dimethyl-, cis-1,620.0 112 C8H16 2207-01-4 V0883 8.79 T Octane 2,000.0 114 C8H18 111-65-9 V0883 9.59 I Chlorobenzene-d5 .0 112 C6D5C1 3114-55-4 sum: 14,775,350

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 FILE: datafile
 RT: retention time
 MQ: T=target compound or ##=PBM library match quality

 Flg: nd=not detected
 U=non-target compound or Unknown

 MDL: method detection limit
 MW: molecular weight
 CAS: chemical abstracts service

Certified For: Yogesh Kumar, Business Unit Manager Environmental Monitoring By: Alberta Research Council Vegreville, Alberta T9C 1T4 Date: Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Contact Person: Grant Prill Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4

University of Alberta - Isotope Lab

Sample ID	date sampled	Well Location	Sample Type	$\delta^{13}C C_1$	$\delta^{13}C C_2$	$\delta^{13}C C_3$	$\delta^{13}$ C i-C <sub>4</sub>	$\delta^{13}$ C n-C <sub>4</sub>	$\delta^{13}C\ CO_2$
	19-Mar-09	11-18-?	PRODUCTION CASING	-43.6	-26.84	-24.44	-29.61	-22.93	8.6
	19-Mar-09	11-18-?	EXSOLVED GAS	-43.49	-26.27	-26.33	-27.08	-27.75	5.3
	19-Mar-09	11-18-?	PRODUCTION TUBING	-44.01	-26.4	-26.61	-25.76	-28.62	
	19-Mar-09	6-12-?	SCV	-47.86	-30.98	-27.32	-28.21	-28.35	

### Appendix B

### Gas Migration Analytical Reports, Location Maps and Field LEL Data



THE STATED SCALE MAY NOT AGREE WITH SCALEBAR WHEN THIS DOCUMENT IS VIEWED IN PDF FORMAT OR IS PRINTED FROM A PDF FORMA



THE STATED SCALE MAY NOT AGREE WITH SCALEBAR WHEN THIS DOCUMENT IS VIEWED IN PDF FORMAT OR IS PRINTED FROM A PDF FORMA

#### Alberta Research Council Environmental Monitoring Vegreville, Alberta

## Tabular Data Report

page 1

Sample	ample No: T09-2954 Comments: Baseline- ERCB-PennWest 6-12-078-08 W6M							
			SmpDate: 16-Sep-09	Time: 0910 By:	CRG Matrix: S	ILCO		
			Canister #: 2443	User Sample No:	SVW-15			
	SubC	Group	: TCD		Concentration			
FILE	RT	MQ	NAME		percent	MW	MolFormula	CAS
	Analysis	Date	:					
G2954	0.00	Т	Carbon monoxide		0.0	28	CO	630-08-0
G2954	1.52	Т	Carbon dioxide		.5	44	CO2	124-38-9
G2954	3.33	Т	Oxygen		20.9	32	02	7782-44-7
G2954	6.07	Т	Nitrogen		77.6	28	N2	7727-37-9
				sum:	99			
	SubG	Group	: clc4		Concentration			
FILE	RT	MQ	NAME		ppmv	MW	MolFormula	CAS
	Analysis	Date	: 22-SEP-2009 00:00					
c2954	0.00	Т	Ethane		0.0	30	C2H6	74-84-0
c2954	0.00	Т	Propane		0.0	44	СЗН8	74-98-6
c2954	0.00	Т	Propylene		0.0	42	СЗН6	115-07-1
c2954	0.00	Т	Isobutane		0.0	58	C4H10	75-28-5
c2954	0.00	Т	Acetylene		0.0	26	C2H2	74-86-2
c2954	0.00	Т	Butane		0.0	58	C4H10	106-97-8
c2954	0.00	Т	trans-2-Butene		0.0	56	C4H8	624-64-6
c2954	0.00	Т	1-Butene		0.0	56	C4H8	106-98-9
c2954	0.00	Т	Isobutylene		0.0	56	C4H8	115-11-7
c2954	0.00	Т	cis-2-Butene		0.0	56	C4H8	590-18-1
c2954	0.00	Т	Propyne		0.0	40	C3H4	74-99-7
c2954	0.00	Т	1,3-Butadiene		0.0	54	C4H6	106-99-0
c2954	0.00	Т	Ethylacetylene		0.0	54	C4H6	107-00-6
c2954	1.74	Т	Methane		2.1	16	CH4	74-82-8
c2954	2.87	Т	Ethylene		0.0	28	C2H4	74-85-1
				sum:	2			
page 2

Sample No: T09-2955

Comments: Baseline- ERCB-PennWest 6-12-078-08 W6M SmpDate: 16-Sep-09 Time: 0925 By: CRG Matrix: SILCO Canister #: 2518 User Sample No: SVW-6

	SubG	Group	: TCD	Concentration			
FILE	RT	MQ	NAME	percent M	W	MolFormula	CAS
	Analysis	Date	:				
G2955	0.00	Т	Carbon monoxide	0.0	28	со	630-08-0
G2955	1.51	Т	Carbon dioxide	2.1	44	CO2	124-38-9
G2955	3.34	Т	Oxygen	5.3	32	02	7782-44-7
G2955	6.12	Т	Nitrogen	38.7	28	N2	7727-37-9
			sum:	46			

	SubC	Froup	: clc4	Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysis	Date	: 22-SEP-2009 00:00				
c2955	0.00	Т	Ethylene	0.0	28	C2H4	74-85-1
c2955	0.00	Т	Propylene	0.0	42	C3H6	115-07-1
c2955	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c2955	0.00	Т	Butane	0.0	58	C4H10	106-97-8
c2955	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
c2955	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
c2955	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c2955	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c2955	0.00	Т	Propyne	0.0	40	C3H4	74-99-7
c2955	0.00	Т	1,3-Butadiene	0.0	54	C4H6	106-99-0
c2955	0.00	Т	Ethylacetylene	0.0	54	C4H6	107-00-6
c2955	1.73	Т	Methane	580,000.0	16	CH4	74-82-8
c2955	2.20	Т	Ethane	5,750.0	30	C2H6	74-84-0
c2955	4.23	Т	Propane	9.2	44	СЗН8	74-98-6
c2955	7.81	Т	Isobutane	11.4	58	C4H10	75-28-5
			sum:	585,771			

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Sample	No: T09-:	2956	Comments: SmpDate: Canister #:	Baseline- 16-Sep-09 2654	ERCB-Pe Time: User S	ennWest ( 0940 By Sample No	5-12-078- 7: CRG 5: SVW-2	08 W6M Matrix: SI	LCO		
	Sub	Group	: TCD				Conc	entration			
FILE	RT	MQ	NAME					percent	MW	MolFormula	CAS
	Analysis	Date	:								
G2956	0.00	Т	Carbon monoxide					0.0	28	CO	630-08-0
G2956	1.52	Т	Carbon dioxide					.4	44	C02	124-38-9
G2956	3.33	Т	Oxygen					19.6	32	02	7782-44-7
G2956	6.07	Т	Nitrogen					71.9	28	N2	7727-37-9
						sum:		92			
	Sub0	Group	: clc4				Conc	entration			
FILE	RT	MQ	NAME					ppmv	MW	MolFormula	CAS
	Analysis	Date	: 22-SEP-2009 00:	00							
c2956	0.00	Т	Ethylene					0.0	28	C2H4	74-85-1
c2956	0.00	Т	Propylene					0.0	42	СЗНб	115-07-1
c2956	0.00	Т	Acetylene					0.0	26	C2H2	74-86-2
c2956	0.00	Т	trans-2-Butene					0.0	56	C4H8	624-64-6
c2956	0.00	Т	1-Butene					0.0	56	C4H8	106-98-9
c2956	0.00	Т	Isobutylene					0.0	56	C4H8	115-11-7
c2956	0.00	т	cis-2-Butene					0.0	56	C4H8	590-18-1
c2956	0.00	Т	Propyne					0.0	40	СЗН4	74-99-7
c2956	0.00	т	1,3-Butadiene					0.0	54	C4H6	106-99-0
c2956	0.00	т	Ethvlacetvlene					0.0	54	C4H6	107-00-6
c2956	1.73	т	Methane					64,900.0	16	CH4	74-82-8
c2956	2.20	Т	Ethane					290.0	30	C2H6	74-84-0
c2956	4.22	T	Propane					29.8	44	СЗНЯ	74-98-6
02956	7.79	Ť	Isobutane					6.2	58	C4H10	75-28-5
trade of addition	1 . 1 .	±.	2002000000000					V . L	00	U . III U	10 40 0

sum:

65,229

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Sample No: T09-2957

Comments: Baseline- ERCB-PennWest 6-12-078-08 W6M SmpDate: 16-Sep-09 Time: 0950 By: CRG Matrix: SILCO Canister #: 2447 User Sample No: SVW-1

	Sub	Group	: TCD	Concentration				
FILE	RT	MQ	NAME	percent	MW	MolFormula	CAS	
	Analysis	Date	;					
G2957	0.00	Т	Carbon monoxide	0.0	28	CO	630-08-0	
G2957	1.51	Т	Carbon dioxide	. 8	44	C02	124-38-9	
G2957	3.32	Т	Oxygen	12.2	32	02	7782-44-7	
G2957	6.09	Т	Nitrogen	55.2	28	N2	7727-37-9	
			sum:	68				

	SubC	Group	: clc4	Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysis	Date	: 22-SEP-2009 00:00				
c2957	0.00	Т	Ethylene	0.0	28	C2H4	74-85-1
c2957	0.00	Т	Propylene	0.0	42	C3H6	115-07-1
c2957	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c2957	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
c2957	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
c2957	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c2957	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c2957	0.00	Т	Propyne	0.0	40	C3H4	74-99-7
c2957	0.00	Т	1,3-Butadiene	0.0	54	C4H6	106-99-0
c2957	0.00	Т	Ethylacetylene	0.0	54	C4H6	107-00-6
c2957	1.73	Т	Methane	326,000.0	16	CH4	74-82-8
c2957	2.20	Т	Ethane	2,700.0	30	C2H6	74-84-0
c2957	4.23	Т	Propane	358.0	44	СЗН8	74-98-6
c2957	7.81	Т	Isobutane	55.4	58	C4H10	75-28-5
c2957	8.24	Т	Butane	19.5	58	C4H10	106-97-8
			sum:	329,133			

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Sample No: T09-3038

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Comments: Baseline- ERCB-PennWest 11-12-078-08 W6M SmpDate: 16-Sep-09 Time: 1055 By: CRG Matrix: SILCO Canister #: 1964 User Sample No: SVW-15

	Sub	Group	: TCD	Concentration			
FILE	RT	MQ	NAME	percent	MW	MolFormula	CAS
	Analysis	Date	:				
G3038	0.00	Т	Carbon monoxide	0.0	28	CO	630-08-0
G3038	1.53	Т	Carbon dioxide	.2	44	C02	124-38-9
G3038	3.32	Т	Oxygen	20.6	32	02	7782-44-7
G3038	6.05	Т	Nitrogen	78.5	28	N2	7727-37-9
			. sum:	99			

	SubGroup: clc4			Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysis	Date	: 24-SEP-2009 00:00				
c3038	0.00	Т	Ethane	0.0	30	C2H6	74-84-0
c3038	0.00	Т	Ethylene	0.0	28	C2H4	74-85-1
c3038	0.00	Т	Propane	0.0	44	C3H8	74-98-6
c3038	0.00	Т	Propylene	0.0	42	СЗНб	115-07-1
c3038	0.00	Т	Isobutane	0.0	58	C4H10	75-28-5
c3038	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c3038	0.00	Т	Butane	0.0	58	C4H10	106-97-8
c3038	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
c3038	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
c3038	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c3038	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c3038	0.00	Т	Propyne	0.0	40	C3H4	74-99-7
c3038	0.00	Т	1,3-Butadiene	0.0	54	C4H6	106-99-0
c3038	0.00	Т	Ethylacetylene	0.0	54	C4H6	107-00-6
c3038	1.75	Т	Methane	5.6	16	CH4	74-82-8
			sum:	6			

Sample No: T09-3039

Comments: Baseline- ERCB-PennWest 11-12-078-08 W6M SmpDate: 16-Sep-09 Time: 1110 By: CRG Matrix: SILCO Canister #: 1680 User Sample No: SVW-9

	SubC	Group	: TCD	Concentration			
FILE	RT	MQ	NAME	percent	MW	MolFormula	CAS
				*************			
	Analysis	Date	:				
G3039	0.00	Т	Carbon monoxide	0.0	28	CO	630-08-0
G3039	1.53	Т	Carbon dioxide	.2	44	C02	124-38-9
G3039	3.33	Т	Oxygen	21.1	32	02	7782-44-7
G3039	6.06	Т	Nitrogen	78.5	28	N2	7727-37-9
			sum:	100			

	SybC	Group	: clc4	Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysis	Date	: 24-SEP-2009 00:00				
c3039	0.00	Т	Propylene	0.0	42	C3H6	115-07-1
c3039	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c3039	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
c3039	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
c3039	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c3039	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c3039	0.00	Т	Propyne	0.0	40	C3H4	74-99-7
c3039	0.00	Т	1,3-Butadiene	0.0	54	C4H6	106-99-0
c3039	0.00	Т	Ethylacetylene	0.0	54	C4H6	107-00-6
c3039	1.74	Т	Methane	3.2	16	CH4	74-82-8
c3039	2.20	Т	Ethane	6.0	30	C2H6	74-84-0
c3039	2.89	Т	Ethylene	. 9	28	C2H4	74-85-1
c3039	4.23	Т	Propane	. 4	44	СЗН8	74-98-6
c3039	7.80	Т	Isobutane	. 3	58	C4H10	75-28-5
c3039	8.24	Т	Butane	.5	58	C4H10	106-97-8
			sum:	11			

page 7

Sample No: T09-3040

Comments: Baseline- ERCB-PennWest 11-12-078-08 W6M SmpDate: 16-Sep-09 Time: 1120 By: CRG Matrix: SILCO

Canister #: 1843 User Sample No: SVW-1

	Sub	Group	: TCD	Concentration			
FILE	RT	MQ	NAME	percent	MW	MolFormula	CAS
	Analysis	Date	:				
G304	0.00	Т	Carbon monoxide	0.0	28	CO	630-08-0
G304	0 1.52	Т	Carbon dioxide	. 9	44	C02	124-38-9
G304	0 3.33	Т	Oxygen	20.9	32	02	7782-44-7
G304	0 6.07	Т	Nitrogen	78.2	28	N2	7727-37-9
			sum:	100			

	Sub(	Group	: clc4	Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysis	Date	: 24-SEP-2009 00:00				
c3040	0.00	т	Ethylene	0.0	28	C2H4	74-85-1
c3040	0.00	Т	Propylene	0.0	42	СЗН6	115-07-1
c3040	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c3040	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
c3040	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
c3040	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c3040	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c3040	0.00	Т	Propyne	0.0	40	C3H4	74-99-7
c3040	0.00	Т	1,3-Butadiene	0.0	54	C4H6	106-99-0
c3040	0.00	Т	Ethylacetylene	0.0	54	C4H6	107-00-6
c3040	1.74	Т	Methane	717.0	16	CH4	74-82-8
c3040	2.21	Т	Ethane	26.6	30	C2H6	74-84-0
c3040	4.23	Т	Propane	52.3	44	СЗН8	74-98-6
c3040	7.81	Т	Isobutane	20.9	58	C4H10	75-28-5
c3040	8.24	Т	Butane	19.5	58	C4H10	106-97-8
			sum:	836			

page 8

Sample No: T09-3049

Comments: Baseline- ERCB- PennWest 11-12-078-08 W6M SmpDate: 16-Sep-09 Time: 1130 By: CRG Matrix: SILCO Canister #: 1152 User Sample No: SVW-8

	SubG	Group	: TCD		Concentration				
FILE	RT	MQ	NAME		percent	MW	MolFormula	CAS	
	Analysis	Date	:						
G3049	0.00	т	Carbon monoxide		0.0	28	CO	630-08-0	
G3049	1.51	Т	Carbon dioxide		2.7	44	C02	124-38-9	
G3049	3.33	Т	Oxygen		18.0	32	02	7782-44-7	
G3049	6.06	Т	Nitrogen		79.6	28	N2	7727-37-9	
				sum:	100				

	SubG	Group	: clc4	Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysis	Date	: 29-SE9-2009 00:00				
c3049	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c3049	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c3049	0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c3049	0.00	Т	1,3-Butadiene	0.0	54	C4H6	106-99-0
c3049	0.00	Т	Ethylacetylene	0.0	54	C4H6	107-00-6
c3049	1.73	Т	Methane	2,540.0	16	CH4	74-82-8
c3049	2.19	Т	Ethane	1,800.0	30	C2H6	74-84-0
c3049	2.87	Т	Ethylene	48.2	28	C2H4	74-85-1
c3049	4.22	Т	Propane	122.0	44	СЗН8	74-98-6
c3049	6.57	Т	Propylene	2.9	42	СЗН6	115-07-1
c3049	7.80	Т	Isobutane	40.6	58	C4H10	75-28-5
c3049	8.23	Т	Butane	41.7	58	C4H10	106-97-8
c3049	10.35	Т	trans-2-Butene	. 8	56	C4H8	624-64-6
c3049	10.47	Т	1-Butene	.6	56	C4H8	106-98-9
c3049	12.41	Т	Propyne	21.5	40	C3H4	74-99-7
			sum:	4,618		-	

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Sample No: T09-3050

Comments: Baseline- ERCB- PennWest 11-12-078-08 W6M SmpDate: 16-Sep-09 Time: 1130 By: CRG Matrix: SILCO Canister #: H2799 User Sample No: SVW-108

	Sub	Group	: TCD	Concentration			
FILE	RT	MQ	NAME	percent	MW	MolFormula	CAS
2	Analysis	Date	:				
G3050	0.00	Т	Carbon monoxide	0.0	28	CO	630-08-0
G3050	1.52	Т	Carbon dioxide	2.0	44	CO2	124-38-9
G3050	3.33	Т	Oxygen	18.5	32	02	7782-44-7
G3050	6.06	Т	Nitrogen	76.8	28	N2	7727-37-9
			sum:	97			
2	Analysis	Date	:				
G3050Z	0.00	T	Carbon monoxide	0.0	28	CO	630-08-0
G3050Z	1.52	Τ	Carbon dioxide	2.2	44	CO2	124-38-9
G3050Z	3.33	т	Oxygen	18.8	32	02	7782-44-7
G3050Z	6.06	T	Nitrogen	77.4	28	N2	7727-37-9
			sum:	98			

	Sul	oGroup	: clc4	Concentration			
FILE	RT	MQ	NAME	ppmv	MW	MolFormula	CAS
	Analysi	s Date	: 29-SE9-2009 00:00				x
c3050	0.00	Т	Ethylene	0.0	28	C2H4	74-85-1
c3050	0.00	Т	Acetylene	0.0	26	C2H2	74-86-2
c3050	0.00	Т	trans-2-Butene	0.0	56	C4H8	624-64-6
c3050	0.00	Т	1-Butene	0.0	56	C4H8	106-98-9
c3050	0.00	Т	Isobutylene	0.0	56	C4H8	115-11-7
c3050	. 0.00	Т	cis-2-Butene	0.0	56	C4H8	590-18-1
c3050	0.00	T	1,3-Butadiene	0.0	54	C4H6	106-99-0
c3050	1.72	Т	Methane	12,700.0	16	CH4	74-82-8
c3050	2.18	Т	Ethane	982.0	30	C2H6	74-84-0
c3050	4.20	Т	Propane	552.0	44	СЗН8	74-98-6
c3050	6.55	Т	Propylene	.8	42	СЗН6	115-07-1
c3050	7.78	Τ	Isobutane	114.0	58	C4H10	75-28-5
c3050	8.21	Т	Butane	115.0	58	C4H10	106-97-8
c3050	12.38	Т	Propyne	31.8	40	C3H4	74-99-7
c3050	15.58	Т	Ethylacetylene	17.1	54	C4H6	107-00-6
			sum:	14,513			

FILE: datafile RT: retention time MQ: T=target compound or ##=PBM library match quality
Flg: nd=not detected U=non-target compound or Unknown
MDL: method detection limit MW: molecular weight CAS: chemical abstracts service

Certified For: Yogesh Kumar, Business Unit Manager Environmental Monitoring By: Alberta Research Council Vegreville, Alberta T9C 1T4 Date: Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill Environmental Monitoring Alberta Research Council Vegreville, Alberta T9C 1T4 Contact Person: Grant Prill (780) 632-8455

#### University of Alberta - Isotope laboratory 16-Oct-09 ISD SEC D

10 0 1 00	100	CEC.	-	D		0	C I I I D	1201	1000	1202	10:04
16-Oct-09	LSD	SEC	1	ĸ	Iv1	Company	Sample ID	13C1	1302	1303	13164
SVW-15	6	12	78	8	6	ARC	T09=2954				
svw-6	6	12	78	8	6	ARC	T09=2955	-49.04	-33.82		
svw-2	6	12	78	8	6	ARC	T09=2956	-53.14	-31.9	-31.17	
svw-1	6	12	78	8	6	ARC	T09=2957	-49.03	-32.51	-31.51	-31.3
svw-15	11	12	78	8	6	ARC	T09-3038				
svw-9	11	12	78	8	6	ARC	T09-3038		-26.38		
svw-1	11	12	78	8	6	ARC	T09-3040	-37.43	-25.26		
svw-8	11	12	78	8	6	ARC	T09-3049	-40.62	-27.35	-26.14	-27.03
svw-108	11	12	78	8	6	ARC	T09-3050	-40.62	-27.91	-25.9	-26.85



### Attention: HAROLD SLATER

PENN WEST PETROLEUM LTD. CALGARY ENVIRONMENTAL 200. 207-9th Avenue SW CALGARY, AB T2P 1K3 CANADA

#### Report Date: 2010/10/12

Job/Samp	le	AnalysisType	
B091320/	X19661	Carbon Isotope PPM	
B091320/	X19662	Carbon Isotope Gas	
B091320/	X19663	Carbon Isotope Gas	

PENN WEST LIBRARY PENN WEST LIBRARY PENN WEST LIBRARY

Well Name/Sample ID

Sample Point

SOIL GAS (0.3M NE OF WELLHEAD) SOIL GAS (1.8M NE OF WELLHEAD) SOIL GAS (0.3M SE OF WELLHEAD)

Encryption Key

Please direct all questions regarding this Certificate of Analysis to your Project Manager.

DARLYCE SIMPSON, Email: dsimpson@maxxamanalytics.com Phone# (780) 378-8500

Maxxam has procedures in place to guard against improper use of the electronic signature and have the required "signatories", as per section 5.10.2 of ISO/IEC 17025:2005(E), signing the reports. For Service Group specific validation please refer to the Validation Signature Page.

**Report Distribution** eports(B091320)HAROLD SLATER

PENN WEST PETROLEUM LTD.

200, 207-9th Avenue SW

CALGARY, CANADA

Date of Issue

2010/10/12

All analyses are performed according to internal procedures that are based on current published reference methods.



### GAS ANALYSIS PPM

								B091320:X1966	1
	MaxxID	Client ID			Meter Numb	ber		Laboratory Number	
PENN WEST	PEIROLEUM	LID.					Well ID	ŝ	
PFNN WEST	LIBRARY				PK/DS		MAXXA	M ANALYTICS	
Well Name					Initials of Sample	ər	Sampling C	Company	
				SOIL GAS	(0.3M NE OF	WELLHEAD)	TEDLA	R BAG	
Field or Area		F	Pool or Zone	Sample Point			Container I	dentity	Percent Full
Test Recovery			Interval	Eleva	tions (m)	Sample Gathering Poi	nt	Solution	n Gas
			From:						
Test Type	No. Mul	tiple Recovery	10.	КВ	GRD	N 0.000000		W 0.000000	
	<ul> <li>Production Rates</li> </ul>		Gauge Pressures kPa	Temp	erature °C	GPS		GPS	
Water m3/d	Oil m3/d	Gas 1000m3/d	Source As Received	Source	As Received	Well Fluid Type		Licence No.	
2010/09/22 1	1:50		2010/09/24	2010/10/08	2010/10/1	12	MW		
Date Sampled Start	Di	ate Sampled End	Date Received	Date Reported	Date Reissued	1	Analyst		
	COMPO	SITION			1	PROPERTIES			
Component	Mole Fraction	í	Cal	culated Molar Mass	Calculated Gro	ss Heating Value (MJ/m3)		Calculated Relative	Densities
componione	As Rec`d	ppm (v/v)		Moisture Free as Sampled	@ 101.3	325 kPa & 15°C		Relative to A	ri @15°C
	· · · ·			18.5	30	0.99		0.640	
H2	Trace			Total		GPA 2172		Moisture Free as S	ampled .
He	Trace								
02	0.1886			On Site	Hy	drogen Sulphide	In	Lab	
N2	0.7199			On Sile				Lab	
CO2	0.0016					<1			
H2S	0.0000		G	Gastec (ppm v/v) Tut	weiler (mole%)	Gastec (ppm v/v)	Tutweiler (r	mole%) H2S from GC (m	ole%)
				- the samely size is seen time of the					
C1	0.0894			S degrades variably in all sam	ple containers and is al	so matrix dependant.			6
C2	0.0004	350							
C3	0.0001	110							
IC4	Trace	10							
NC4	Trace	10	<i>2</i>						
IC5	Trace	Trace							
NC5	0.0000								
C6	0.0000								
C7+	0.0000								
TOTAL	1,0000		OC Chec	k Std # 5878/8167	Date 2010/0	19/27 OC Pag	sed Yes		
	1.0000								
s				** Information not suppl	ied by client dat	a derived from LSD i	nformation	Results relat	te only to items tes

Remarks:

CALGARY 2021-41 Avenue N.E., Calgary, Canada T2E 6P2 Tel: (403) 291-3077 Fax (403) 291-30-45634 NDE PRAIRIE #101,7002 - 98 Street, Clairmont, Canada T0H 0W0 Tel: (780) 532-0227 Fax (780) 532-0288 EDMONTON 6744-50 Street, Edmonton, Canada T6B 3M9 Tel: (780) 378-8500 Fax (780) 378-86550 DEER Bay #3, 4845 79 Street, Red Deer, Canada T4P 2T4 Tel: (403) 341-8811 Fax (403) 341-881

To view or Download your data on-line via MaxxLINK, please call 1-800-386-7247



### GAS MIGRATION ANALYSIS

								BO	91320:X19662	
	MaxxID	Client ID				Meter Nun	nber	Lab	ooratory Number	
INN WEST	F PETROLEUM	LTD.								
Operator Name						LSD		Well ID		
PENN WEST	LIBRARY					PK/DS		MAXXAM	ANALYTICS	
Well Name					0011 010	Initials of Samp	ler	Sampling Com	pany	
					SOIL GAS	(1.8M NE OF	WELLHEAD)	TEDLAR	BAG	
Field or Area			Pool or Zone		Sample Point			Container Iden	tity	Percent Full
Test Recovery			From:	erval	Eleva	ations (m)	Sample Gathering Poi	nt	Solution	Gas
Test Type	No. Mult	tiple Recovery	To:		КВ	GRD				
	Braduction Poton		Gauga Ba			ereture °C	→ N 0.000000		W 0.000000	
	- Floudcuon Rales		Gauge Fil	5350163 NI 8	16mp	21	GPS		GPS	
Water m3/d	Oil m3/d	Gas 1000m3/d	Source	As Received	Source	As Received	Well Fluid Type		Licence No.	
2010/09/22 1	2:50		2010	)/09/24	2010/10/08	2010/10/	'12	MW ,DT1		
Date Sampled Start	Da	ate Sampled End	Date R	Received	Date Reported	Date Reissue	ed	Analyst	1	
	COMPO	SITION					PROPERTIES			
0	Mala Franking	I		Calcu	ated Molar Mass		oss Heating Value (M 1/m3)		- Calculated Polative De	notice
Component	As Rec'd	ppm (v/v)	δ <sup>13</sup> C7 <sub>00</sub>	Mo	bisture Free as Sampled	@ 101	.325 kPa & 15°C		Relative to Ari	@15°C
				17	7.5	3	4.38		0.604	
H2	Trace				Total		GPA 2172		Moisture Free as San	npled
He	Trace									
02	0.1366				On Site	—— Н	ydrogen Sulphide	In Lat	2	
N2	0.5389				onoite			in La	5	
CO2	0.0040		-36.14				<1			
H2S	0.0000			Gas	tec (ppm v/v) Tut	weiler (mole%)	Gastec (ppm v/v)	Tutweiler (mole	%) H2S from GC (mol	e%)
					a na a na marana		1			
1	0.3179		-54.94	Unsi	te analysis is required f	or accurate source	e H2S content.			
-C2	0.0025		-36.07							
C3	0.0001	110	-30.21	δ"C %.=	= [( <sup>13</sup> C / <sup>12</sup> C	- "C / "C	)/("C/"C	)] * 1000	ר	
					EC (Guipte	II	TERPRETATION	131172	-	
1C4	Trace	30	-28.68							
NCA	Trace	10	-24 52							
1104	indoo	10	21.02							
IC5	Trace	10								
NC5	Trace	Trace								
18 - 18 C										
C6	0.0000									
C7+	0.0000									
TOTAL	1.0000			QC Check S	Std # 5878/8167	Date 2010/0	)9/27 QC Pas	sed Yes		
				**	Information not suppl	ied by client da	ta derived from LSD i	nformation	Results relate	only to items test
Contraction of the local division of the loc				1						, to nom o teate

Remarks:

CALGARY 2021-41 Avenue N.E., Calgary, Canada T2E 6P2 Tel: (403) 291-3077 Fax (403) 291-3645630 DE PRAIRIE #101, 7002 - 98 Street, Clairmont, Canada T0H 01W0 Tel: (780) 532-0227 Fax (780) 532-0288 EDMONTON 6744-50 Street, Edmonton, Canada T6B 3M9 Tel: (403) 378-8500 Fax (780) 378-865590 DEER Bay #3, 4845 79 Street, Red Deer, Canada T4P 2T4 Tel: (403) 341-8811 Fax (403) 341-8815

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Ma

### **GAS MIGRATION ANALYSIS**

								B	091320:X19663
and Married States	MaxxID	Client ID				Meter Nun	nber	La	boratory Number
PENN WEST	PETROLEUM	I LTD.							(
Operator Name			а.			LSD			
PENN WEST	LIBRARY				>	- PK/DS		MAXXAN	ANALYTICS
Well Name						Initials of Samp			npany
					SOIL GAS	(0.3M SE OF	WELLHEAD)	TEDLAR	BAG
Field or Area		,	ool of Zone		Sample Point			Container Ide	ntity Percent Full
Test Recovery			Inter	val	Eleva	tions (m)	Sample Gathering Poir	nt	Solution Gas
			From:			-			
Test Type	No. Mu	tiple Recovery	10:		KB	GRD	N 0 000000		W 0 00000
[	- Production Rates -		Gauge Pres	sures kPa ———	Tempe	rature °C	GPS		GPS
						21			
Water m3/d	Oil m3/d	Gas 1000m3/d	Source	As Received	Source	As Received	Well Fluid Type		Licence No.
2010/09/22 1	1:40		2010/	09/24	2010/10/08	2010/10/	12	MW ,DT1	
Date Sampled Start	D	ate Sampled End	Date Re	ceived	Date Reported	Date Reissue	d	Analyst	
	COMPO	SITION					PROPERTIES		
		I	. 0	Calau	Inted Malor Mass	Coloulated Co			Coloristed Delative Dec. 11
Component	As Rec'd	ppm (v/v)	δ <sup>13</sup> C <sup>7</sup> 0	Mc	bisture Free as Sampled	@ 101	.325 kPa & 15°C		Relative to Ari @15°C
				20	0.3	2	5.72		0.701
110	Trace				Total	-	GPA 2172		Moisture Free as Sampled
	Trace					L			
He	Irace					u	udrogon Sulphido		P
02	0.1754				On Site	11		In La	b
N2	0.7031								
CO2	0.0036		-40.49				<1		
H2S	0.0000			Gas	tec (ppm v/v) Tutv	reiler (mole%)	Gastec (ppm v/v)	Tutweiler (mo	e%) H2S from GC (mole%)
				Onci	to applyzic in required f		 H2C content		
C1	0.1174		-53.39	H2S	degrades variably in all same	or accurate source	BISO matrix dependent		
C2	0.0004	450	-37.93						
C3	0.0001	120	-33.93	8°C% :	= [( <sup>13</sup> C / <sup>13</sup> C	- 1ºC / 1ºC	MCC/PC	) 1 * 100	Ω
00	0.0001	120	00.00		Lt C t C; anple	IN		aan 100	
104	-	10	24.22						
104	Trace	10	-31.33						
NC4	Trace	10							
IC5	Trace	Trace							
NC5	0.0000				2				
C6	0.0000								
C7+	Trace	20	5. 						
TOTAL	1 0000	20			NA # E070/0407	D-1- 0040"	0/07 005	1.27	
IUIAL	1.0000			QC Check S		Date 2010/0	QC Pas	sed Yes	
				**	Information not suppli	ed by client da	ta derived from LSD in	nformation	Results relate only to items te

Remarks:

CALGARY 2021-41 Avenue N.E., Calgary, Canada T2E 6P2 Tel: (403) 291-3077 Fax (403) 291-34656NDE PRAIRIE #101,7002 - 98 Street, Clairmont, Canada T0H 0W0 Tel: (780) 532-0227 Fax (780) 532-0286 EDMONTON 6744-50 Street, Edmonton, Canada T6B 3M9 Tel: (780) 378-8500 Fax (780) 378-85509 DEER Bay #3, 4845 79 Street, Red Deer, Canada T4P 2T4 Tel: (403) 341-8811 Fax (403) 341-8815

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### Dr. Karlis Muehlenbachs' Carbon Isotope Abundance

28-Sep-10	LSD	Company	Sample ID	13C1	13C2	13C3	13iC4	13nC4	CO2
Soil Gas (1.8m NE of WH)	Penn West Library	Penn West	B091320-X19662	-54.94	-36.07	-30.21	-28.68	-24.52	-36.14
Soil Gas (0.3m SW of WH)	Penn West Library	Penn West	B091320-X19663	-53.39	-37.93	-33.93	-31.33		-40.49

Carbon Isotope Abundance done by Dr. Karlis Muehlenbachs at the University of Alberta Email: karlis.muehlenbachs@ualberta.ca



## Soil Gas Summary

Maxxam File #: B091320 - X19662, X19663 Sampling Date: 2010/09/22

Location: Penn West Library

Interpretation:

### Maxxam

SCV Gas Possible Depth:

400 – 600 m

SCV Gas Geologic Formation: From or near Dunvegan

### Maxxam's Remarks:

- These gases have similar signatures to the soil gas samples taken previously from this area.
- Based on the gas and carbon isotopic data, a possible source for these soil gases is from the Dunvegan formation, at a depth of 400 to 600 m.

Maxxam's interpretation and remarks were done by Darlyce Simpson

Email: darlyce.simpson@maxxamanalytics.com

Max

### **Gas Migration Data Sheet**

Company Name: Penn West Contact Name: Harold Slater Phone: 403-777-2500 Email: harold.slater@pennwest.com

Field Contact: Larry Gordey Phone: 780-818-2330

#### Maxxam Job Number: B091320

UWI: Penn West Library Pad/Surface: N/A License: N/A Well Name: N/A Date: 2010/09/22 Initials: PK/DS

### SOIL GAS SURVEY DATA

Note: Please record observed "zero" readings for soil gas Lower Explosive Limit (LEL). Do not leave blanks. The centre dot in the diagram represents the wellbore/casing.



On the diagram, indicate the location of sample points and record percentage LEL readings, including observed "zero" readings. Briefly describe test results and the condition of the area around the wellbore and on lease,

e.g., vegetation growth, bare spots, contaminated soil, etc.

Note: Postion and number of test points are subject to the proximity of buried facilities.

Gravel pad, clay underneath

Surrounded by pasture, no apparent vegetation stress

0.3m NE of wellhead - clay has oily smell

### **GAS MIGRATION TEST REPORT**

WELL NAME:	NORTHSTAR ET AL SI	PIRITR 6-12-78-8		UNIQUE WELL IDENTIFIER :	00/06-12-078-08W6/
GAS MIGRA DISTANCE At Wellhead At 2m from Well At 4m from Well At 6m from Well	IDON TEST % LEL I           NORTH         EAS	MEASUREMEN           ST         SOUTH         W           0.0         %         0.0         %           0.0         %         0.0         %           0.0         %         0.0         %           0.0         %         0.0         %           0.0         %         0.0         %           0.0         %         0.0         %	TS /EST % % % %	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	6 m 4 m 2 m 0.0 0.0 0.0 0.0
Is Gas Migratio	on a fire, public safety, c hazard?	or off lease	NO	0.0	
EUB Classifica	ation: <u>NO GM</u>			0.0	

### COMMENTS:

The RKI eagle has an accuracy level of +/- 5% LEL.

A CCS Company

There is suspected Hydrocarbon contamination visible around wellhead to depths of 1m and greater.

The test points were also checked with methane elimination procedure to find that the readings were still zero % LEL. There is no gas migration found along NESW axis points at 1.5-2m's below surface.

### PICTURE(S):



MARK HUNT	TEST DATE:	11-Nov-10
	MARK HUNT	MARK HUNT TEST DATE:

### **NORTHSTAR ET AL SPIRITR 6-12-78-8**



6-12-78-8W6 Penn West 11-Nov-10 Hydro vac nearby pipeline



175cm - 200cm Deep North Of Well

### **NORTHSTAR ET AL SPIRITR 6-12-78-8**



**NORTHSTAR ET AL SPIRITR 6-12-78-8** 



6-12-78-8W6 Penn west 11-Nov-10 Excavated furrows south 175cm-200cm

**NORTHSTAR ET AL SPIRITR 6-12-78-8** 



NORTHSTAR ET AL SPIRITR 6-12-78-8



6-12-78-8W6 Penn West 11-Nov-10 Found cason drainage culvert while digging south furrow



6-12-78-8W6 Penn West 11-Nov-10 Excavated furrows near well south 175cm-200cm

## NORTHSTAR ET AL SPIRITR 6-12-78-8



6-12-78-8W6 Penn West 11-Nov-10 Excavated furrows near well west 175cm-200cm



P.O. Box 75003, RPO Cambrian Calgary, AB T2K 6J8 Phone: (403) 282-3999 Fax: (403) 282-0815

November 25, 2009

Energy Resources Conservation Board 640 – 5<sup>th</sup> Avenue SW Calgary, AB T2P 3G4

Attention: Michael Bevan, M.Sc., P.Geol.

### RE: PENN WEST 06/11-12-078-08 W6M Gas Migration Assessments

Baseline Water Resource Inc. (BWRI) was retained by the Energy Resources Conservation Board (ERCB) to conduct gas migration assessments on September 15-16, 2009, at two well sites near Spirit River, Alberta (Figure 1). The purpose of the assessments were to determine if formation gas from the Penn West 06-12-078-08 W6M and Penn West 11-12-078-08 W6M energy wells was migrating to surface. These locations were selected by the ERCB because of their proximity to a water well known to contain thermogenic gases. The location of the two energy wells and the approximate location of the water well are presented in Figure 2.

The following report consists of details of the field investigation, however, discussion or interpretation of results is not included. Photographs of the gas migration assessments are provided in Appendix A.

### **SCOPE OF WORK**

The Scope of Work included the following:

- Review and coordinate ground disturbance procedures with Penn West operations. Arrange for professional line locators to locate underground facilities.
- Install and purge Soil Vapour Wells (SVW) at each site.
- Collect Organic Vapour Measurements (OVM) from each SVW to determine lab sample submission requirements.

- Collect soil gas samples from the three SVW's with the greatest OVM readings and a background sample at each site for submission to the Alberta Research Council (ARC) in Vegreville, Alberta for gas composition and to the University of Alberta (UofA) in Edmonton, Alberta for isotopic analysis.
- Collect one field duplicate gas sample for submission to the ARC for gas composition and the UofA for isotopic analysis.
- Collect two soil samples (06-12-078-08 W6M (SVW-1) and 11-12-078-08 W6M SVW-1)) for submission to ARC for Petroleum Hydrocarbon Fractions (PHC) (F1-F4) and BTEX analysis.
- Decommission all SVW's and remove all flags, stakes and paint prior to leaving site.
- Prepare a report outlining the findings of the gas migration assessments.

### FIELD METHODOLOGIES - STANDARD OPERATING PROCEDURES

### SAFETY

Prior to commencing work, BWRI ensured that all client and BWRI Ground Disturbance procedures were followed. The most stringent protocols were used. BWRI Ground Disturbance procedures are outlined in the "Company Policy and Procedure Manual: Safe Work and Environmental Operations" (Sec. 7.2). In addition to the Alberta 1 Call (Ticket # 9248852), a professional line locating contractor, Red Alta, based out of Edmonton, Alberta was retained to complete a 30 m radius sweep at each subject site (wellhead). BWRI field personnel reviewed all available surveys and drawings with site operators to ensure all underground facilities had been identified. Ground disturbance activities were not conducted without prior approval from both the client and the owner of the underground facility. In addition, a Penn West consultant was onsite for the duration of all ground disturbance activities.

### PROCEDURE

Gas migration testing was conducted in accordance with the standardized procedures provided in ERCB's Directive 020: Well Abandonment Guide (revised edition July 24, 2007).

Fifteen boreholes were hand augured to a minimum depth of 80 cm below ground surface (bgs). Two locations were bored within 30 cm of the wellhead. Twelve locations were bored at 2 m intervals every  $90^{\circ}$  to a maximum distance of 6 m. One sample location was selected to provide a background measurement. Excess fill (gravel) was removed prior to completing the borehole to ensure the slotted area of the SVW was completed in native subsoil. Whenever possible, the sample bearing ( $0^{\circ}$ ) was followed for SVW placement, however, site specific conditions (i.e. pipelines, above ground facilities, boulders) may have resulted in field personnel adjusting the location.

### INSTALLATION

The SVW's are comprised of a 70 cm long by 10 cm diameter slotted PVC pipe equipped with a sampling port and valve. One SVW was inserted into each borehole. Frac sand (10/20 size) was placed around the slotted portion of the SVW and sealed at ground surface with hydrated bentonite powder and soil cuttings to ensure that no mixing of soil and atmospheric gases occurred. Each SVW was purged for a minimum of 10 minutes using an SKC transfer pump at a rate of 3 L/min. SVW's were then left to equilibrate overnight. The next day, OVM's were collected from each SVW using an RKI Eagle gas detector. Field calibration of the RKI Eagle was conducted prior to measurement collection.

### SOIL GAS SAMPLING

The three SVW's with the greatest OVM readings and the background SVW were selected for soil gas sample collection at each site. One field duplicate sample was collected for QA/QC purposes. Samples submitted to ARC were collected using 6 L Silco Steel canisters provided by ARC. Samples submitted to the UofA were collected in 3 L high quality foil lined gas bags provided by BWRI.

### SILCO STEEL CANISTERS

Nine gas samples were collected using 6 L Silco Steel canisters and submitted to ARC for gas composition analysis. A short piece of 1/8 inch Teflon tubing was attached to the canister and connected to the SVW sample port with a 1 inch piece of Tygon tubing provided by ARC. New Teflon and Tygon tubing was used at each sample location. Once connected, the canister was opened and allowed to fill for 5 minutes. Samples were placed in a cooler and shipped to ARC on September 17, 2009 via Purolator Courier based in Edmonton, Alberta.

### FOIL GAS BAGS

Nine gas samples were collected using 3 L high quality foil gas bags and submitted to UofA for isotopic analysis. A lung sampler connected to an SKC pump was used to create negative pressure and draw the soil gas sample from the SVW sample port into the foil gas bag. New ¼ inch Teflon and Tygon tubing was attached to the sampling port and the lung sampler at each sample location. The SKC pump was set at a rate of 3 L/min and ran for 30 seconds to draw approximately 1500 mL of soil gas into each bag. Samples were delivered to the UofA in Edmonton, Alberta on September 17, 2009.

### SOIL SAMPLING

A soil sample (06-12-078-08 W6M (SVW-1) and 11-12-078-08 W6M (SVW-1)) was collected from a depth of approximately 50 cm bgs at each site. Soil was placed into two 250

mL jars at each sample location and submitted to ARC for PHC F1-F4 and BTEX analysis. The samples were placed in a cooler with ice and shipped to ARC on September 17, 2009 via Purolator Courier based in Edmonton, Alberta.

### COMPLETION

All SVW's were removed and backfilled with soil cuttings, fill and bentonite to previous grade. SVW's were cleaned of any soil and bentonite with a bleach solution. All stakes, flags and paint from utility locates were removed prior to leaving site.

### RESULTS

### PENN WEST 06-12-078-08 W6M

The Penn West 06-12-078-08 W6M wellsite consisted of a wellhead, well shed, propane bullet, and underground pipelines at the time of assessment (Figure 3).

On September 15, 2009 BWRI installed fifteen SVW's. A sample bearing of  $0^{\circ}$  was implemented on site. Subsurface soil conditions ranged from dry to moist throughout the site and were deemed suitable to complete the assessment. A previous precipitation event in the vicinity of the site was on September 7, 2009.

On September 16, 2009 OVM measurements were collected to determine sampling requirements. Measurements ranged from 25 parts per million (ppm) (SVW-13 and SVW-14) to 100% lower explosive limit (LEL) (SVW-1). The locations selected for soil gas sampling were SVW-1 (100% LEL), SVW-2 (225 ppmv), SVW-6 (72% LEL), and the background location (SVW-15) (30 ppm). The background location (SVW-15) was installed approximately 20.5 m northeast (45°) of the wellhead.

A soil sample was collected from the SVW-1 borehole and submitted to ARC for PHC F1-F4 and BTEX analysis. The soil sample was labeled SS-1 on the chain of custody.

Sample locations and OVM measurements are provided in Figure 3.

### PENN WEST 11-12-078-08 W6M

The Penn West 11-12-078-08 W6M wellsite consisted of a wellhead, concrete pad, aboveground pipelines, and underground pipelines at the time of assessment (Figure 4).

On September 15, 2009 BWRI installed fifteen SVW's. A sample bearing of  $0^{\circ}$  was implemented on site. Subsurface soil conditions ranged from dry to moist throughout the site and were deemed suitable to complete the assessment. A previous precipitation event in the vicinity of the site was on September 7, 2009.

On September 16, 2009 OVM measurements were collected to determine sampling requirements. Measurements ranged from 10 ppm (SVW-11) to 6% LEL (SVW-8). The locations selected for soil gas sampling were SVW-1 (400 ppm), SVW-8 (6% LEL), SVW-9 (75 ppm), and the background location (SVW-15) (25 ppm). The background location (SVW-15) was installed approximately 20.97 m northwest (315°) of the wellhead. A field duplicate was collected from SVW-8 and labeled SVW-108 on the chain of custody.

A soil sample was collected from the SVW-1 borehole and submitted to ARC for PHC F1-F4 and BTEX analysis. The soil sample was labeled SS-1 on the chain of custody.

Sample locations and OVM measurements are provided in Figure 4.

### REFERENCES

Alberta Energy and Utilities Board, 2007. "EUB Directive 020: Well Abandonment Guide" July, 2007. Appendix 2.

### CONCLUSION

BWRI appreciates the opportunity to assist the ERCB with this project. If you have any comments or suggestions regarding the report, please do not hesitate to contact either of the undersigned at 403-282-3999.

Respectfully submitted,

S. Brent Bowerman, P.Geol. President

Clint R. Ganes, B.Sc. Operations Manager

### DISCLAIMER

Baseline Water Resource Inc. has used proficient skill and diligence conducting the gas migration assessment and preparation of this report. This report is a representation of the conditions and information present and available at the time of the assessment. Information received from all other sources is considered to be accurate but cannot be guaranteed. Baseline Water Resource Inc. is not responsible for any individual interpretation of this material nor any decisions based upon the findings in this report.

# FIGURES

Figure 1. Site Location Figure 2. Water Well Location Figure 3. 06-12-078-08 W6M Soil Vapour Well Locations Figure 4. 00/11-12-078-08 W6M Soil Vapour Well Locations









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# **APPENDIX A**

Site Photographs





Photograph 1. The 06-12-078-08 W6M wellsite, view northeast.



Photograph 2. The 06-12-078-08 W6M wellsite, view southwest.

BWRI Project: 09-7003 Date: October 28, 2009 **Baseline** 



Photograph 3. The 00/11-12-078-08 W6M wellsite, view northeast.



Photograph 4. The 00/11-12-078-08 W6M wellsite, view southwest.

BWRI Project: 09-7003 Date: October 28, 2009




Photograph 5. View of an open borehole with SVW.



Photograph 6. View of an installed SVW with 10/20 frac sand.



BWRI Project: 09-7003 Date: October 28, 2009



Photograph 7. View of SVW sealed with hydrated bentonite powder.



Photograph 8. View of SVW being purged after installation.

BWRI Project: 09-7003 Date: October 28, 2009





Photograph 9. Soil gas sampling with a Silco Steel canister.



Photograph 10. Soil gas sampling with a lung sampler and foil gas sampling bag.



### Appendix C

Isotopic Mudlog (100/02-04-078-07 W6M) Analytical Results

#### MORNING REPORT OCT. 28, 29, 2009

	EXSHAW OIL SPIRIT 2-4-78-7			GAS SAMPLING		
DEPTH(m)	SAMPLE #	DATE	TIME	CO2	GAS (unit)	
340.0m	1	Oct. 28,2009	15:54hr	131ppm	1324	
407.0m	2	Oct.28, 2009	17:28hr	116.0ppm	752	
657.0m	3	Oct.29, 2009	01:56hr	179.0ppm	735	
703.0m	Missed	Oct.29, 2009	05:08hr	93.0ppm	1097	
734.0m	4	Oct.29, 2009	05:58hr	98.0ppm	444	
844.0m	5	Oct.29,2009	08:40hr	115.0ppm	705	
871.0m	6	Oct. 29, 2009	09:34hr	111ppm	607	
920.0m	7	Oct.29,2009	11:16hr	ppm	442	
925.0m	8	Oct.29, 2009	11:29hr	ppm	846	
END						

Sample Depth	Date sampled	CO2 (ppm)	Total Gas (unit)	δ <sup>13</sup> C C <sub>1</sub>	δ <sup>13</sup> C C <sub>2</sub>	δ <sup>13</sup> C C <sub>3</sub>	δ <sup>13</sup> C i-C <sub>4</sub>	δ <sup>13</sup> C n-C <sub>4</sub>	δ <sup>13</sup> C CO <sub>2</sub>
340	28-Oct-09	131	1324	-56.78	-36.29	-32.02	-30.49	-33.44	-21.95
407	28-Oct-09	116	752						
657	29-Oct-09	179	735	-55.18	-40.72	-38.40	-35.16		-18.40
734	29-Oct-09	98	444	-51.45	-35.00	-36.32	-28.88	-36.64	-12.20
844	29-Oct-09	115	705	-44.65	-28.11	-29.04	-30.49		-15.07
844 lab duplicate	29-Oct-09	115	705	-44.77	-28.22	-28.90	-29.23	-31.49	-15.57
871	29-Oct-09	111	607	-45.76	-29.37	-30.89	-29.31	-33.60	-29.13
920	29-Oct-09		442	-42.64	-29.01	-28.57	-27.99	-30.05	-11.74
925	29-Oct-09	(100.000)	846	-42.60	-28.91	-28.22	-28.05	-29.92	

Gas mudlog data from 100/02-04-078-07 W6M

JackWellPhaseII\_8789028\_CBM\Data\Lab\_Data\SpiritRiverMudlog\mud log 1-4-78-7W6.xls

## Appendix D

### **Pumping Test Results**













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